



Docket No.: 05542/071002  
(PATENT)

**IN THE UNITED STATES PATENT AND TRADEMARK OFFICE**

In re Patent Application of:  
Robert L. Horton et al.

Application No.: 10/620,041

Confirmation No.: 7482

Filed: July 15, 2003

Art Unit: 1712

For: SURFACTANT-POLYMER COMPOSITIONS  
FOR ENHANCING THE STABILITY OF  
VISCOELASTIC-SURFACTANT BASED  
FLUID

Examiner: C. R. Richard

**UNAVAILABILITY OF INVENTOR PETITION UNDER 37 C.F.R. § 1.47**

MS AF  
Commissioner for Patents  
P.O. Box 1450  
Alexandria, VA 22313-1450

Dear Sir:

In response to the Petition Dismissal of January 11, 2006, please consider the following reply.

The Application lists seven inventors: Robert L. Horton, Bethicia B. Prasek, Frederick B. Growcock, David Kippie, John W. Vian, Kamila B. Abdur-Rahman, and Morris Arvie, Jr. A signed Declaration for Robert L. Horton, Bethicia B. Prasek, Frederick B. Growcock, David Kippie, John W. Vian, and Morris Arvie, Jr. in compliance with 37 C.F.R. § 1.68 was submitted with the filing of the application on July 15, 2003. Kamila B. Abdur-Rahman signed that Declaration but failed to date her signature. On November 14, 2005, a first Petition was filed, including chronological proof of the pertinent facts relating to the petitioners efforts to obtain the supplemental Declaration.

Subsequently, the first petition was dismissed for failure to submit a copy of the entire application, evidence of that transmittal submitted, proof of reasonable effort to ascertain a

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current or last known address, and proof of due diligence effort ascertaining the same. Thus, in response to the dismissal, Petitioner submits a Declaration, made by Leah Bardin, which details attempts to contact Kamila B. Abdur-Rahman by additional means.

After a diligent effort, Leah Bardin has been unable to obtain a signed and dated supplemental Declaration from Kamila B. Abdur-Rahman. Leah Bardin has attempted to contact Kamila B. Abdur-Rahman via mail to her last known address, via mail to a second address on file with M-I L.L.C., via telephone book, internet, and telephone. Please find appended hereto an updated chronological proof of the pertinent facts relating to the petitioners efforts to obtain the supplemental Declaration:

Tab No.	Date(s)	Description
Tab 1	Assignment signed July 1, 2003	Assignment of rights to the invention known as "Surfactant-Polymer Compositions For Enhancing the Stability of Viscoelastic-Surfactant Based Fluid" to M-I L.L.C. by Kamila B. Abdur-Rahman
Tab 2	Declaration filed July 15, 2003	Signed Declaration of Robert L. Horton, Bethicia B. Prasek, Frederick B. Growcock, David Kippie, John W. Vian, and Morris Arvie, Jr. and undated Declaration of Kamila B. Abdur-Rahman accompanied the filing of the patent application titled "Surfactant-Polymer Compositiions For Enhancing the Stability of Viscoelastic-Surfactant Based Fluid," later designated by the U.S. Patent and Trademark Office as U.S. Patent Serial No. 10/620,041
Tab 3	Letter sent September 21, 2005	Certified letter containing application as filed and new Declaration sent to Kamila B. Abdur-Rahman at 12707 Bellaire Blvd. No. 303, Houston, Texas 77072. Certified letter returned to sender as "Attempted Not Known."
Tab 4	Letter sent April 21, 2006.	Certified letter containing application as filed and new Declaration sent to Kamila B. Abdur-Rahman at P.O. Box 1603, Alief, Texas 77411 Certified letter has not currently been received.
Tab 5	Declaration signed May 11, 2006	Declaration of Leah Bardin declaring first-hand knowledge of the pertinent facts

The last known residential address of Kamila B. Abdur-Rahman is:

12707 Bellaire Blvd. No. 303,  
Houston, Texas 77072

A second known address for Kamila B. Abdur-Rahman is:

PO Box 1603  
Alief, Texas 77411

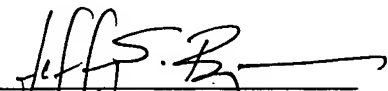
In view of the petitioner's diligent attempts to obtain the signature of the unavailable inventor, enclosed documentation, and pursuant to 37 C.F.R. § 1.47(a), the petitioner requests that that the referenced application be examined without the signed Declaration of Kamila B. Abdur-Rahman.

A credit card form SB-08 for the petition fee is enclosed along with this Petition. Please apply any charges not covered, or any credits, to Deposit Account No. 50-0591, under Order No. 05542/071002 from which the undersigned is authorized to draw.

Dated: May 11, 2006

Respectfully submitted,

By

  
Jeffrey S. Bergman  
Registration No.: 45,925  
OSHA LIANG LLP  
1221 McKinney St., Suite 2800  
Houston, Texas 77010  
(713) 228-8600  
(713) 228-8778 (Fax)

## ASSIGNMENT

In consideration of the sum of one dollar (\$1.00) or equivalent and other good and valuable consideration paid to each of the undersigned: Robert L. Horton; Bethicia Prasek; Frederick B. Growcock; David Kippie; John W. Vian; Kamila B. Abdur-Rahman; Morris Arvie, Jr. the undersigned hereby sell(s) and assign(s) to M-I LLC. (the "Assignee"), a corporation of Delaware, having a place of business at 5950 North Course Drive, Houston, TX 77072, his/her entire right, title and interest

check applicable box(es)      ☒: for the United States of America (as defined in 35 U.S.C. § 100),  
    ☒: and throughout the world,

in the invention(s) known as SURFACTANT-POLYMER COMPOSITIONS FOR ENHANCING THE STABILITY OF VISCOELASTIC-SURFACTANT BASED FLUID for which application(s) for Letters Patent in the United States of America has (have) been executed by the undersigned on event date (also known as United States Patent Application No. 10/, filed 2003), in any and all applications thereon, in any and all Letters Patent(s) therefor, and in any and all reissues, extensions, renewals, reexaminations of such applications or Letters Patent(s) and divisional and continuation applications thereof, to the full end of the term or terms for which such Letters Patent(s) issue, including all claims, if any, that may have arisen for infringement prior to the date of this assignment, such entire right, title and interest to be held and enjoyed by the above-named Assignee to the same extent as they would have been held and enjoyed by the undersigned had this assignment and sale not been made.

The undersigned agree(s) to execute all papers necessary in connection with the application(s) and any continuing (continuation, divisional, or continuation-in-part), reissue, reexamination or corresponding application(s) thereof and also to execute separate assignments in connection with such applications as the Assignee may deem necessary or expedient.

The undersigned agree(s) to execute all papers necessary in connection with any interference that may be declared concerning the application(s) or any continuing (continuation, divisional, or continuation-in-part), reissue or reexamination application thereof and to cooperate with the Assignee in every way possible in obtaining evidence and going forward with such interference.

The undersigned hereby authorizes and requests the government authority to issue any and all Letters Patent granted on such inventions to the Assignee.

The undersigned hereby warrants and represents that the undersigned has full right to convey the entire interest herein assigned, and that the undersigned has not executed, and will not execute, any agreement in conflict therewith.

The undersigned hereby expressly intends this agreement to be binding on the heirs, assigns, representatives and successors of the undersigned and extend to the successors, assigns and nominees of the Assignee.

IN WITNESS WHEREOF, executed by the undersigned on the date(s) opposite their name(s).



Signature: Robert L. Horton Date: 07/01/2003  
 Printed Name: Robert L. Horton

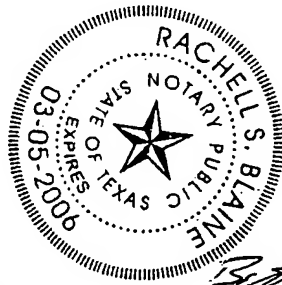
COUNTY OF Harris )  
 STATE OF Texas ) ss.

BEFORE ME, the undersigned authority, on this 1<sup>st</sup> day of July, 2003, personally appeared Robert Horton, known to me to be the person whose name is subscribed to the foregoing instrument and acknowledged to me that he executed the same of his own free will for the purposes and consideration therein expressed.

Rachell S. Blaine

Notary or Consular Officer

[SEAL]



\*\*\*\*\*

Signature: Bethicia Prasek Date: July 1, 2003  
 Printed Name: Bethicia Prasek

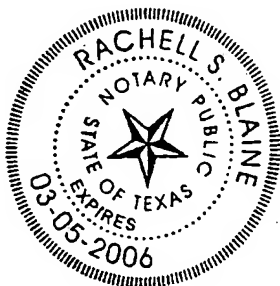
COUNTY OF Harris )  
 STATE OF Texas ) ss.

BEFORE ME, the undersigned authority, on this 1<sup>st</sup> day of July, 2003, personally appeared Bethicia Prasek, known to me to be the person whose name is subscribed to the foregoing instrument and acknowledged to me that she executed the same of her own free will for the purposes and consideration therein expressed.

Rachell S. Blaine

Notary or Consular Officer

[SEAL]



\*\*\*\*\*

Signature: Frederick B. Growcock Date: 7-1-2003  
 Printed Name: Frederick B. Growcock

COUNTY OF Harris )  
 STATE OF Texas ) ss.

BEFORE ME, the undersigned authority, on this 1<sup>st</sup> day of July, 2003, personally appeared Fred Growcock, known to me to be the person whose name is subscribed to the foregoing instrument and acknowledged to me that he executed the same of his own free will for the purposes and consideration therein expressed.

Rachell S. Blaine  
 Notary or Consular Officer

[SEAL]



\*\*\*\*\*

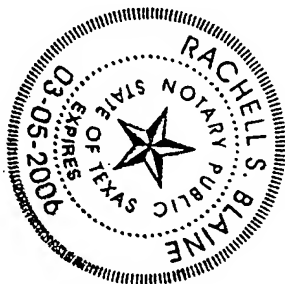
Signature: David Kippie Date: 7/1/03  
 Printed Name: David Kippie

COUNTY OF Harris )  
 STATE OF Texas ) ss.

BEFORE ME, the undersigned authority, on this 1<sup>st</sup> day of July, 2003, personally appeared David Kippie, known to me to be the person whose name is subscribed to the foregoing instrument and acknowledged to me that he executed the same of his own free will for the purposes and consideration therein expressed.

Rachell S. Blaine  
 Notary or Consular Officer

[SEAL]



\*\*\*\*\*

Signature: [Signature] Date: 1 July 2003  
 Printed Name: John W. Vian

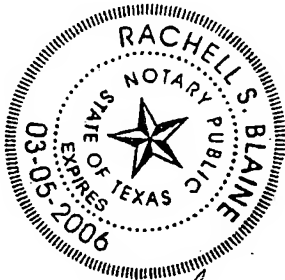
COUNTY OF Harris )  
 STATE OF Texas ) ss.

BEFORE ME, the undersigned authority, on this 1<sup>st</sup> day of July, 2003, personally appeared John Vian, known to me to be the person whose name is subscribed to the foregoing instrument and acknowledged to me that he executed the same of his own free will for the purposes and consideration therein expressed.

Rachell S. Blaine

Notary or Consular Officer

[SEAL]



\*\*\*\*\*

Signature: Kamila Abdur-Rahman Date: July 1, 2003  
 Printed Name: Kamila B. Abdur-Rahman

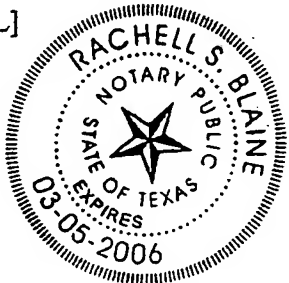
COUNTY OF Harris )  
 STATE OF Texas ) ss.

BEFORE ME, the undersigned authority, on this 1<sup>st</sup> day of July, 2003, personally appeared Kamila Abdur-Rahman known to me to be the person whose name is subscribed to the foregoing instrument and acknowledged to me that he executed the same of his own free will for the purposes and consideration therein expressed.

Rachell S. Blaine

Notary or Consular Officer

[SEAL]



\*\*\*\*\*

Signature: Morris Arvie Jr. Date: 7/1/03  
Printed Name: Morris Arvie, Jr.

COUNTY OF Harris )

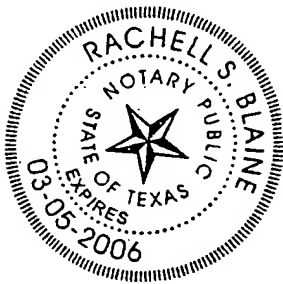
STATE OF Texas ) ss.

BEFORE ME, the undersigned authority, on this 1 day of July, 2003, personally appeared Morris Arvie, Jr., known to me to be the person whose name is subscribed to the foregoing instrument and acknowledged to me that he executed the same of his own free will for the purposes and consideration therein expressed.

Rachell S. Blaine

Notary or Consular Officer

[SEAL]



\*\*\*\*\*



## DECLARATION FOR PATENT APPLICATION

As a below named inventor, I hereby declare as evidenced by my signature that:

My residential address, mailing address and citizenship are as stated below next to my name,

I believe that the below named inventor(s) are the original, first and joint inventor(s) of the subject matter which is claimed and for which a patent is sought on the invention entitled

SURFACTANT-POLYMER COMPOSITIONS FOR ENHANCING THE STABILITY OF VISCOELASTIC-SURFACTANT BASED FLUID

the specification of which

☒ is attached hereto.

☐ was filed on \_\_\_\_\_ as

Application Serial No. \_\_\_\_\_

and was amended \_\_\_\_\_

(if applicable)

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose all information that may be material to the examination of this application in accordance with Title 37, Code of Federal Regulations, §1.56(a).

\*\*\*\*\*

PRIOR UNITED STATES PROVISIONAL APPLICATION(S):

☒ I hereby claim domestic priority benefits under Title 35, United States Code, §119(e) of any provisional application(s) for patent listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, §112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, §1.56(a) which occurred between the filing date of the prior provisional application and the national or PCT international filing date of this application:

60/ 319,575	September 25, 2002	copending
(Application Serial No.)	(Filing Date)	(Status)

\*\*\*\*\*

I hereby appoint the following, each associated with Customer Number 27551, to prosecute this application and to transact all business in the Patent and Trademark Office connected with this application:

STEPHEN H. CAGLE, Attorney (Reg. No. 26,445), CARTER J. WHITE, Attorney (Reg. No. 41374), each an attorney or agent with the law firm of HOWREY SIMON ARNOLD & WHITE, LLP, and each associated with Customer Number 27551, as its attorney or agent so long as they remain with such law firm.

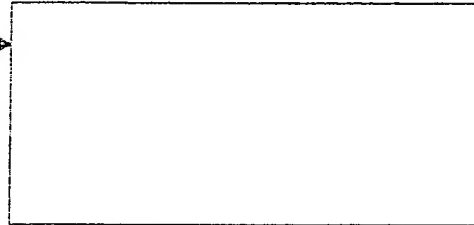
\*\*\*\*\*

I direct that:

All correspondence be directed to:

Customer No. 27551 Place label here →

HOWREY SIMON ARNOLD & WHITE, LLP  
750 Bering Dr.  
Houston, TX 77057



All telephone calls are to be directed to: Carter White, at telephone number: 713 268 1372.

\*\*\*\*\*

**I hereby declare, as evidenced by my signature below, that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under §1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.**

\*\*\*\*\*

First Inventor's signature

Robert L. Horton

07/01/2003  
(Date)

First Inventor's Full Name  
(Printed)

Robert L. HORTON

Residential Address

1103 Springdale Ct.  
Sugar Land, TX 77479  
USA

Mailing Address

Same as above

Citizenship

US

\*\*\*\*\*

Second Inventor's signature  
(if any)

Bethia B. Prasek 7-1-2003  
Date

Second Inventor's Full Name  
(Printed)

Bethicia  
Bethicia PRASEK

Residential Address

2215 Avendia La Quinta, Apt 903  
Houston, TX 77077  
USA

Mailing Address

Same as above

Citizenship

US

\*\*\*\*\*

Third Inventor's signature  
(if any)

Frederick B. Growcock 7-1-2003  
Date

Third Inventor's Full Name  
(Printed)

Frederick B. GROWCOCK

Residential Address

1514 Dodd Lane  
Houston, TX 77077  
USA

Mailing Address

Same as above

Citizenship

US

\*\*\*\*\*

Fourth Inventor's signature  
(if any)

David Kippie 7/1/03  
Date

Fourth Inventor's Full Name  
(Printed)

David KIPPIE

Residential Address

3610 Pedernales Trails Ln.  
Katy, TX 77450  
USA

Mailing Address

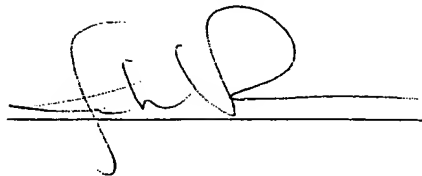
Same as above

Citizenship

US

\*\*\*\*\*

Fifth Inventor's signature  
(if any)



1 July 2003  
Date

Fifth Inventor's Full Name  
(Printed)

John W. VIAN

Residential Address

3317 Greenwood Pl.  
Deer Park, TX 77536  
USA

Mailing Address

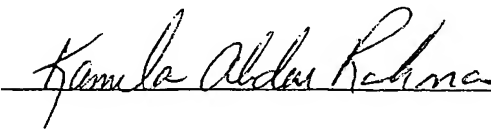
Same as above

Citizenship

US

\*\*\*\*\*

Sixth Inventor's signature  
(if any)



Date

Sixth Inventor's Full Name  
(Printed)

Kamila B. ABDUR-RAHMAN

Residential Address

12707 Bellaire Blvd. # 303  
Houston, TX 77072  
USA

Mailing Address

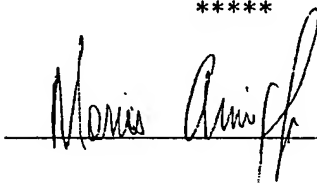
Same as above

Citizenship

USA

\*\*\*\*\*

Seventh Inventor's signature  
(if any)



7/1/03

Date

Seventh Inventor's Full Name  
(Printed)

Morris ARVIE, Jr.

Residential Address

11655 Briar Forest, #83  
Houston, TX 77077  
USA

Mailing Address

Same as above

Citizenship

US



Writer's Direct Dial:  
281-561-1451

Lbardin@miswaco.com

**Via Certified Mail**

September 21, 2005

Ms. Kamila B. Abdur-Rahman  
12707 Bellaire Blvd. #303  
Houston, TX 77072

Re: *U.S. Patent Application No. 10/620,041*  
*Entitled: Surfactant-Polymer Compositions for Enhancing the Stability of Viscoelastic-Surfactant Based Fluid*  
*Inventors: Robert Horton, Bethicia B. Prasek, Frederick Growcock, David P. Kippie, John Vian, Kamila Abdur-Rahman, Morris Arvie*  
*M-I L.L.C. Ref: PA-00146US*

Dear Ms. Abdur-Rahman:

We write you regarding the above-referenced U.S. application in which you are listed as an inventor. In July 2003, when the application was filed, you as well as the other inventors signed the Declaration. However, we have recently received a notice from the Patent Office that your signature is not dated, therefore they are requiring we submit a new declaration with a new signature and date.

We have enclosed a new Declaration for your execution as well as a copy of the application and original Declaration you signed. Please sign and date the new declaration and forward to us in the enclosed return FedEx envelope as soon as possible.

Please do not hesitate to contact me or Carter White at 281-561-1450 or [cjwhite@miswaco.com](mailto:cjwhite@miswaco.com) should you have any questions. We appreciate your help with this matter.

Regards,

Leah S. Bardin  
Patent Paralegal

Encl.

cc: Carter J. White, Ph.D.

- Item 4 if Restricted Delivery is desired.
- Print your name and address on the reverse so that we can return the card to you.
  - Attach this card to the back of the mailpiece, or on the front if space permits.

1. Article Addressed to:

Kamila B. Abdur-Rahman  
12707 Bellaire Blvd.  
#303  
Houston TX 77072

C. Signature

X

☐ Agent

☐ Addressee

D. Is delivery address different from Item 1?

☐ Yes

If YES, enter delivery address below:

☐ No

3. Service Type

☒ Certified Mail

☐ Express Mail

☐ Registered

☒ Return Receipt for Merchandise

☐ Insured Mail

☐ C.O.D.

4. Restricted Delivery? (Extra Fee)

☐ Yes

2. Article Number (Copy from service label)

7099 3400 0016 8879 5899

PS Form 3811, July 1999

Domestic Return Receipt

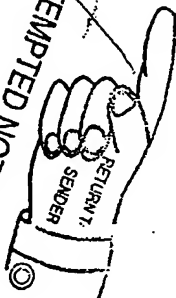
102595-00-M-0952

**MSWACO**

Legal Dept. - Patents  
5930 North Course Drive  
Houston, Texas 77072

Ms. Kamila B. Abdur-Rahman  
12707 Bellaire Blvd. #303  
Houston, TX 77072

ATTEMPTED NOT KNOWN



7099 3400 0016 8879 5899



*WVW*

I hereby certify that this correspondence is being deposited with the U.S. Postal Service as Express Mail, Airbill No. \_\_\_\_\_, in an envelope addressed to: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450, on the date shown below.

Dated: \_\_\_\_\_ Signature: \_\_\_\_\_ 0

**IN THE UNITED STATES PATENT AND TRADEMARK OFFICE**

**DECLARATION FOR PATENT APPLICATION**

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name.

I believe I am an original, first and joint inventor of the subject matter which is claimed and for which a patent is sought on the invention entitled:

**SURFACTANT-POLYMER COMPOSITIONS FOR ENHANCING THE STABILITY OF  
VISCOELASTIC-SURFACTANT BASED FLUID**

the specification of which was filed on July 15, 2003 as Application No. 10/620,041.

In the event that the filing date and/or Application No. are not entered above at the time I execute this document, and if such information is deemed necessary, I hereby authorize and request my attorneys/agent(s) at **Osha • Liang LLP**, 1221 McKinney St., Suite 2800, Houston, Texas 77010, to insert above the filing date and/or Application No. of said application.

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by an amendment, if any, specifically referred to herein.

I acknowledge the duty to disclose all information known to me that is material to patentability in accordance with Title 37, Code of Federal Regulations, § 1.56.

**FOREIGN PRIORITY CLAIM**

I hereby claim foreign priority benefits under Title 35, United States Code § 119(a)-(d) of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

☒ no such foreign applications have been filed

☐ such foreign application have been filed as follows:

**EARLIEST FOREIGN APPLICATION(S), IF ANY FILED WITHIN 12 MONTHS  
(6 MONTHS FOR DESIGN) PRIOR TO THIS U.S. APPLICATION**

Application Number	Country	Date of Filing	Priority Claimed Under 35 USC 119

**ALL FOREIGN APPLICATION(S), IF ANY FILED MORE THAN 12 MONTHS  
(6 MONTHS FOR DESIGN) PRIOR TO THIS U.S. APPLICATION**

Application Number	Country	Date of Filing

**CLAIM FOR BENEFIT OF EARLIER U.S. PROVISIONAL APPLICATIONS**

I hereby claim priority benefits under Title 35, United States Code §119(e), of any United States provisional patent application(s) listed below:

☐ no such U.S. provisional applications have been filed.

☒ such U.S. provisional application have been filed as follows:

Application Number	Date of Filing	Priority Claimed Under 35 USC 119
60/319,575	September 25, 2002	<input checked="" type="checkbox"/> Yes <input type="checkbox"/> No

**CLAIM FOR BENEFIT OF EARLIER U.S./PCT APPLICATION(S)**

I hereby claim the benefit under Title 35, United States Code, §120 of the United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, §112, I acknowledge the duty to disclose all information that is material to patentability in accordance with Title 37, Code of Federal Regulations, §1.56 which became available to me between the filing date of the prior application and the national or PCT international filing date of this application:

☒ no such U.S./PCT applications have been filed.

☐ such U.S./PCT application have been filed as follows:

Application Number	Relationship	Parent Application	Date of Filing


I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

Please mail all correspondence to Jeffrey S Bergman, whose address is:

**Osha · Liang LLP**  
 1221 McKinney St., Suite 2800  
 Houston, Texas 77010

Please direct telephone calls to: Jeffrey S. Bergman at (713) 228-8600.

Please direct facsimiles to: (713) 228-8778

Full name of sole or first inventor <b>Robert L. Horton</b>	
Sole or first inventor's signature	Date
Residence <b>Sugar Land, Texas</b>	
Citizenship <b>US</b>	
Mailing Address  <b>1103 Springdale Ct. Sugar Land, Texas 77479</b>	

Full name of second inventor, if any <b>Bethicia B. Prasek</b>	
Second inventor's signature	Date
Residence <b>Houston, Texas</b>	
Citizenship <b>US</b>	
Mailing Address  <b>2215 Avenida La Quinta, Apt. 903 Houston, Texas 77077</b>	

Full name of third inventor, if any <b>Frederick B. Growcock</b>	
Third inventor's signature	Date
Residence <b>Houston, Texas</b>	
Citizenship <b>US</b>	
Mailing Address  <b>1514 Dodd Lane Houston, Texas 77077</b>	

Full name of fourth inventor, if any <b>David P. Kippie</b>	
Fourth inventor's signature	Date
Residence <b>Katy, Texas</b>	
Citizenship <b>US</b>	
Mailing Address  <b>3610 Perdarnales Trails Lane Katy, Texas 77450</b>	

Full name of fifth inventor, if any <b>John W. Vian</b>	
Fifth inventor's signature	Date
Residence <b>Deer Park, Texas</b>	
Citizenship <b>US</b>	
Mailing Address  <b>3317 Greenwood Pl. Deer Park, Texas 77536</b>	

Full name of sixth inventor, if any <b>Kamila B. Abdur-Rahman</b>	
Sixth inventor's signature	Date
Residence <b>Houston, Texas</b>	
Citizenship <b>US</b>	
Mailing Address  <b>12707 Bellaire Blvd. #303 Houston, Texas 77072</b>	

Full name of seventh inventor, if any <b>Morris Arvie</b>	
Seventh inventor's signature	Date
Residence <b>Houston, Texas</b>	
Citizenship <b>US</b>	
Mailing Address  <b>11655 Briar Forest #83 Houston, Texas 77077</b>	

Full name of eighth inventor, if any	
Eighth inventor's signature	Date
Residence	
Citizenship	
Mailing Address	

**APPLICATION FOR UNITED STATES LETTERS PATENT**

**for**

**SURFACTANT-POLYMER COMPOSITIONS FOR ENHANCING THE STABILITY  
OF VISCOELASTIC-SURFACTANT BASED FLUID**

**by**

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1 This application claims the benefit of U. S. Provisional Application No. 60/319,575, filed  
2 September 25, 2002, the entire contents of which are incorporated by reference herein.

### 3 4 BACKGROUND

5 When drilling or completing wells in earth formations, various fluids typically are  
6 used in the well for a variety of reasons. For the purposes of the present disclosure, such  
7 a fluid will be referred to as a "well fluid." Common uses for well fluids include:  
8 lubrication and cooling of drill bit cutting surfaces while drilling generally or drilling-in  
9 (i.e., drilling into a targeted formation), transportation of cuttings to the surface,  
10 controlling formation pressure to prevent blowouts, maintaining well stability,  
11 suspending solids in the well, minimizing fluid loss into and stabilizing the formation  
12 through which a well is being drilled, fracturing the formation in the vicinity of a well,  
13 displacing the fluid within a well with another fluid, cleaning a well, testing a well,  
14 emplacing spacer or fluid loss control pills at various points in the displacement,  
15 completion, or work-over process, emplacing a packer fluid in the completed wellbore  
16 during production, preparing the well for abandonment, abandoning the well or,  
17 otherwise treating the well or the formation. A commonly used type of well fluid is  
18 based on water-based solutions including brines. Brines, such as  $\text{CaBr}_2$  brine, are  
19 commonly used as well fluids because of the ability to control the density of the solution  
20 over a wide density range. Further the brines are typically substantially free of suspended  
21 solids and brines typically do not damage the more common types of subterranean  
22 formations.

23 When drilling progresses to the level of penetrating a hydrocarbon-bearing  
24 formation, special care may be required to maintain the stability of the wellbore.  
25 Examples of formations in which problems often arise are highly permeable and/or  
26 poorly consolidated formation and thus a technique known as "under-reaming" may be  
27 employed. In conducting the under-reaming process, the wellbore is drilled to penetrate  
28 the hydrocarbon-bearing zone using conventional techniques. A casing generally is set in  
29 the wellbore to a point just above the hydrocarbon-bearing zone. The hydrocarbon-  
30 bearing zone then may be re-drilled to a wider diameter, for example, using an  
31 expandable under-reamer that increases the diameter of the wellbore. Under-reaming

usually is performed using special "clean" drilling fluids. Typically the "clean" drilling fluids used in under-reaming are aqueous, dense brines that are viscosified with a gelling and/or cross-linked polymer to aid in the removal of formation cuttings. The expense of such fluids limits their general use in the drilling process.

When the target subterranean formation has a high permeability a significant quantity of the drilling fluid may be lost into the formation. Once the drilling fluid is lost into the formation, it becomes difficult to remove. Removal of the aqueous based well fluids is desired to maximize the production of the hydrocarbon in the formation. It is well known in the art that calcium- and zinc-bromide brines can form highly stable, acid insoluble compounds when reacted with the formation rock itself or with substances contained within the formation. These reactions often may substantially reduce the permeability of the formation to any subsequent out-flow of the desired hydrocarbons. As should be well known in the art, it is widely and generally accepted that the most effective way to prevent such damage to the formation is to limit fluid loss into the formation. Thus, providing effective fluid loss control is highly desirable to prevent damaging the hydrocarbon-bearing formation. For example such damage may occur during, completion, drilling, drill-in, displacement, hydraulic fracturing, work-over, packer fluid emplacement or maintenance, well treating, or testing operations.

Techniques that have been developed to control fluid loss include the use of fluid loss control "pills." As the term is used in this disclosure a "pill" is a quantity of fluid added to the well fluid so as to temporarily change the properties of the well bore fluid at or near a specific point in the well bore. Significant research has been directed to determining suitable materials for the fluid loss pills, as well as controlling and improving the properties of the fluid loss pills. Typically, fluid loss pills work by enhancing filter-cake buildup on the face of the formation to inhibit fluid flow into the formation from the wellbore; however the fluids in accordance with the claimed subject matter are effective by developing extremely high viscosity in the environment at and just within the face of the formation to inhibit fluid flow into the formation from the wellbore. Because of the high temperatures, high shear (caused by the pumping and placement of the pill), high pressures, and low pH to which well fluids may be exposed (i.e., "stress

conditions”), synthetic polymeric materials typically used to form fluid loss pills and to viscosify the well fluids tend to degrade rather quickly.

One class of viscosifiers commonly used in the petroleum industry comprises polymeric structures starting with molecular weights of hundreds of thousands to several million grams per mole. These large, chemically bonded structures are often crosslinked to further increase molecular weight and effective viscosity per gram of polymer added to the fluid. These large molecules are quite stable under the thermal conditions typically encountered in a subterranean reservoir. However, this thermal stability is believed to contribute to decreased well productivity. As a result, expensive and often corrosive breakers have been designed to destroy the molecular backbone of these polymeric structures. These breakers are typically oxidizers or enzymes and are at best only partially effective with typical reservoir cleanup less than 80% complete and more usually much less than 50% complete. It is also reported in the literature that the long term stability of polymeric based thickening agents is shortened by the high temperature, high shear, high pressures, and low pH to which well fluids may be exposed (i.e., “stress conditions”).

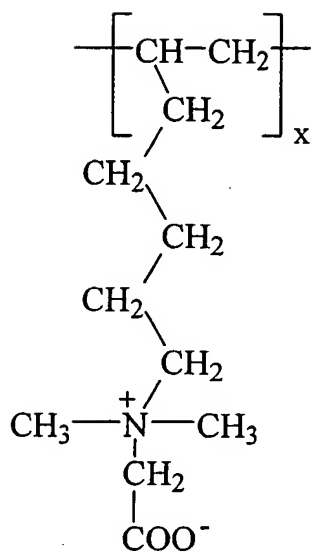
Viscoelastic surfactants are commonly used in the petroleum industry as an alternative to the above mentioned polymeric thickening agents. Viscoelastic surfactants are relatively small molecules with each molecule being typically less than 500 grams per mole (i.e., molecular weight less than 500). These small molecules will associate under certain conditions to form structures which resemble the polymer molecules but which are not stable structures. The individual molecules of surfactant begin to associate to form rod-like or spiraling-cylinder-like micelles. These micelle structures are always in an equilibrium state of breaking and reforming. As dynamic structures, these polymer-shaped micelles are readily destroyed by shear, presence of hydrocarbons or increased temperature. While these features are desirable especially in a hydrocarbon-bearing formation, there is minimal control over the conditions under which micelle breakup occurs. Therefore, under conditions of exposure to oil, high temperature, high shear, or other “stress conditions”, the viscoelastic surfactants rapidly return to their original small independent spherical micellar state. When the viscoelastic micelles are broken down to

1 this small independent spherical micellar state, the desired viscous nature of the well fluid  
2 is lost. In some cases the loss is temporary, in others the loss may be more permanent.

3  
4 Presently there exists an unmet need for a simple, inexpensive way to increase the  
5 thermal range for viscoelastic-surfactant-based viscosifying agents used in downhole  
6 applications. Preferably, this thermal extender would be applicable to various  
7 viscoelastic-surfactant-based viscosifying agents.

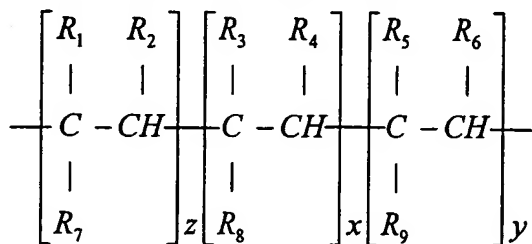
## 8 9 SUMMARY

10 Upon review of the following disclosed and claimed subject matter, one of skill in  
11 the art should appreciate and understand that one illustrative embodiment is a wellbore  
12 fluid that includes: an aqueous based continuous phase; a viscoelastic surfactant; and a  
13 surfactant-polymer compound soluble in an aqueous solution. The surfactant-polymer  
14 compound has a molecular structure including a hydrophobic backbone and a plurality of  
15 hydrophilic functional groups attached to the hydrophobic backbone. The hydrophobic  
16 backbone is the reaction product of one or more molecules having polymerizable alkene  
17 or alkyne functional groups, for example an oligo- or poly-ethylene structure. In contrast  
18 the hydrophilic functional groups can be zwitterionic surfactant functional groups,  
19 anionic surfactant functional groups, cationic surfactant functional groups, and nonionic  
20 surfactant functional groups. The illustrative embodiment is molecularly designed such  
21 that the combination of the viscoelastic surfactant and surfactant-polymer compound  
22 forms micellar assemblies in the wellbore fluid. In one illustrative and exemplary  
23 embodiment the acid form of the surfactant-polymer compound has the structure:

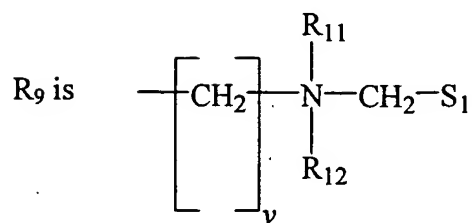
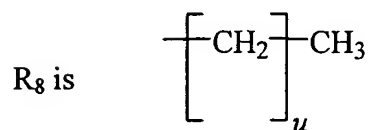
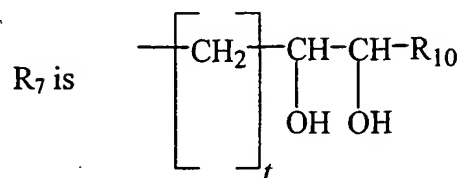


in which x has a value of 2 to 300,000 and preferably x has a value of 2 to 36  
 Alternatively, the surfactant-polymer compound can be a salt of oligo- or poly-( $\alpha$ -alkenyl  
 - $\omega$ - or  $\alpha$ -alkynyl - $\omega$ -quaternary-ammonio-N,N-dialkyl-N-alkylcarboxylate) or a mixture  
 further comprising a salt of N-alkyl-N-carboxymethyl-N,N-dimethylammonium chloride.  
 Another alternative and illustrative embodiment is where the surfactant-polymer  
 compound is a salt of oligo- or poly-(1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-  
 methylcarboxylate) or a mixture further comprising a salt of N-hexadecyl-N-  
 carboxymethyl-N,N-dimethylammonium chloride. A third alternative is where the  
 surfactant-polymer compound is a salt of oligo- or poly-( $\alpha$ -alkenyl - $\omega$ - or  $\alpha$ -alkynyl - $\omega$ -  
 quaternary-ammonio-N,N-dialkyl-N-alkylcarboxylate), or is a salt of oligo- or poly-(1-  
 hepten-7-quaternary-ammonio-N,N-dimethyl-N-methylcarboxylate).

Additional embodiments of the claimed subject matter include zwitterionic  
 surfactant heads such that the polymers or oligomers have the following structures:



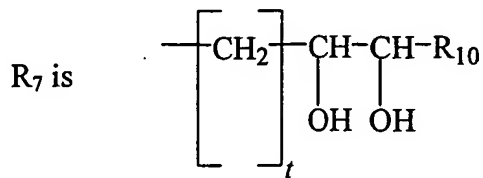
in which  $R_1, R_2, R_3, R_4, R_5, R_6 = H$  or  $CH_3$

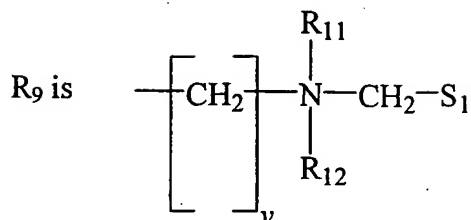
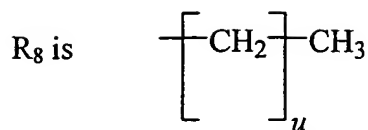


and in which  $R_{10}, R_{11}, R_{12} = H$  or  $CH_3$ , and  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  $3$ , and  $x + y + z = 3$  to  $300,000$  and  $S_1 = CO_2^-$  or  $SO_3^-$ . In a preferred illustrative embodiment,  $t = 12$  to  $16$ ,  $\mu = 6$  to  $12$ ;  $v = 12$  to  $18$ ,  $w = 1$  to  $3$ ,  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$  and  $z = 0$  to  $10,000$

Alternatively the oligomer or polymer compound can be cationic in the surfactant head and thus have a structure such as:

in which  $R_1, R_2, R_3, R_4, R_5, R_6 = H$  or  $CH_3$

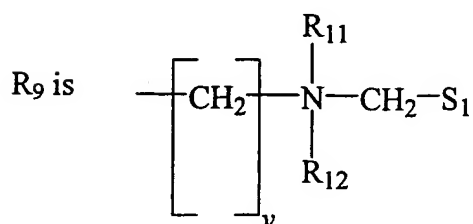
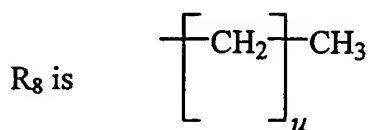
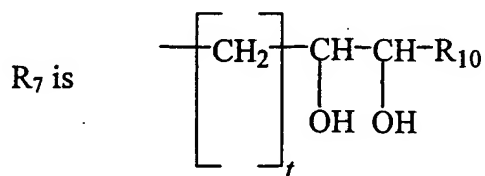




10  
11 and in which  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ , and  $x + y + z$   
12  $= 3$  to  $300,000$ . An especially preferred and illustrative embodiment includes an  
13 oligomer or polymer in which  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $w = 1$  to  $3$ , and  $x =$   
14  $0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

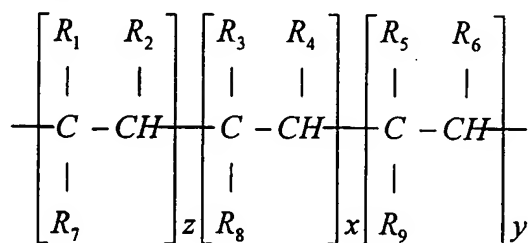
15 In yet another illustrative embodiment, the oligomer or polymer can have a  
16 molecular structure that includes an anionic surfactant functional group such as:

17  
18  
19 where  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$

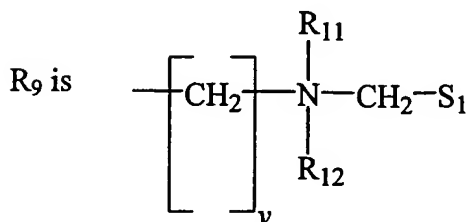
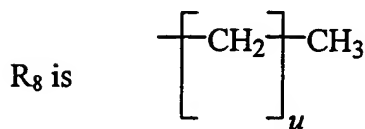
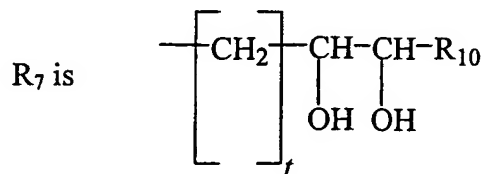


in which  $R_{10}, R_{11}, R_{12} = \text{H or CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $x + y + z = 3$  to  $300,000$ , and  $S_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ . In one such illustrative embodiment, it is preferred that  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

Further as noted above, the illustrative oligomer or polymer can have a nonionic surfactant group and preferably has a molecular structure such as:



in which  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H or CH}_3$

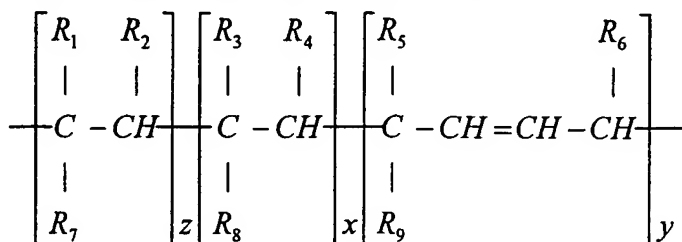


and in which,  $R_{10}, R_{11}, R_{12} = \text{H or CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  $12$ , and  $x + y + z = 3$  to  $300,000$ . In such instances, a preferred illustrative embodiment is



achieved when t = 12 to 16, u = 6 to 12, v = 12 to 18, w = 1 to 3, and x = 0 to 10,000, y = 2 to 300,000, and z = 0 to 10,000.

Fundamentally the polymeric backbone can be saturated as noted in the above illustrative examples or unsaturated. In such illustrative embodiments, the oligomer or polymer has a back bone structure such as the following:



in which  $R_1, R_2, R_3, R_4, R_5, R_6 = H$  or  $CH_3$ . As for the  $R_7, R_8$  and  $R_9$  groups, these may be the same as those disclosed above. Thus one of skill in the art should appreciate that zwitterionic, cationic, anionic and nonionic surfactant groups may be attached to the unsaturated backbone structure shown above and that such compounds are illustrative of the disclosed and claimed subject matter.

As discussed above, the novel oligomers and polymers taught in accordance with the claimed subject matter contain chemical functional groups that are similar to those found in conventional viscoelastic surfactants and thus are highly compatible with conventional viscoelastic surfactant systems. Accordingly, the claimed subject matter teaches blends of the viscoelastic oligomers or polymers with conventional viscoelastic surfactant systems. The conventional viscoelastic-surfactant-based fluids useful for the claimed subject matter are those in the following references, all of which are incorporated herein by reference – Canadian Patent 1,298,697, U. S. Patents 4,615,825, 4,695,389, 4,725,372, 5,258,137, 5,551,516, 5,691,292, 5,964,295, 5,965,502, 5,979,555, 5,979,557, 6,140,277, 6,194,355, 6,194,356, 6,211,120, 6,232,274, 6,239,183, Paper SPE 17,168, Paper SPE 30,098, Paper SPE 30,114, Paper SPE 30,458, Paper SPE 31,114, Paper SPE 38,622, Paper SPE 56,467, Paper SPE 57,432, Paper SPE 59,478, and Paper SPE 60,322. Of these, the preferred viscoelastic-surfactant-based fluids are those based on anionic, cationic, or zwitterionic surfactants or mixtures of anionic and nonionic surfactants or mixtures of cationic and nonionic surfactants or mixtures of zwitterionic and nonionic surfactants. And of these, the particularly preferred viscoelastic-surfactant-based fluids

1 are those based on zwitterionic surfactants or mixtures of zwitterionic and nonionic  
2 surfactants. In both the preferred viscoelastic-surfactant-based fluids, and in the  
3 particularly preferred viscoelastic-surfactant-based fluids, a minority amount of an  
4 additional surfactant, termed a "co-surfactant", such as, for example, 2-ethylhexanol or  
5 SDBS may optionally be employed. The viscoelastic oligomers or polymers of the  
6 claimed subject matter may be created in the presence of conventional viscoelastic  
7 surfactant systems or may be synthesized in separate steps, optionally processed or dried,  
8 and then subsequently admixed into a solution of conventional viscoelastic surfactants.

9 It has been found that the oligomers or polymers of the illustrative drilling fluids  
10 can be crosslinked with polyvalent metal ions, formaldehyde, or glutaraldehyde. In one  
11 such embodiment, the polyvalent metal ions are selected from:  $\text{Fe}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Ca}^{2+}$ ,  
12  $\text{Cu}^{2+}$ ,  $\text{UO}_2^{2+}$ ,  $\text{PbO}^{2+}$ ,  $\text{Al}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Cr}^{3+}$ ,  $\text{Ce}^{3+}$ ,  $\text{Ti}^{4+}$ ,  $\text{Zr}^{4+}$ ,  $\text{Sn}^{4+}$  and mixtures thereof.

13 One of skill in the art should also appreciate that the disclosed and claimed  
14 subject matter includes a method of making a wellbore fluid as is disclosed herein. In  
15 one such illustrative embodiment, the method involves the blending of an aqueous fluid  
16 phase, a viscoelastic surfactant, a water-soluble inorganic salt, and an oligomer or  
17 polymer soluble in the aqueous salt solution. The oligomer or polymer includes a  
18 hydrophobic oligomeric or polymeric backbone made from the oligomerization or  
19 polymerization of alkene or alkyne monomer groups, or mixtures thereof. The oligomer  
20 or polymer further comprises surfactant functional groups attached to the hydrophobic  
21 backbone, wherein the oligomer or polymer is hydrophilic in the surfactant functional  
22 groups and hydrophobic in the backbone hydrocarbon chain. Thus the oligomer or  
23 polymer has a molecular structure that promotes the formation of micellar assemblies  
24 such that the oligomers or polymers develop viscoelastic character prior to a  
25 polymerization step.

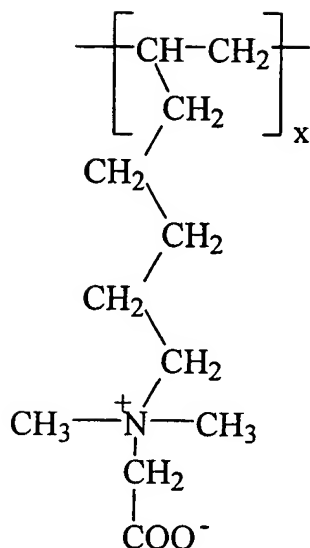
26 The claimed subject matter also encompasses a method of drilling a subterranean  
27 well as well as other uses for the wellbore fluid that should be apparent to one of skill in  
28 the art.

29 Further details regarding the claimed subject matter can be found in the following  
30 description.

## DESCRIPTION

The claimed subject matter relates to compositions for increasing the thermal durability of viscoelastic-surfactant-based well fluids. More specifically, the claimed subject matter relates to the fields of fluid rheology, thickeners, viscosifiers, viscoelastic fluids, viscoelastic surfactant fluids, drilling fluids, well fracturing fluids, well treatment fluids and fluid control pills. Further, the claimed subject matter relates to increasing the thermal and pressure stability of well fluids viscosified using viscoelastic surfactants by including an effective amount of a surfactant-oligomeric or surfactant-polymeric compound that has a thermally stable backbone structure and viscoelastic surfactant appendages. As the term is used in the present disclosure, "effective" simply means an amount sufficient to raise the temperature stability of the viscoelastic-surfactant based well fluid system by a measurable amount.

In accordance with one illustrative embodiment of the claimed subject matter, a sufficient quantity of at least one surfactant-oligomeric or surfactant-polymeric compound that is soluble in an aqueous salt solution is employed to affect the desired viscosity. In the claimed subject matter, the molecules of the surfactant-oligomeric or surfactant-polymeric compound have a hydrophobic oligomeric or polymeric backbone made preferably from the oligomerization or polymerization of alkene and/or alkyne groups. As the term is used herein, "thickener" and surfactant-oligomeric or surfactant-polymeric compound are used interchangeably and are intended to mean the compounds substantially described and claimed herein. The thickener of the claimed subject matter also includes chemical functional groups that are structurally similar to prior art viscoelastic surfactants and therefore these molecules exhibit similar chemical characteristics of prior art viscoelastic surfactants. Thus the hydrophobic backbone is chemically linked to and thus rendered at least in part hydrophilic by the presence of these chemical functional groups. One such illustrative compound is the product of the oligomerization reaction of a monomer such as the sodium salt of N-N-dimethyl-N-methylcarboxylate-N-1-hepten-7-ammonium chloride to give the sodium salt of oligo-(1-hepten-7-quaternary-ammonio-N-N-dimethyl-N-methylcarboxylate). The resulting oligomer is believed to have the simplified structure as indicated below in the acid form rather than the sodium-salt form:



in which x will have a value from about 2 to several hundred thousand, preferably from about 2 to several dozen. The monomer may be prepared, for example, by the reaction of N-hexadecyl-N,N-dimethylamine with chloroacetic acid to produce N-hexadecyl-N-methylcarboxylic acid-N,N-dimethylammonium chloride. Upon neutralization with sodium hydroxide, the final product is the zwitterionic betaine which is the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-dimethylammonium chloride – it has a negative charge on the carboxyl group and the sodium cation associated with it as a counter ion, and a positive charge on the quaternary amine group and the chloride anion associated with it as a counter ion. Alternatively, the sodium and chloride counter ions may be separated therefrom, leaving the negatively charged carboxyl group and the positively charged quaternary amine group as counter ions for each other.

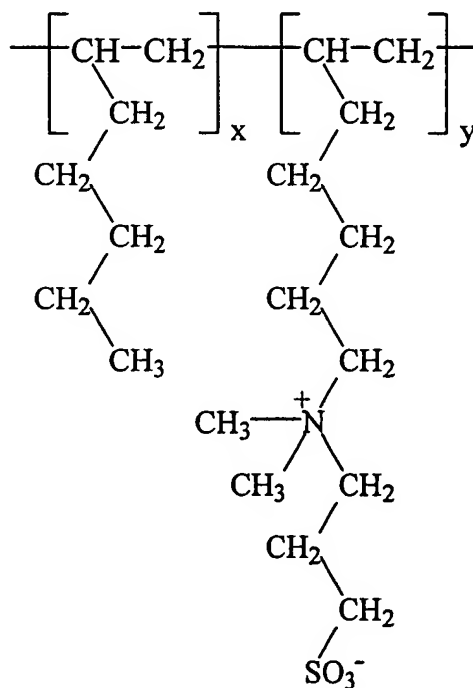
The sodium salt of oligo-(1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-methylcarboxylate) is an example of a salt of oligo- or poly-( $\alpha$ -alkenyl- $\omega$ -quaternary-ammonio-N,N-dialkyl-N-alkylcarboxylate). This example begins to define the terms  $\alpha$ -alkenyl - $\omega$ - and  $\alpha$ -alkynyl - $\omega$ -, wherein the “ $\alpha$ -“ designation denotes a location at or near one end of the hydrocarbonyl chain, such as, for example at the 1-, 2-, or 3- position and wherein the “ $\omega$ -“ designation denotes a location at or near the opposite end of the hydrocarbonyl chain from the  $\alpha$ -position, such as, for example at the very end of the hydrocarbonyl chain, at one carbon group away from the very end of the hydrocarbonyl chain,

1 or at two carbon groups away from the very end of the hydrocarbyl chain. An  $\alpha$ -alkene  
2 and an  $\alpha$ -alkyne are defined similarly. An  $\alpha$ -alkenyl- $\omega$ -carboxylate salt and an  $\alpha$ -  
3 alkynyl- $\omega$ -carboxylate salt are defined in a parallel manner. An  $\alpha$ -alkenyl- $\omega$ -N,N,N-  
4 trialkylammonium salt and an  $\alpha$ -alkynyl- $\omega$ -N,N,N-trialkylammonium salt are defined  
5 similarly. An  $\alpha,\alpha+2$ -alkadienyl- $\omega$ -carboxylate salt and an  $\alpha,\alpha+2$ -alkadiynyl- $\omega$ -  
6 carboxylate salt are defined in a parallel manner.

7 In one illustrative embodiment, a monomer like the sodium salt of N-  
8 carboxymethyl-N,N-dimethyl-N-1-hepten-7-ammonium chloride could be mixed into the  
9 solution of the conventional rod-shaped or spaghetti-like or spiraling-cylinder-like  
10 micelles of the viscoelastic surfactant that is the sodium salt of N-hexadecyl-N-  
11 carboxymethyl-N,N-dimethylammonium chloride. This illustrative monomer, the  
12 sodium salt of N-carboxymethyl-N,N-dimethyl-N-1-hepten-7-ammonium chloride,  
13 differs only in minor ways from the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-  
14 dimethylammonium chloride. Accordingly, the monomer will be readily subsumed into  
15 the conventional rod-shaped or spaghetti-like or spiraling-cylinder-like micelles,  
16 whereupon one can take well-known steps to initiate the oligomerization or  
17 polymerization of the monomer to produce the sodium salt of oligo- or poly-(1-hepten-7-  
18 quaternary-ammonio-N,N-dimethyl-N-methylcarboxylate). This oligomer or polymer  
19 will be inherently hydrophilic in its zwitterionic functional groups and hydrophobic in the  
20 hydrocarbon chains that link all the zwitterionic functional groups to each other. The  
21 oligomer or polymer is believed to be structurally quite similar to the viscoelastic  
22 surfactant molecules in the well fluid and therefore is soluble or dispersible in the well  
23 fluid solution. Similar oligomers or polymers are likewise soluble or dispersible in other  
24 viscoelastic surfactant solutions such as 10% XE862 (a product that is commercially  
25 available from Schlumberger) solution with 0.3% sodium dodecylbenzene sulfonate  
26 (SDBS) in 13.5 pound per barrel  $\text{CaBr}_2$ -based brine. Upon mixing the oligomer or  
27 polymer with the viscoelastic-surfactant-based fluid, the oligomer or polymer is believed  
28 to be subsumed into the rod-shaped or spaghetti-like micelles, whereupon, sitting inside  
29 these rod-shaped or spaghetti-like or spiraling-cylinder-like micelles is one or more such  
30 oligomer or polymer molecules and when configured in this fashion, the oligomer or  
31 polymer molecules impart greater thermal stability to the micelles, greater resistance to

shear stress and other stress conditions acting upon a fluid loss pill – including, for example, exposure to oil, high shear in pumping and placement, high temperature, high differential pressure, and low pH.

Another illustrative embodiment of the compounds of the claimed subject matter includes the oligomeric or polymeric products of the co-oligomerization or co-polymerization of two different monomers such as 1-heptene and the sodium salt of N-carboxymethyl-N,N-dimethyl-N-1-hepten-7-ammonium chloride. Alternatively, 1-heptene can be first co-oligomerized or co-polymerized with N,N-dimethyl-1-hepten-7-amine and in a subsequent reaction the amine groups are reacted with 1-chloro-propane-3-sulfonic acid and then caustic to produce oligo- or poly-(1-heptene-co-1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-propylsulfonate). The resulting oligomer or polymer is believed to have the simplified structure as indicated below:



in which x and y will have values dependent upon the molar ratio of reactants added to the polymerization reaction and the sum of x and y will have a value of about 2 to several hundred thousand, preferably from about 2 to several dozen. While this oligomer or polymer will be inherently hydrophilic in its zwitterionic functional groups, there are two sources of hydrophobicity: in the 1-heptene co-oligomer or co-polymer species and in the hydrocarbon chains that link all the zwitterionic functional groups to each other. By

1 varying the molar ratio of the monomers present during the oligomerization or  
2 polymerization reaction, the hydrophobicity and viscoelastic surfactant properties of the  
3 resulting oligomer or polymer may be controlled. The oligomer or polymer contains a  
4 point of structural similarity to the small molecule viscoelastic surfactants and therefore  
5 is soluble or dispersible in such conventional viscoelastic surfactant solutions. Similar  
6 oligomers or polymers are likewise soluble or dispersible in the 10% XE862 solution  
7 with 0.3% SDBS in 13.5 pounds per gallon  $\text{CaBr}_2$ -based brine discussed above. Upon  
8 admixing the polymer with the viscoelastic-surfactant-based fluid, the polymer is  
9 believed to be subsumed into the rod-shaped or spaghetti-like or spiraling-cylinder-like  
10 micelles, whereupon, sitting inside these rod-shaped or spaghetti-like or spiraling-  
11 cylinder-like micelles is one or more such polymer molecules and when configured in  
12 this fashion, the polymer molecules impart greater thermal stability to the micelles,  
13 greater resistance to shear stress and other stress conditions acting upon a fluid loss pill –  
14 including, for example, exposure to oil, high shear in pumping and placement, high  
15 differential pressure, and low pH.

16 Yet a third illustrative embodiment of the compounds of the claimed subject  
17 matter includes the co-oligomerization or co-polymerization of three or more co-  
18 monomers. For example, one could co-oligomerize or co-polymerize 1-heptene, 1-  
19 heptene-6,7-diol and N,N-dimethyl-1-hepten-7-amine. In such an illustrative  
20 embodiment, the amine groups are reacted with 1-chloro-propane-3-sulfonic acid and  
21 then caustic to produce oligo- or poly- (1-heptene -co-1-heptene-6,7-diol -co-1-hepten-7-  
22 quaternary-ammonio-N,N-dimethyl-N-propylsulfonate). The resulting oligomer or  
23 polymer is believed to have the simplified structure as indicated below:  
24



7



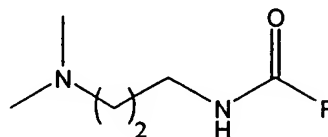
1 vicinal diol functionality to present itself at the outer surface of any rod-shaped micelle in  
2 which it has become subsumed.

3 The vicinal diol functionality provides the chemical functional group that renders  
4 the oligomers or polymers readily crosslinkable with polyvalent metal ions or complexes  
5 such as, for example, a borate, titanate, or zirconate crosslinker as taught in U. S. Patent  
6 5,062,969, (2) divalent, trivalent, or tetravalent cations such as, for example,  $\text{Fe}^{2+}$ ,  $\text{Cd}^{2+}$ ,  
7  $\text{Co}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{UO}_2^{2+}$ ,  $\text{PbO}^{2+}$ ,  $\text{Al}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Cr}^{3+}$ ,  $\text{Ce}^{3+}$ ,  $\text{Ti}^{4+}$ ,  $\text{Zr}^{4+}$ ,  $\text{Sn}^{4+}$ , and the like, (3)  
8 complexes of or other moieties containing the crosslinkers listed above in the first two  
9 categories, such as, for example, the tetrammine complex of the  $\text{Cu}^{2+}$  cation, the  
10 carbonate anion complexes of the  $\text{UO}_2^{2+}$  cation,  $\text{UO}_2(\text{CO}_3)_2^{2-}$  and  $\text{UO}_2(\text{CO}_3)_3^{4-}$ , or the  
11 triethanolamine complex of the  $\text{Ti}^{4+}$  cation, (4) so-called "organic crosslinkers" such as,  
12 for example, formaldehyde, and glutaraldehyde, and (5) mixtures of the crosslinkers  
13 listed above in the first four categories and/or reaction products therefrom. Thus it is  
14 contemplated that one of skill in the art could, if desired, crosslink the illustrative  
15 oligomers or polymers with polyvalent metal ions, complexes, organic crosslinkers, or  
16 mixtures thereof, as described above. This is believed to lead effectively to the  
17 crosslinking of the viscoelastic surfactant assemblies in which the oligomers or polymers  
18 have been subsumed.

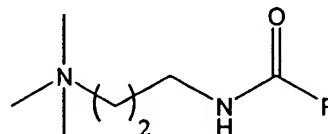
19 One of skill in the art should at this point appreciate that the co-oligomerization or  
20 co-polymerization reactions disclosed above result in the random placement of each co-  
21 monomer along the vinyl backbone of the illustrative oligomers or polymers. However,  
22 it is contemplated that block co-oligomerization or co-polymerization could also be used  
23 to achieve substantially similar results. Alternatively, block co-oligomerization or co-  
24 polymerization could also be used to carefully tailor the properties of the resulting  
25 oligomers or polymers made in accordance with the claimed subject matter. One of skill  
26 in the art should understand and appreciate that by systematically controlling the molar  
27 ratio of and concentration of monomers present in during the oligomerization or  
28 polymerization process, the order of addition, the temperature and duration of  
29 oligomerization or polymerization and the initiators and catalysts used and their  
30 concentrations, the properties of the compounds of the claimed subject matter can be  
31 carefully controlled and tailored.

Other oligomers or polymers formulated in accordance with and illustrative of the claimed subject matter include such oligomers or polymers as oligo- or poly-(N-carboxymethyl-N,N-diallyl-N-methyl ammonium chloride), oligo- or poly-(N,N-diallyl-N,N-dimethyl ammonium chloride-co-N-carboxymethyl-N,N-diallyl-N-methyl ammonium chloride), oligo- or poly-(1-butene-co-N-carboxymethyl-N,N-diallyl-N-methyl ammonium chloride), oligo- or poly- (1-butene-co-1-pentene-4,5-diol- co-N-carboxymethyl-N,N-diallyl-N-methyl ammonium chloride), oligo- or poly-alkenyl or -alkynyl DMAPA amides (see Figure below), oligo- or poly-alkenyl or -alkynyl DMAPA quats, and oligo- or poly-alkenyl or -alkynyl tallow amine quaternary amines such as

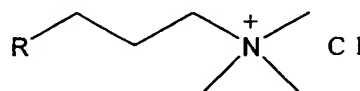
C<sub>12-22</sub> DMAPA Amide



C<sub>12-22</sub> DMAPA Amide Quaternary Amine



Tallow Amine Quaternary Amine



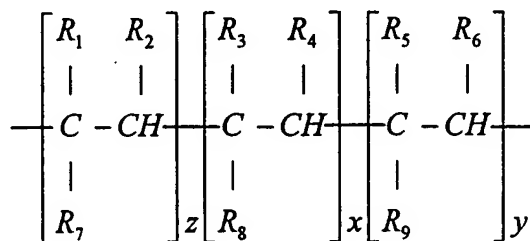
and the like.

Except for the oligo- or poly-alkenyl or -alkynyl DMAPA amides, oligo- or poly-alkenyl or -alkynyl DMAPA quaternary amines, and oligo- or poly-alkenyl or -alkynyl tallow amine quaternary amines, oligomers or polymers of this type may be unlike those of the previous illustrative embodiments of the claimed subject matter in that they are probably not capable of effectively vacuuming up the micelles and packing them about themselves until there are enough surfactant molecules present in the vicinity of the oligomer or polymer molecule so that the spherical micelles can merge into a single rod-shaped or spaghetti-like or spiraling-cylinder-like micelles with the oligomer or polymer molecule subsumed within the rod-shaped or spaghetti-like or spiraling-cylinder-like micelles. Polymers or oligomers in accordance with the present illustrative embodiment

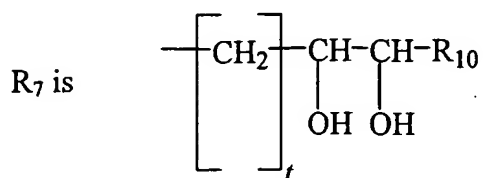
of the claimed subject matter are designed to be compatible with  $\text{CaBr}_2$ -,  $\text{CaBr}_2/\text{CaCl}_2$ -,  $\text{ZnBr}_2/\text{CaBr}_2$ -, and  $\text{ZnBr}_2/\text{CaBr}_2/\text{CaCl}_2$ -based brines.

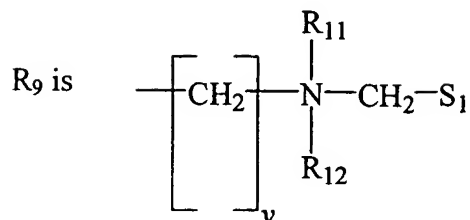
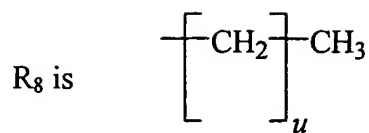
Other oligomers or polymers formulated in accordance with and illustrative of the claimed subject matter include oligomers or polymers such as oligo- or poly-(ethylene-co-N,N-diallyl-N,N-dimethyl ammonium chloride), oligo- or poly-(1-butene-co-N,N-diallyl-N,N-dimethyl ammonium chloride), oligo- or poly-(1-butene-co-1-pentene-4,5-diol-co-N,N-diallyl-N,N-dimethyl ammonium chloride), and the like. Polymers or oligomers of this type are probably not sufficiently compatible with  $\text{CaBr}_2$ -,  $\text{CaBr}_2/\text{CaCl}_2$ -,  $\text{ZnBr}_2/\text{CaBr}_2$ -, and  $\text{ZnBr}_2/\text{CaBr}_2/\text{CaCl}_2$ -based brines; however, they expand the scope of the claimed subject matter into useful dimensions to the extent that these polymers should be quite compatible with  $\text{NaCl}$ -,  $\text{NaBr}$ -,  $\text{NaBr}/\text{NaCl}$ -,  $\text{CsBr}/\text{KBr}$ -, and  $\text{CsBr}/\text{KBr}/\text{NaCl}$ -based brines and the like. Poly- or oligo-(1-butene-co-1-pentene-4,5-diol-co-N,N-diallyl-N,N-dimethyl ammonium chloride) incorporates a vicinal diol functionality, and so one of skill in the art should appreciate that this polymer or oligomer may be crosslinkable with polyvalent metal ions as disclosed above. As noted above, when the oligomers or polymers are crosslinked with polyvalent metal ions, we also effectively crosslink the viscoelastic surfactant assemblies in which the oligomers or polymers have been subsumed.

Additional embodiments of the claimed subject matter include zwitterionic surfactant heads such that the polymers or oligomers have the following structures:



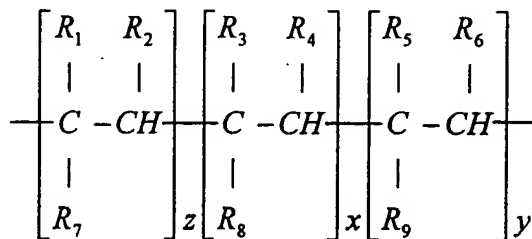
in which  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



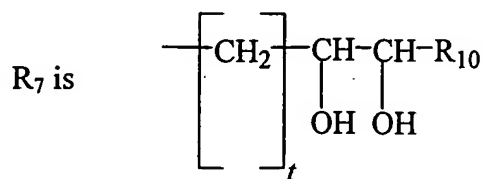


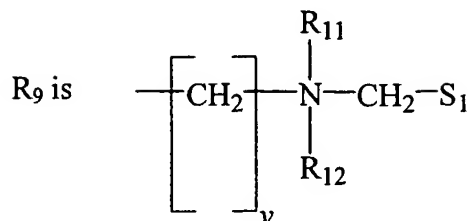
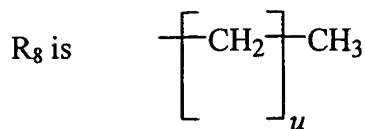
and in which  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ , and  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  $3$ , and  $x + y + z = 3$  to  $300,000$  and  $\text{S}_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ . In a preferred illustrative embodiment,  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ;  $v = 12$  to  $18$ ,  $w = 1$  to  $3$ ,  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$  and  $z = 0$  to  $10,000$

Alternatively the oligomer or polymer compound can be cationic in the surfactant head and thus have a structure such as:



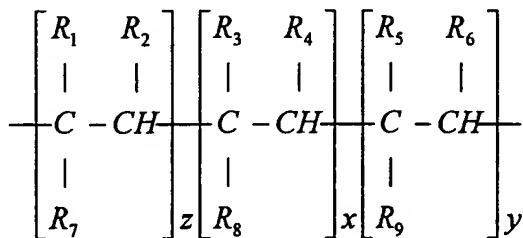
in which  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



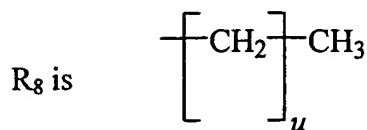
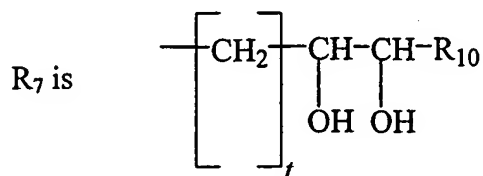


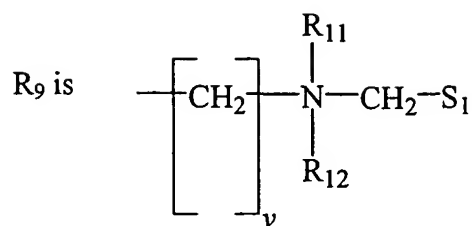
and in which  $R_{10}, R_{11}, R_{12} = \text{H or CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ , and  $x + y + z = 3$  to  $300,000$ . An especially preferred and illustrative embodiment includes an oligomer or polymer in which  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $w = 1$  to  $3$ , and  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

In yet another illustrative embodiment, the oligomer or polymer can have a molecular structure that includes an anionic surfactant functional group such as:



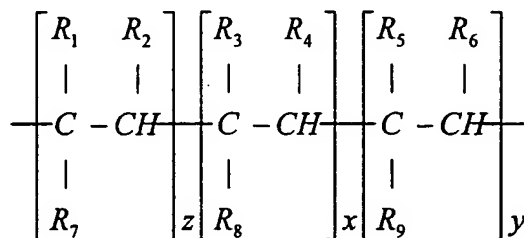
where  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H or CH}_3$



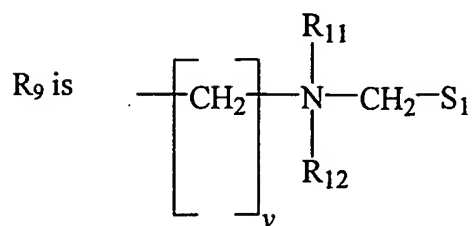
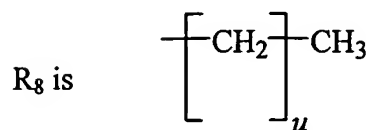
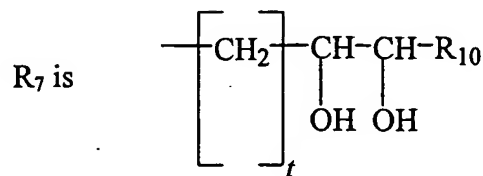


9 in which  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $x + y + z = 3$  to  
10  $300,000$ , and  $\text{S}_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ . In one such illustrative embodiment, it is preferred that  $t$   
11  $= 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  
12  $10,000$ .

13 Further as noted above, the illustrative oligomer or polymer can have a nonionic  
14 surfactant group and preferably has a molecular structure such as:

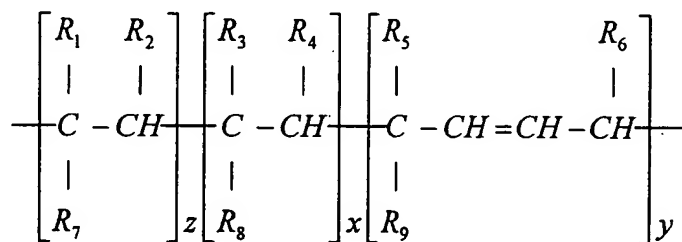


15  
16  
17 in which  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



and in which,  $R_{10}, R_{11}, R_{12} = \text{H or CH}_3$ ,  $t = 1$  to 16,  $u = 6$  to 12,  $v = 1$  to 18,  $w = 1$  to 12, and  $x + y + z = 3$  to 300,000. In such instances, a preferred illustrative embodiment is achieved when  $t = 12$  to 16,  $u = 6$  to 12,  $v = 12$  to 18,  $w = 1$  to 3, and  $x = 0$  to 10,000,  $y = 2$  to 300,000, and  $z = 0$  to 10,000.

Fundementally the polymeric backbone can be saturated as noted in the above illustrative examples or unsaturated. In such illustrative embodiments, the oligomer or polymer has a back bone structure such as the following:



in which  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H or CH}_3$ . As for the  $R_7, R_8$  and  $R_9$  groups, these may be the same as those disclosed above. Thus one of skill in the art should appreciate that zwitterionic, cationic, anionic and nonionic surfactant groups may be attached to the unsaturated backbone structure shown above and that such compounds are illustrative of the disclosed and claimed subject matter.

As discussed above, the novel oligomers and polymers taught in accordance with the claimed subject matter contain chemical functional groups that are similar to those found in conventional viscoelastic surfactants and thus are highly compatible with conventional viscoelastic surfactant systems. Accordingly, the claimed subject matter teaches blends of the viscoelastic oligomers or polymers with conventional viscoelastic surfactant systems. The conventional viscoelastic-surfactant-based fluids useful for the claimed subject matter are those in the following references, all of which are incorporated herein by reference – Canadian Patent 1,298,697, U. S. Patents 4,615,825, 4,695,389, 4,725,372, 5,258,137, 5,551,516, 5,691,292, 5,964,295, 5,965,502, 5,979,555, 5,979,557, 6,140,277, 6,194,355, 6,194,356, 6,211,120, 6,232,274, 6,239,183, Paper SPE 17,168,

1 Paper SPE 30,098, Paper SPE 30,114, Paper SPE 30,458, Paper SPE 31,114, Paper SPE  
2 38,622, Paper SPE 56,467, Paper SPE 57,432, Paper SPE 59,478, and Paper SPE 60,322.  
3 Of these, the preferred viscoelastic-surfactant-based fluids are those based on anionic,  
4 cationic, or zwitterionic surfactants or mixtures of anionic and nonionic surfactants or  
5 mixtures of cationic and nonionic surfactants or mixtures of zwitterionic and nonionic  
6 surfactants. And of these, the particularly preferred viscoelastic-surfactant-based fluids  
7 are those based on zwitterionic surfactants or mixtures of zwitterionic and nonionic  
8 surfactants. In both the preferred viscoelastic-surfactant-based fluids, and in the  
9 particularly preferred viscoelastic-surfactant-based fluids, a minority amount of an  
10 additional surfactant, termed a "co-surfactant", such as, for example, 2-ethylhexanol or  
11 SDBS may optionally be employed. The viscoelastic oligomers or polymers of the  
12 claimed subject matter may be created in the presence of conventional viscoelastic  
13 surfactant systems or may be synthesized in separate steps, optionally processed or dried,  
14 and then subsequently admixed into a solution of conventional viscoelastic surfactants.

15 The aqueous well fluids made in accordance with the claimed subject matter may  
16 optionally include a sufficient quantity of at least one water-soluble inorganic salt to  
17 effect formation stability. Typically, water-soluble potassium and ammonium salts, such  
18 as potassium chloride and ammonium chloride are employed. However, other shale  
19 inhibition compounds may be utilized including organic amine based compounds and  
20 other known shale inhibition agents. Additionally, calcium chloride, calcium bromide  
21 and zinc halide salts may also be used to increase the specific gravity (i.e., the density) of  
22 the solution. Formation stability and in particular clay stability are achieved at a  
23 concentration of a few percent by weight and as such the density of the fluid is not  
24 significantly altered by the presence of the inorganic salt unless fluid density becomes an  
25 important consideration, at which point, heavier inorganic salts may be employed.

26 Other compounds useful in the claimed subject matter include other viscosifiers,  
27 corrosion inhibitors, lubricants, pH control additives, surfactants, solvents, and/or  
28 weighting agents, among other additives. Some typical brine-based well fluid  
29 viscosifying additives include "natural" or biopolymers or derivatives thereof, such as,  
30 for example, xanthan gum and hydroxyethyl cellulose (HEC) or synthetic polymers and  
31 oligomers such as poly(ethylene glycol) (PEG), poly(diallyl amine), poly(acrylamide),



1 poly(aminomethylpropylsulfonate [AMPS]), poly(acrylonitrile), poly(vinyl acetate),  
2 poly(vinyl alcohol), poly(vinyl amine), poly(vinyl sulfonate), poly(styryl sulfonate),  
3 poly(acrylate), poly(methyl acrylate), poly(methacrylate), poly(methyl methacrylate),  
4 poly(vinylpyrrolidone), poly(vinyl lactam) and co-, ter-, and quater-polymers of the  
5 following co-monomers: ethylene, butadiene, isoprene, styrene, divinylbenzene, divinyl  
6 amine, 1,4-pentadiene-3-one (divinyl ketone), 1,6-heptadiene-4-one (diallyl ketone),  
7 diallyl amine, ethylene glycol, acrylamide, AMPS, acrylonitrile, vinyl acetate, vinyl  
8 alcohol, vinyl amine, vinyl sulfonate, styryl sulfonate, acrylate, methyl acrylate,  
9 methacrylate, methyl methacrylate, vinylpyrrolidone, and vinyl lactam. Yet other  
10 viscosifiers include the clay-based viscosifiers, especially laponite and other small  
11 fibrous clays such as the polygorskites (attapulgite and sepiolite).

12 One of ordinary skill in the art should appreciate that the use of "inert" filler  
13 materials can be added to impart strength to a fluid. Examples of such materials include  
14 shredded rubber tires, shredded battery casings, peanut hulls, cotton seed hulls, woody  
15 material, and other plant fibers that should be well known to one of skill in the art.

16 Additional brine-based well fluid surfactant additives useful in the claimed  
17 subject matter include nonionic surfactants, such as ethoxylated nonylphenols containing  
18 about 6 to 20 moles of ethylene oxide, or alkyl polyethyleneoxyalcohols, ethoxylated  
19 linear alcohols, ethoxylated tridecyl alcohols, ethoxylated phenols such as nonylphenols  
20 and dodecylphenols and fatty dialkanol amides. Hydroxyethyl fatty amines are also  
21 classified as nonionic surfactants, although at low pH, they may take on some cationic  
22 character. Emulsifying surfactants including oil soluble surfactants, such as fatty  
23 diethanolamides, sorbitan fatty acid esters, and ethoxylated sorbitan fatty acid esters such  
24 as sorbitan monooleate and sorbitan sesquioleate, because of their limited solubility in  
25 typical brines, may however be incidentally included in the formulation of other products  
26 – polymer solutions, emulsions, or slurries, corrosion inhibitors, lubricants, solvents, or  
27 weighting agents – that are used as additives to brine-based well fluids.

28 Other surfactants useful in the claimed subject matter are those listed in  
29 "McCutcheon's Emulsifiers and Detergents 1999: North American Edition" (ISBN:  
30 0944254624), incorporated herein by reference. They are classified as anionic, nonionic,  
31 amphoteric, zwitterionic, alcohols, alkanolamides, alkanolamines, alkylaryl sulfonates,

1 alkylaryl sulfonic acids, amine acetates, amine oxides, amines, sulfonated amines,  
2 sulfonated amides, betaine derivatives, block polymers, carboxylated alcohols,  
3 alkylphenol ethoxylates, carboxylic acids, fatty acids, ethoxylated alcohols, ethoxylated  
4 alkylphenols, ethoxylated amines, ethoxylated amides, ethoxylated fatty acids,  
5 ethoxylated fatty esters, fatty esters, fluorocarbon-based surfactants, glycerol esters,  
6 glycol esters, heterocyclic surfactants, imidazolines, imidazoline derivatives, isethionates,  
7 lanolin-based derivatives, lechithin, lechithin derivatives, methyl esters, monoglycerides,  
8 monoglyceride derivatives, olefin sulfonates, phosphate esters, phosphorous organic  
9 derivatives, polyethylene glycols, polymeric surfactants (polysaccharides, polyacrylic  
10 acids, polyacrylamides), propoxylated alcohols, propoxylated alkylphenols, propoxylated  
11 amines, propoxylated amides, propoxylated fatty acids, propoxylated fatty esters, protein-  
12 based surfactants, quaternary surfactants, sarcosamine derivatives, silicone-based  
13 surfactants, soaps, sodium isethionate, sorbitan derivatives, sucrose and glucose esters  
14 and derivatives, sulfates and sulfonates of oils and fatty acids, sulfates and sulfonates of  
15 ethoxylated alkylphenols, sulfates of alcohols, sulfates of ethoxylated alcohols, sulfates  
16 of fatty esters, sulfonates of hydrocarbons and petroleum, sulfosuccinates and derivatives,  
17 taurates, thio and mercapto derivatives, and tridecyl and dodecyl benzene sulfonic acids.

18 As mentioned earlier, the claimed subject matter concentrates on those micellar  
19 assemblies that are dominated by (1) one-dimensional extensions of the spherical micelle  
20 into rod-shaped or spaghetti-like or spiraling-cylinder-like micelles, (2) two-dimensional  
21 extensions of the spherical micelle into planar micelles or, more likely, stacks of planar  
22 micelles (as in liquid crystals), or (3) vesicular micelles. Vesicular micelles may be  
23 spherical at their very centers, but each center sphere will be surrounded by a second,  
24 third, and possibly more layers of surfactants. Moving outward radially from the central  
25 sphere of surfactant molecules, each successively outer layer of surfactant molecules will  
26 be oriented oppositely from next inner-most layer. Vesicular micelles may also be rod-  
27 shaped at their very centers, but each central rod-shaped assembly will be surrounded by  
28 a second, third, and possibly more layers of surfactants. A stack of planar micelles may  
29 be thought of as the vesicular form of a single planar micelle. These concepts comprise  
30 yet another illustrative embodiment of the claimed subject matter.

1       The above-mentioned embodiments may also comprise oligomers or polymers  
2       created by first carrying out a micellar polymerization to produce an oligomer or  
3       polymer, and then optionally processing or drying the product, and then subsequently  
4       admixing it into a solution of conventional surfactants – (1) non-viscoelastic or (2) non-  
5       viscoelastic and viscoelastic in nature. The micelles of the conventional surfactants  
6       would, in preferred instances, subsume the oligomers or polymers into the conventional  
7       surfactant micelles to create mixed micelles having superior viscoelastic properties in  
8       comparison to those of equivalent solutions not comprising the conventional surfactants  
9       and/or in comparison to those of equivalent solutions not comprising the oligomers or  
10      polymers.

11       It should also be appreciated by one of skill in the art that the claimed subject  
12      matter also relates to methods of using fluid loss control pills and similar fluids  
13      containing the compounds of the claimed subject matter, that can sustain stress conditions  
14      for extended periods of time without significant fluid loss or loss of desirable rheological  
15      properties. The stress conditions may include, for example, exposure to oil, high shear in  
16      pumping and placement, exposure to oxidizing breakers (including oxygen dissolved in  
17      the fluid), exposure to brines having high divalent cation content, high temperature, high  
18      differential pressure, low pH, extended time, and a combination of two or more of such  
19      stress conditions. These pills and fluids are advantageously applied in or in connection  
20      with drilling, drill-in, displacement, completion, hydraulic fracturing, work-over, packer  
21      fluid emplacement or maintenance, well treating, testing, or abandonment.

22       Exemplary viscoelastic surfactant compositions of improved stability in fluid loss  
23      control in accordance with the claimed subject matter are given in the following  
24      examples. The following examples are included to demonstrate preferred embodiments  
25      of the claimed subject matter. It should be appreciated by those of skill in the art that the  
26      techniques disclosed in the examples which follow represent techniques discovered by  
27      the inventors to function well in the practice of the claimed subject matter, and thus can  
28      be considered to constitute preferred modes for its practice. However, those of skill in  
29      the art should, in light of the present disclosure, appreciate that many changes can be  
30      made in the specific embodiments which are disclosed and still obtain a like or similar  
31      result without departing from the scope of the claimed subject matter.

**Example 1:** The oligomerization of the surfactant N-dodecene-1-yl-N,N-bis(2-hydroxyethyl)-N-methylammonium chloride:



in an aqueous solution was achieved by joining a few of the carbon-carbon double bonds by a free radical oligomerization reaction within the micelles using the following processes. A viscoelastic surfactant solution is produced using 30 g/l of the surfactant N-dodecene-1-yl-N,N-bis(2-hydroxyethyl)-N-methylammonium chloride with 40 g/l ammonium chloride. A volume of 100 ml of the viscoelastic surfactant solution was placed in a bottle purged with oxygen-free, dry nitrogen gas to remove any dissolved oxygen. After sufficient purging 10 mg of the free radical initiator 2,2'-azo(bis-amidinopropane) dihydrochloride was added to the viscoelastic surfactant solution and mixed thoroughly. The surfactant solution was heated at 60°C for 30 minutes under an atmosphere of oxygen-free, dry nitrogen. The resulting oligomer can be thought of as related to oligo-ethylene, a relatively short-chained polyethylene, to which relatively long pendant surfactant groups -  $[(\text{CH}_2)_{18}][\text{HO}(\text{CH}_2)_2]_2\text{CH}_3\text{N}^+\text{Cl}^-$  - are linked. Oligomerization of the surfactant monomers in micelles resulted in the viscosity of the gel becoming largely insensitive to contact with hydrocarbon. The viscosity of the surfactant gel was not materially altered by the oligomerization of the surfactant monomers. The oligomerized surfactant gel retained its gel strength after prolonged contact with water. A volume of 50 ml of the oligomerized viscoelastic surfactant solution was blended with 50 ml of the original un- oligomerized viscoelastic surfactant solution. The resulting solution was blended with 8 volume % of n-hexane with no material reduction in the viscosity of the mixture being observed, whereas mixing only 6 volume % of n-hexane with the original un-oligomerized viscoelastic surfactant solution led to loss of essentially all viscosity in the mixture. Advantageously, therefore, compositions formed in accordance with the claimed subject matter provide viscoelastic surfactant fluids that are capable of controlling fluid loss, and that are capable of dissolving in fluid without leaving substantial amounts of residue. Further, the compositions are capable of controlling fluid loss at temperatures up to about 350°F or

higher, and may also control fluid loss for an extended period of at least 4 days, and finally, do not interfere with other chemical additives commonly used in the petroleum industry.

**Example 2:** The oligomerization of the surfactant potassium octadec-1-ene-18-oate:



in an aqueous solution was achieved using the following processes. The viscoelastic surfactant solution was formed by mixing 60 g/l potassium oleate with 60 g/l potassium chloride. A sample of 100 ml of the viscoelastic surfactant solution was purged with oxygen-free, dry nitrogen and mixed with 10 mg of the initiator 2,2'-azo(bisamidinopropane) dihydrochloride. The solution is heated at 60°C for 30 minutes under an atmosphere of oxygen-free, dry nitrogen. The resulting solution of oligomerized surfactants was about equally viscoelastic as the original monomeric solution but the observed viscoelasticity was insensitive to contact with hydrocarbon. The resulting oligomer can be thought of as related to oligo-ethylene, a relatively short-chained polyethylene, to which relatively long pendant groups –  $(\text{CH}_2)_{16}\text{CO}_2\text{K}^+$  – are linked. The gel formed by the oligomerized surfactant retained its viscoelasticity after prolonged contact with water. The observations led to the conclusion that the oligomer was viscoelastic in character, much as was the original un-oligomerized viscoelastic surfactant solution; but the oligomerized solution was more stable than the original un-oligomerized solution.

**Example 3:** The oligomerization of a long-chain vinyl surfactant, the potassium salt of octadeca-1,3-diene-17-oate:



in a viscoelastic solution was carried out at a concentration of 60 g/l in the presence of 40 g/l ammonium chloride. The surfactant monomers were oligomerized using 10 mg of the free radical initiator 2,2'-azo(bis-amidinopropane) dihydrochloride in 100 ml of viscoelastic surfactant solution which had been purged with oxygen-free, dry nitrogen

gas. The solution was heated at 60°C for 30 minutes under an atmosphere of oxygen-free, dry nitrogen. The resulting oligomer can be thought of as related to oligo-butadiene, a relatively short-chained polybutadiene, to which relatively long pendant groups –  $(\text{CH}_2)_{14}\text{CO}_2^-\text{K}^+$  – are linked. Oligomerization of the surfactant resulted in a semi-rigid gel that retained the viscoelasticity of the original monomeric surfactant solution but showed none of its sensitivity to contact with hydrocarbon or water. The observations led to the conclusion that the oligomer was viscoelastic in character, much as was the original un-oligomerized viscoelastic surfactant solution; but the oligomerized solution was more stable than the original un-oligomerized solution.

**Example 4:** Oligomers of N-(undec-11-enyl)-N,N-dimethylammonium chloride were synthesized by linking the head groups with a branched  $\text{C}_9$  bridge. First, 5-methylnonane-1,8-diamine was reacted with 11-chloroundec-1-ene,  $\text{CH}_2=\text{CH}(\text{CH}_2)_9\text{Cl}$ , and subsequently quaternized to produce the following monomer:



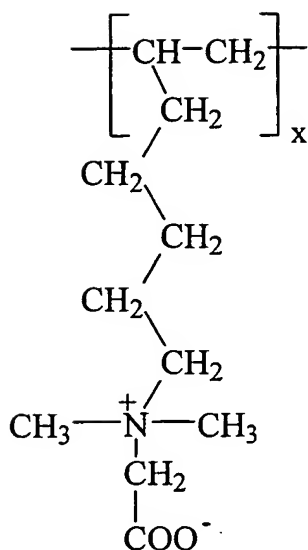
When this monomer was oligomerized, the product was a surfactant oligomer in which the  $\text{N}^+\text{Cl}^-$  head groups, again, are almost completely free relative to their neighboring  $\text{N}^+\text{Cl}^-$  head groups – each being linked to each other through relatively long  $(\text{CH}_2)_4\text{CHCH}_3(\text{CH}_2)_4$  chains or to the oligo-ethylene backbone through relatively long pendant  $(\text{CH}_2)_9$  chains.

The preparation of the monomer described above began with 5-methylnonane-1,8-diamine. This precursor was reacted with 11-chloroundec-1-ene,  $\text{CH}_2=\text{CH}(\text{CH}_2)_9\text{Cl}$ , under conditions such that each diamine group was be converted to secondary amines. In this first step in the synthesis, incompletely reacted material was separated from the di-secondary amine by distillation. The small part of the 5-methylnonane-1,8-diamine that had reacted too much and had become tertiarized or quaternized at one or both ends by excessive reaction with the 11-chloroundec-1-ene were separated by fractional crystallization from the desired N,N'-di-(1-decenyl)-5-methylnonane-1,8-diamine. Finally, this product was then reacted with an excess of methylbromide and quaternized to produce the monomer stated above. The oligomerization of this monomer:



in a viscoelastic solution was carried out at a concentration of 60 g/l in the presence of 40 g/l ammonium chloride. A volume of 100 ml of the viscoelastic surfactant solution was placed in a bottle purged with oxygen-free, dry nitrogen gas to remove any dissolved oxygen. After sufficient purging 10 mg of the free radical initiator 2,2'-azo(bis-amidinopropane) dihydrochloride was added to the viscoelastic surfactant solution and mixed thoroughly. The surfactant solution was heated at 60°C for 30 minutes under an atmosphere of oxygen-free, dry nitrogen. Oligomerization of the surfactant monomers in micelles resulted in the viscosity of the gel becoming largely insensitive to contact with hydrocarbon. The viscosity of the surfactant gel was not materially altered by the oligomerization of the surfactant monomers. The oligomerized surfactant gel retained its gel strength after prolonged contact with water. The observations led to the conclusion that the oligomer was viscoelastic in character, much as was the original un-oligomerized viscoelastic surfactant solution; but the oligomerized solution was more stable than the original un-oligomerized solution.

**Example 5:** The oligomerization of the viscoelastic surfactant monomer, the sodium salt of N-N-dimethyl-N-methylcarboxylate-N-1-hepten-7-ammonium chloride, was carried out to give oligo(1-hepten-7-quaternary-ammonio-N-N-dimethyl-N-methylcarboxylate), sodium salt. The resulting oligomer is believed to have the simplified structure as indicated below in the acid form rather than the sodium-salt form:



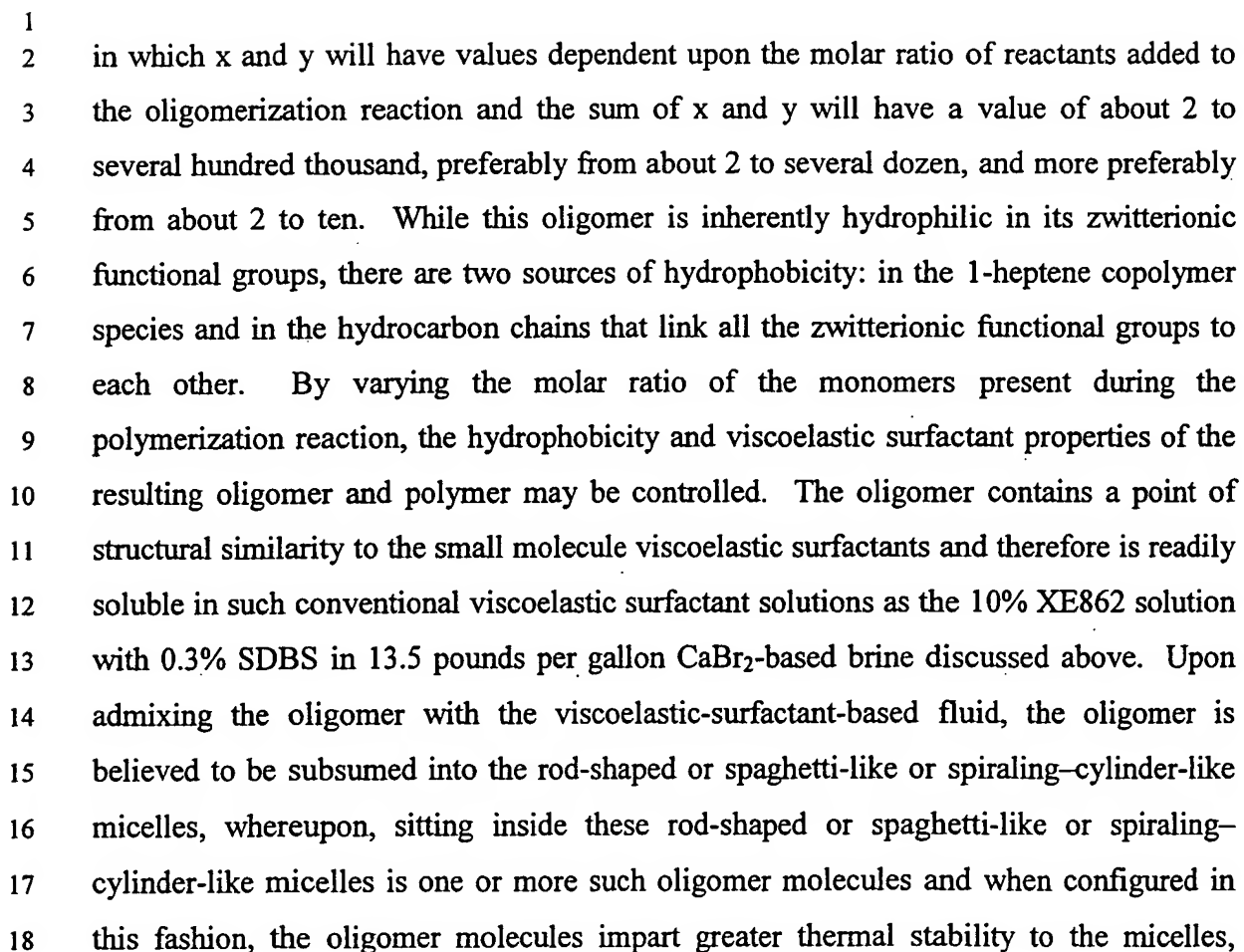
1  
2 in which x will have a value from about 2 to several hundred thousand, preferably from  
3 about 2 to several dozen, and more preferably from about 2 to perhaps 4. The monomer  
4 was prepared by the reaction of N-1-hept-6-enyl-N,N-dimethylamine with chloroacetic  
5 acid to produce N-1-hept-6-enyl-N-methylcarboxylic acid-N,N-dimethylammonium  
6 chloride. Upon neutralization with sodium hydroxide, the final product was the  
7 zwitterionic betaine which is the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-  
8 dimethylammonium chloride – it has a negative charge on the carboxyl group and the  
9 sodium cation associated with it as a counter ion, and a positive charge on the quaternary  
10 amine group and the chloride anion associated with it as a counter ion.

11 **Example 6:** The viscoelastic monomer, the sodium salt of N-carboxymethyl-N,N-  
12 dimethyl-N-1-hepten-7-ammonium chloride was mixed into the solution of the  
13 conventional rod-shaped or spaghetti-like or spiraling-cylinder-like micelles of the  
14 viscoelastic surfactant that is the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-  
15 dimethylammonium chloride. This illustrative monomer, the sodium salt of N-  
16 carboxymethyl-N,N-dimethyl-N-1-hepten-6-ammonium chloride, differs only in minor  
17 ways from the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-dimethylammonium  
18 chloride. Accordingly, the monomer was readily subsumed into the conventional rod-  
19 shaped or spaghetti-like micelles, whereupon oligomerization was initiated to produce the  
20 sodium salt of oligo- (1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-



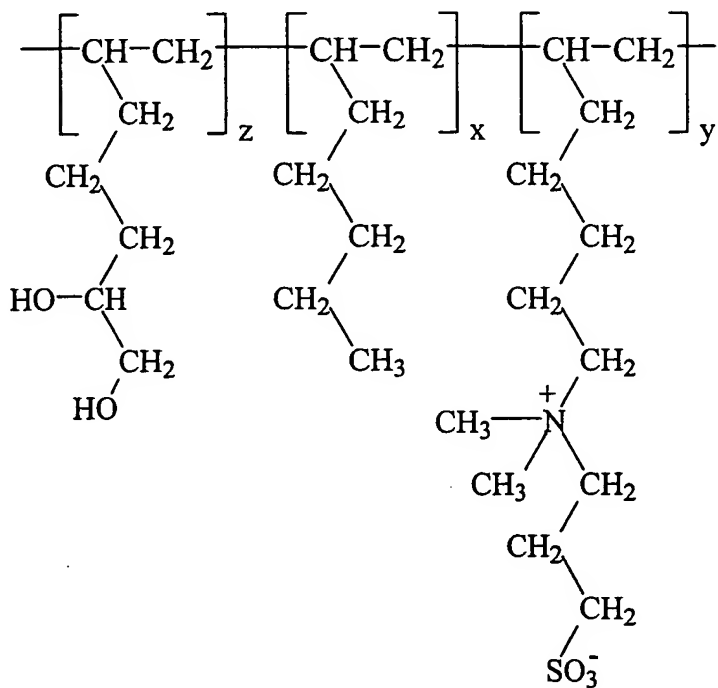
1 methylcarboxylate). This oligomer is inherently hydrophilic in its zwitterionic functional  
2 groups and hydrophobic in the hydrocarbon chains that link all the zwitterionic functional  
3 groups to each other. The oligomer is believed to be structurally quite similar to the  
4 viscoelastic surfactant molecules in the well fluid and therefore is readily soluble in the  
5 well fluid solution. the oligomers are likewise soluble or dispersible in other viscoelastic  
6 surfactant solutions such as 10% XE862 (a product that is commercially available from  
7 Schlumberger) solution with 0.3% SDBS in 13.5 pound per barrel  $\text{CaBr}_2$ -based brine.  
8 Upon mixing the oligomer or polymer with the viscoelastic-surfactant-based fluid, the  
9 oligomer or polymer is believed to be subsumed into the rod-shaped or spaghetti-like or  
10 spiraling-cylinder-like micelles, whereupon, sitting inside these rod-shaped or spaghetti-  
11 like or spiraling-cylinder-like micelles is one or more such oligomer or polymer  
12 molecules and when configured in this fashion, the oligomer or polymer molecules  
13 impart greater thermal stability to the micelles, greater resistance to shear stress and other  
14 stress conditions acting upon a fluid loss pill – including, for example, exposure to oil,  
15 high shear in pumping and placement, high temperature, high differential pressure, and  
16 low pH.

17 **Example 7:** In another illustrative embodiment of the claimed subject matter, a sufficient  
18 quantity, about 10 volume%, of the co-monomer 1-heptene was mixed into a viscoelastic  
19 solution of the sodium salt of 1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-  
20 propylsulfonate chloride, causing the solution to lose most or all of its viscoelastic  
21 character. Then the 1-heptene was co-oligomerized with the sodium salt of 1-hepten-7-  
22 quaternary-ammonio-N,N-dimethyl-N-propylsulfonate chloride to produce oligo-(1-  
23 heptene-co-1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-propylsulfonate). The  
24 resulting oligomer is believed to have the simplified structure as indicated below:



greater resistance to shear stress and other stress conditions acting upon a fluid loss pill – including, for example, exposure to oil, high shear in pumping and placement, high temperature, high differential pressure, and low pH.

**Example 8:** In another illustrative embodiment of the claimed subject matter, a sufficient quantity, about 8 volume%, of the co-monomer 1-heptene was mixed into a viscoelastic solution of 1-heptene-6,7-diol and the sodium salt of 1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-propylsulfonate chloride, causing the solution to lose most or all of its viscoelastic character. The resulting oligomer, oligo-(1-heptene-co-1-heptene-6,7-diol-co-1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-propylsulfonate), is believed to have the simplified structure as indicated below:



in which x, y and z will have values dependent upon the molar ratio of reactants added to the oligomerization reaction and the sum of x, y, and z will have a value of about 3 to several hundred thousand, preferably from about 3 to several dozen, and more preferably from about 3 to ten. As with the other polymers of the claimed subject matter, the above illustrative polymer is soluble in conventional viscoelastic surfactant well fluids such as the 10% XE862 solution with 0.3% SDBS in 13.5 pounds per gallon  $\text{CaBr}_2$ -based brine discussed above. Upon admixing the illustrative polymer with such a viscoelastic-

1 surfactant-based fluid, the polymer is believed to be subsumed into the rod-shaped or  
2 spaghetti-like or spiraling-cylinder-like micelles, whereupon, sitting inside these rod-  
3 shaped or spaghetti-like or spiraling-cylinder-like micelles is one or more such polymer  
4 molecules and when configured in this fashion, the polymer molecules impart greater  
5 thermal stability to the micelles, greater resistance to shear stress and other stress  
6 conditions acting upon a fluid loss pill – including, for example, exposure to high shear in  
7 pumping and placement, high temperature, high differential pressure, and low pH.  
8 Additionally, the co-monomer 1-heptene-6,7-diol incorporates a vicinal diol functionality  
9 into the polymer. Surface forces should cause the vicinal diol functionality to present  
10 itself at the outer surface of any rod-shaped micelle in which it has become subsumed.  
11 The vicinal diol functionality was subsequently crosslinked with borate, leading  
12 effectively to the crosslinking of the viscoelastic surfactant assemblies in which the  
13 oligomers had been subsumed.

14 As discussed above, the novel oligomers or polymers taught in accordance with  
15 the claimed subject matter contain chemical functional groups that are similar to those  
16 found in conventional viscoelastic surfactants and thus are highly compatible with  
17 conventional viscoelastic surfactant systems. Accordingly, the claimed subject matter  
18 teaches blends of the viscoelastic oligomers or polymers with conventional viscoelastic  
19 surfactant systems. The viscoelastic-surfactant-based fluids useful for the claimed  
20 subject matter are those in the following citations, all of which are incorporated herein by  
21 reference – Canadian Patent 1,298,697, U. S. Patents 4,615,825, 4,695,389, 4,725,372,  
22 5,258,137, 5,551,516, 5,691,292, 5,964,295, 5,965,502, 5,979,555, 5,979,557, 6,140,277,  
23 6,194,355, 6,194,356, 6,211,120, 6,232,274, 6,239,183, Paper SPE 17,168, Paper SPE  
24 30,098, Paper SPE 30,114, Paper SPE 30,458, Paper SPE 31,114, Paper SPE 38,622,  
25 Paper SPE 56,467, Paper SPE 57,432, Paper SPE 59,478, and Paper SPE 60,322.

26 Furthermore, the oligomeric or polymeric viscoelastic surfactants may be added  
27 to a well fluid in substantially any convenient manner. Thus, the oligomeric or polymeric  
28 viscoelastic surfactants may be added directly to the well fluid either in solid form or in  
29 the form of an aqueous solution. Alternatively, the oligomeric or polymeric viscoelastic  
30 surfactants may be separately added to a solution already containing other surfactants or  
31 viscoelastic surfactants to provide a fluid loss controlling base solution, with the optional

1 crosslinking agents thereafter being added to the fluid loss controlling base solution  
2 immediately prior to use.

3 In preferred embodiments, the oligomeric or polymeric viscoelastic surfactants,  
4 including optional crosslinking agents, are supplied to the well fluid at a level of about 20  
5 ppm to about 20 parts per 100 parts of the well fluid, more preferably about 100 ppm to  
6 about 15 parts per 100 parts of the well fluid.

7 In more preferred embodiments, the oligomeric or polymeric viscoelastic  
8 surfactants, including optional crosslinking agents, are supplied to the well fluid already  
9 containing an amount of other surfactants or viscoelastic surfactants. When these more  
10 preferred embodiments are exercised, the oligomeric or polymeric viscoelastic surfactants  
11 are supplied at a level of about 20 ppmw to about 10 parts per 100 parts of the well fluid,  
12 preferably about 100 ppmw to about 5 parts per 100 parts of the well fluid.

13 Techniques for measuring fluid loss control are well known in the art and should  
14 be well known to one of skill and knowledge of the formulation of drilling fluids.  
15 Specifically, the fluid loss measurements of the present disclosure were made with  
16 reference to API Recommended Practice RP 13B-1, Second Edition, September 1997, pp.  
17 9-11, the contents of which are incorporated by reference.

18 While the apparatus, compositions and methods of the claimed subject matter  
19 have been described in terms of preferred or illustrative embodiments, it will be apparent  
20 to those of skill in the art that variations may be applied to the process described herein  
21 without departing from the concept and scope of the claimed subject matter. All such  
22 similar substitutes and modifications apparent to those skilled in the art are deemed to be  
23 within the scope and concept of the claimed subject matter.

24

1 **What is claimed is:**

2 1. A wellbore fluid comprising:

3 an aqueous based continuous phase;

4 a viscoelastic surfactant; and

5 a surfactant-polymer compound soluble in an aqueous solution, the surfactant-  
6 polymer compound having a hydrophobic backbone and a plurality of hydrophilic  
7 functional groups attached to the hydrophobic backbone,

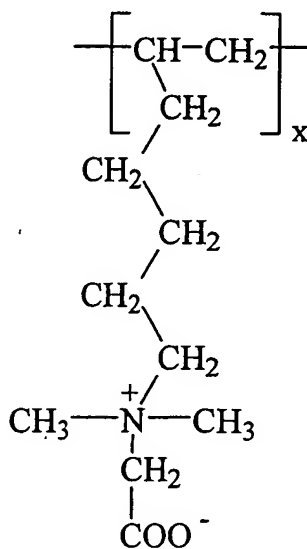
8 wherein the hydrophobic backbone is the reaction product of one or more  
9 molecules having polymerizable alkene or alkyne functional groups;

10 wherein the hydrophilic functional groups are selected from: zwitterionic  
11 surfactant functional groups, anionic surfactant functional groups, cationic surfactant  
12 functional groups, and nonionic surfactant functional groups; and,

13 wherein the combination of the viscoelastic surfactant and surfactant-polymer  
14 compound form micellar assemblies.

15  
16 2. The wellbore fluid of claim 1 further comprising a water-soluble inorganic salt.

17  
18 3. The wellbore fluid of claim 1 wherein the acid form of the surfactant-polymer  
19 compound has the structure:



wherein  $x = 2$  to 300,000.

4. The wellbore fluid of claim 3 wherein  $x = 2$  to 36.

5. The wellbore fluid of claim 1 wherein the surfactant-polymer compound is a salt of oligo- or poly-( $\alpha$ -alkenyl - $\omega$ - or  $\alpha$ -alkynyl - $\omega$ -quaternary-ammonio-N,N-dialkyl-N-alkylcarboxylate) or a mixture further comprising a salt of N-alkyl-N-carboxymethyl-N,N-dimethylammonium chloride.

6. The wellbore fluid of claim 1 wherein the surfactant-polymer compound is a salt of oligo- or poly-(1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-methylcarboxylate) or a mixture further comprising a salt of N-hexadecyl-N-carboxymethyl-N,N-dimethylammonium chloride.

7. The wellbore fluid of claim 1 wherein the surfactant-polymer compound is a salt of oligo- or poly-( $\alpha$ -alkenyl - $\omega$ - or  $\alpha$ -alkynyl - $\omega$ -quaternary-ammonio-N,N-dialkyl-N-alkylcarboxylate).

8. The wellbore fluid of claim 1 wherein the surfactant-polymer compound is a salt of oligo- or poly-(1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-methylcarboxylate).

9. A wellbore fluid comprising:

an aqueous fluid;

a viscoelastic surfactant;

a thickener soluble in the aqueous fluid,

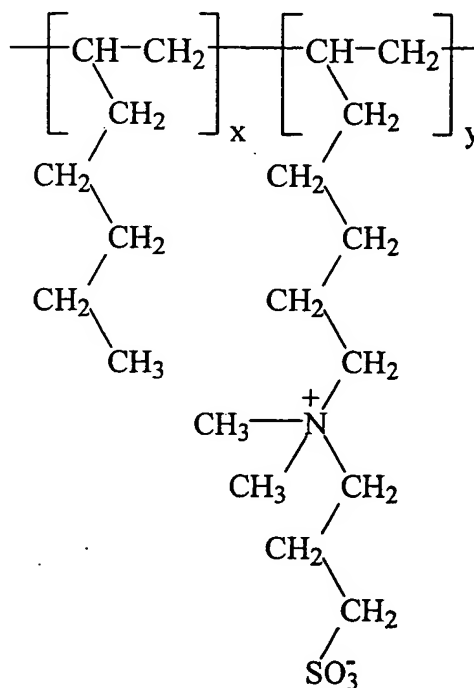
wherein the thickener has a hydrophobic oligomeric or polymeric backbone made from the reaction of alkene monomer or alkyne monomer, or mixtures thereof, and

wherein surfactant functional groups are attached to the hydrophobic backbone, wherein the surfactant functional group is selected from: zwitterionic surfactant functional groups, anionic surfactant functional groups, cationic surfactant functional groups, and nonionic surfactant functional groups; and,

wherein the thickener has a molecular confirmation such that the surfactant functional groups are hydrophilic and the hydrophobic oligomeric or polymeric backbone is hydrophobic; and

wherein the combination of viscoelastic surfactant and thickener form micellar assemblies such that the wellbore fluid thickener develops viscoelastic characteristics.

10. The wellbore fluid of claim 9 wherein the thickener has the following structure:



wherein  $x + y = 2$  to 300,000.

11. The wellbore fluid of claim 10 wherein  $x + y = 2$  to 36.

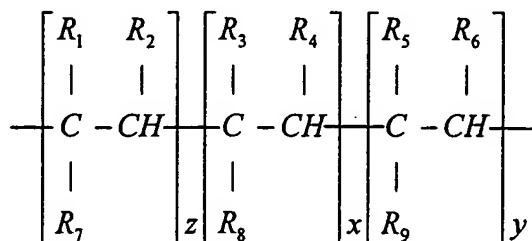
12. The wellbore fluid of claim 11 wherein the thickener is a salt of oligo- or poly-( $\alpha$ -alkene - $\omega$ - or  $\alpha$ -alkyne-co- $\alpha$ -alkenyl - $\omega$ - or  $\alpha$ -alkynyl - $\omega$ -quaternary-ammonio-N,N-dialkyl-N-alkylsulfonate).



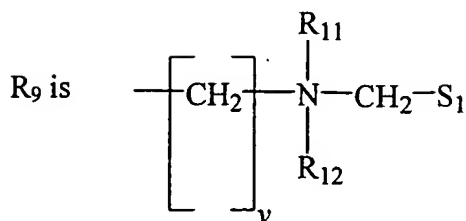
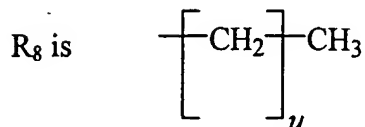
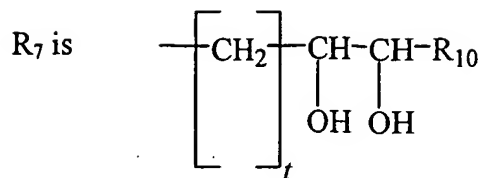
1 13. The wellbore fluid of claim 11 wherein the thickener is oligo- or poly-(1-heptene-co-  
2 1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-propylsulfonate).

3  
4 14. The wellbore fluid of claim 9 wherein the viscoelastic surfactant is selected from the  
5 following: anionic, nonionic, amphoteric, zwitterionic, alcohols, alkano-lamides,  
6 alkanolamines, alkylaryl sulfonates, alkylaryl sulfonic acids, amine acetates, amine  
7 oxides, amines, sulfonated amines, sulfonated amides, betaines, block polymers,  
8 carboxylated alcohols, alkylphenol ethoxylates, carboxylic acids, fatty acids, ethoxylated  
9 alcohols, ethoxylated alkylphenols, ethoxylated amines, ethoxylated amides, ethoxylated  
10 fatty acids, ethoxylated fatty esters, fatty esters, fluorocarbon-based surfactants, glycerol  
11 esters, glycol esters, heterocyclic surfactants, imidazolines, isethionates, lanolins,  
12 lechithins, methyl esters, monoglycerides, olefin sulfonates, phosphate esters,  
13 polyethylene glycols, polysaccharides, polyacrylic acids, polyacrylamides, propoxylated  
14 alcohols, propoxylated alkylphenols, propoxylated amines, propoxylated amides,  
15 propoxylated fatty acids, propoxylated fatty esters, protein-based surfactants, quaternary  
16 surfactants, sarcosamines, silicone-based surfactants, soaps, sodium isethionate,  
17 sorbitans, sucrose and glucose esters, sulfates and sulfonates of oils and fatty acids,  
18 sulfates and sulfonates of ethoxylated alkylphenols, sulfates of alcohols, sulfates of  
19 ethoxylated alcohols, sulfates of fatty esters, sulfonates of hydrocarbons and petroleum,  
20 sulfosuccinates, taurates, and tridecyl and dodecyl benzene sulfonic acids and mixtures  
21 thereof.

22  
23 15. The wellbore fluid of claim 9 wherein the thickener has the following structure:



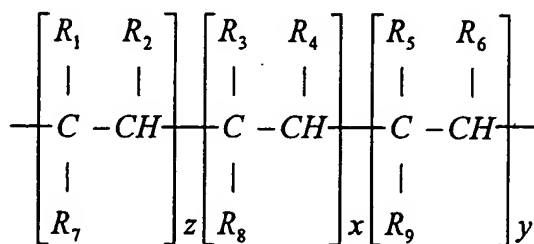
where  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H or } \text{CH}_3$



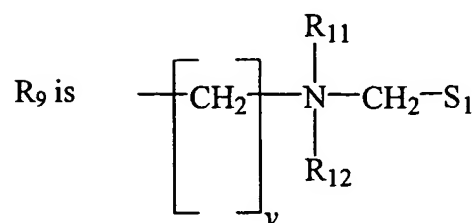
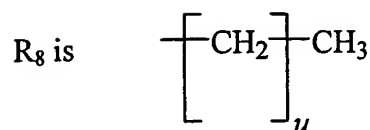
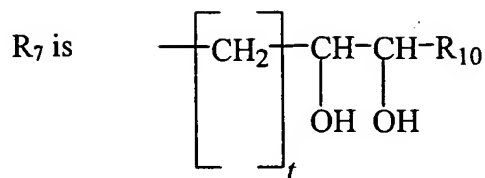
15 wherein  $R_{10}$ ,  $R_{11}$ ,  $R_{12}$  = H or  $\text{CH}_3$ , and  $t = 1$  to 16,  $u = 6$  to 12,  $v = 1$  to 18,  $w = 1$  to 3,  
16 and  $x + y + z = 3$  to 300,000 and  $\text{S}_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ .

17  
18 16. The wellbore fluid of claim 15 wherein  $t = 12$  to 16,  $u = 6$  to 12;  $v = 12$  to 18,  $w = 1$   
19 to 3,  $x = 0$  to 10,000,  $y = 2$  to 300,000 and  $z = 0$  to 10,000

20  
21 17. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following  
22 structure:



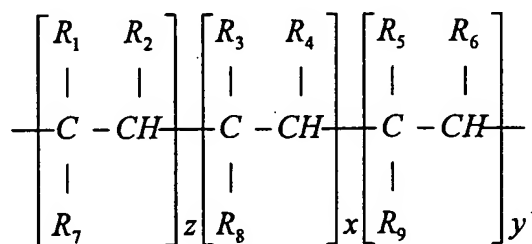
wherein  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  = H or  $\text{CH}_3$



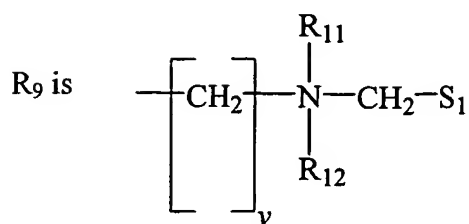
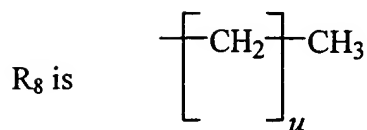
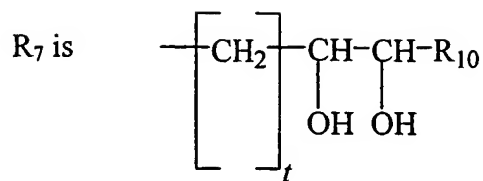
15 wherein  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ , and  $x + y + z = 3$  to  
16  $300,000$ .

17  
18 18. The wellbore fluid of claim 17 wherein  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $w = 1$   
19 to  $3$ , and  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

20  
21 19. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following  
22 structure



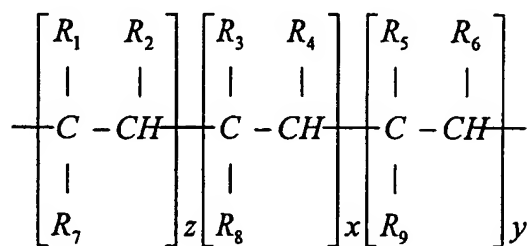
24  
25 where  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



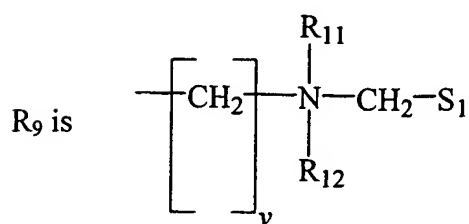
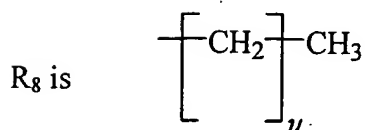
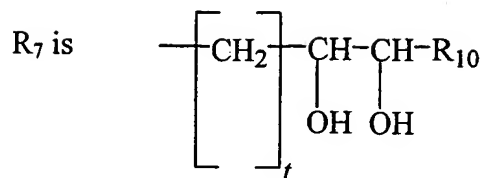
17 wherein  $R_{10}$ ,  $R_{11}$ ,  $R_{12}$  = H or  $\text{CH}_3$ ,  $t$  = 1 to 16,  $u$  = 6 to 12,  $v$  = 1 to 18,  $x + y + z$  = 3 to  
18 300,000, and  $\text{S}_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ .

19  
20 20. The wellbore fluid of claim 19 wherein  $t$  = 12 to 16,  $u$  = 6 to 12,  $v$  = 12 to 18,  $x$  = 0 to  
21 10, 000,  $y$  = 2 to 300,000, and  $z$  = 0 to 10,000.

22  
23 21. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following  
24 structure:



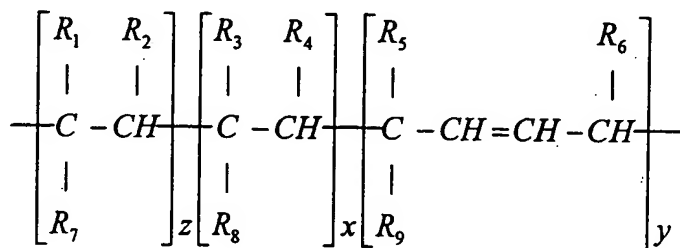
26  
27  
28 wherein  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  = H or  $\text{CH}_3$



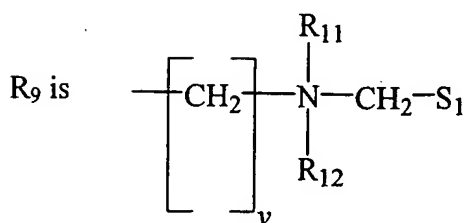
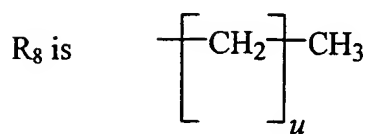
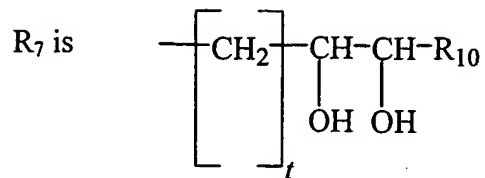
16 wherein,  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  $12$ , and  $x$   
17  $+ y + z = 3$  to  $300,000$ .

18  
19 22. The wellbore fluid of claim 21 wherein  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $w = 1$   
20 to  $3$ , and  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

21  
22 23. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following  
23 structure:



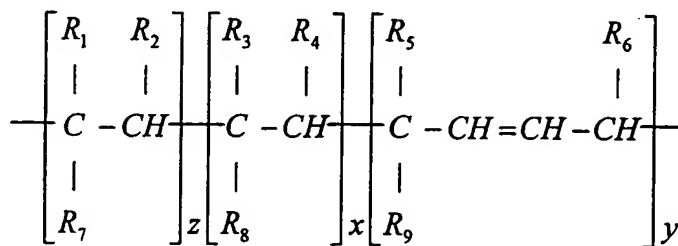
wherein  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



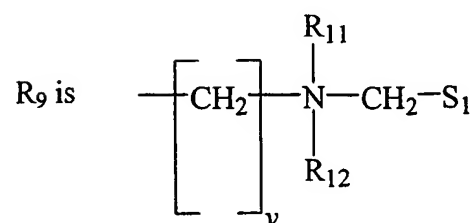
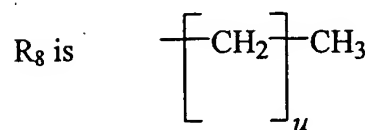
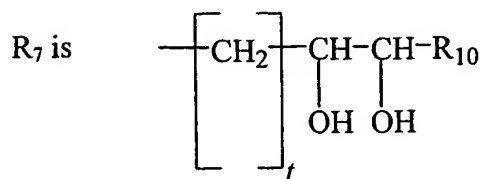
16 wherein  $R_{10}$ ,  $R_{11}$ ,  $R_{12} = \text{H}$  or  $\text{CH}_3$ , and  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  $3$ ,  
17 and  $x + y + z = 3$  to  $300,000$  and  $\text{S}_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ .  
18

19 24. The wellbore fluid of claim 23 wherein  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ;  $v = 12$  to  $18$ ,  $w = 1$   
20 to  $3$ ,  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$  and  $z = 0$  to  $10,000$   
21

22 25. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following  
23 structure:



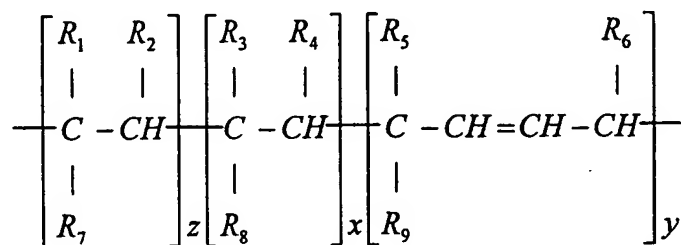
wherein  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



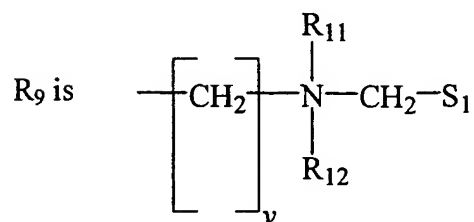
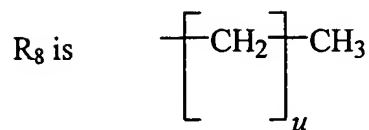
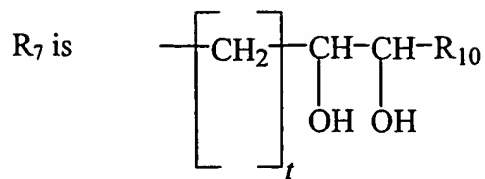
15  
16 wherein  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ , and  $x + y + z = 3$   
17 to  $300,000$ .  
18

19 28. The wellbore fluid of claim 25 wherein  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $w$   
20  $= 1$  to  $3$ , and  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .  
21

22 27. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following  
23 structure:



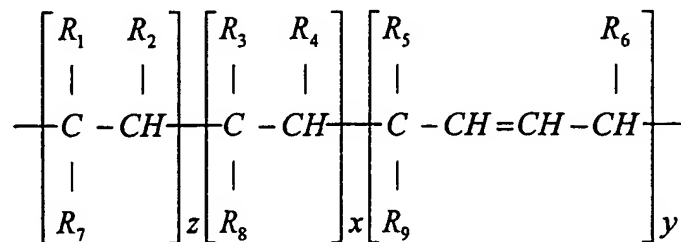
wherein  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



wherein  $R_{10}$ ,  $R_{11}$ ,  $R_{12}$  = H or  $\text{CH}_3$ ,  $t$  = 1 to 16,  $u$  = 6 to 12,  $v$  = 1 to 18,  $x + y + z$  = 3 to 300,000, and  $S_1$  =  $\text{CO}_2^-$  or  $\text{SO}_3^-$ .

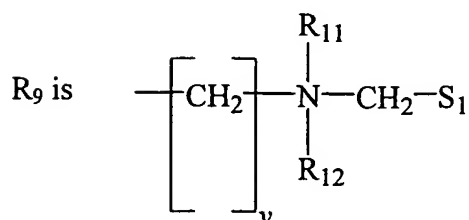
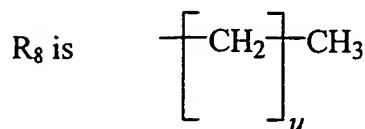
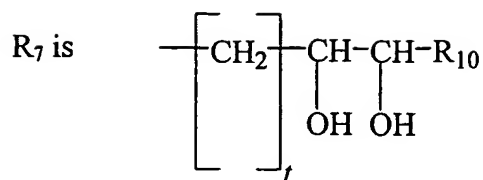
28. The wellbore fluid of claim 27 wherein  $t$  = 12 to 16,  $u$  = 6 to 12,  $v$  = 12 to 18,  $w$  = 1 to 3, and  $x$  = 0 to 10,000,  $y$  = 2 to 300,000, and  $z$  = 0 to 10,000.

29. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following structure:



wherein  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  = H or  $\text{CH}_3$





wherein,  $R_{10}$ ,  $R_{11}$ ,  $R_{12}$  = H or  $\text{CH}_3$ ,  $t$  = 1 to 16,  $u$  = 6 to 12,  $v$  = 1 to 18,  $w$  = 1 to 12, and  $x + y + z$  = 3 to 300,000.

30. The wellbore fluid of claim 29 wherein  $t$  = 12 to 16,  $u$  = 6 to 12,  $v$  = 12 to 18,  $w$  = 1 to 3, and  $x$  = 0 to 10,000,  $y$  = 2 to 300,000, and  $z$  = 0 to 10,000.

31. The wellbore fluid of claim 9 wherein the oligomers or polymers are crosslinked with polyvalent metal ions, formaldehyde, or glutaraldehyde.

32. The wellbore fluid of claim 31 wherein the polyvalent metal ions are selected from the following:  $\text{Fe}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{UO}_2^{2+}$ ,  $\text{PbO}^{2+}$ ,  $\text{Al}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Cr}^{3+}$ ,  $\text{Ce}^{3+}$ ,  $\text{Ti}^{4+}$ ,  $\text{Zr}^{4+}$ ,  $\text{Sn}^{4+}$  and mixtures thereof.

33. A method of making a wellbore fluid comprising blending:

an aqueous fluid phase;

a viscoelastic surfactant;

a water-soluble inorganic salt;

1 an oligomer or polymer soluble in an aqueous salt solution, the oligomer or  
2 polymer comprising a hydrophobic oligomeric or polymeric backbone made from the  
3 oligomerization or polymerization of alkene or alkyne monomer groups, or mixtures  
4 thereof, the oligomer or polymer further comprising zwitterionic functional groups  
5 attached to the hydrophobic backbone, wherein the oligomer or polymer is hydrophilic in  
6 the zwitterionic functional groups and hydrophobic in the backbone hydrocarbon chain to  
7 form micellar assemblies such that the oligomers or polymers develop viscoelastic  
8 character prior to a polymerization step.

9  
10 34. The method of claim 33 further comprising a polymerization step of the polymer or  
11 oligomer, then drying the product and subsequently admixing it into a solution of  
12 conventional surfactants.

13  
14 35. A method of drilling a subterranean well, the method comprising:  
15 drilling the subterranean well using a rotary drilling rig and circulating a drilling  
16 fluid in the subterranean well, wherein the drilling fluid is the wellbore fluid of claim 1.

17  
18 36. A method of reducing the loss of fluid out of a subterranean well, the method  
19 comprising injecting into the subterranean well a wellbore fluid as recited in claim 1.

20

## ABSTRACT

1  
2 Compositions for increasing the thermal and pressure stability of well fluids  
3 viscosified using viscoelastic surfactants, the compositions including an effective amount  
4 of an oligomeric or polymeric compound that has a thermally stable backbone structure  
5 and at least one pendent viscoelastic surfactant functional group. Preferred compositions  
6 for increasing the stability of well fluids viscosified using monomeric viscoelastic  
7 surfactants include an effective amount of an oligomeric or polymeric compound that has  
8 a thermally stable backbone structure and a multiplicity of pendent viscoelastic surfactant  
9 functional groups attached to said backbone structure through relatively long  
10 hydrocarbon chains, 1 to 18 carbons in length.



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Via Certified Mail No. 70033110000420008757

April 21, 2006

Ms. Kamila B. Abdur-Rahman  
PO Box 1603  
Alief, TX 77411

Re: *U.S. Patent Application No. 10/620,041*  
*Entitled: Surfactant-Polymer Compositions for Enhancing the Stability of Viscoelastic-Surfactant Based Fluid*  
*Inventors: Robert Horton, Bethicia B. Prasek, Frederick Growcock, David P. Kippie,*  
*John Vian, Kamila Abdur-Rahman, Morris Arvie*  
*M-I L.L.C. Ref: PA-00146US*

Dear Ms. Abdur-Rahman:

We write you regarding the above-referenced U.S. application in which you are listed as an inventor. In July 2003, when the application was filed, you as well as the other inventors signed the Declaration and Assignment. However, we have received a notice from the Patent Office that your signature was not dated, therefore they are requiring we submit a new declaration with a new signature and date.

We have enclosed a new Declaration for your execution as well as a copy of the application, original Declaration and Assignment you previously executed. Please sign and date the new declaration (yellow tab) and forward to us in the enclosed return FedEx envelope as soon as possible.

Please do not hesitate to contact me or Carter White at 281-561-1450 or [cjwhite@miswaco.com](mailto:cjwhite@miswaco.com) should you have any questions. We appreciate your help with this matter.

Regards,

Leah S. Bardin  
Patent Paralegal

Encl.

cc: Carter J. White, Ph.D.

I hereby certify that this correspondence is being deposited with the U.S. Postal Service as Express Mail, Airbill No. \_\_\_\_\_, in an envelope addressed to: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450, on the date shown below.

Dated: \_\_\_\_\_ Signature: \_\_\_\_\_ 0

**IN THE UNITED STATES PATENT AND TRADEMARK OFFICE**

**DECLARATION FOR PATENT APPLICATION**

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name.

I believe I am an original, first and joint inventor of the subject matter which is claimed and for which a patent is sought on the invention entitled:

**SURFACTANT-POLYMER COMPOSITIONS FOR ENHANCING THE STABILITY OF VISCOELASTIC-SURFACTANT BASED FLUID**

the specification of which was filed on July 15, 2003 as Application No. 10/620,041.

In the event that the filing date and/or Application No. are not entered above at the time I execute this document, and if such information is deemed necessary, I hereby authorize and request my attorneys/agent(s) at **Osha · Liang LLP**, 1221 McKinney St., Suite 2800, Houston, Texas 77010, to insert above the filing date and/or Application No. of said application.

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by an amendment, if any, specifically referred to herein.

I acknowledge the duty to disclose all information known to me that is material to patentability in accordance with Title 37, Code of Federal Regulations, § 1.56.

**FOREIGN PRIORITY CLAIM**

I hereby claim foreign priority benefits under Title 35, United States Code § 119(a)-(d) of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

☒ no such foreign applications have been filed

☐ such foreign application have been filed as follows:

**EARLIEST FOREIGN APPLICATION(S), IF ANY FILED WITHIN 12 MONTHS  
(6 MONTHS FOR DESIGN) PRIOR TO THIS U.S. APPLICATION**

Application Number	Country	Date of Filing	Priority Claimed Under 35 USC 119

**ALL FOREIGN APPLICATION(S), IF ANY FILED MORE THAN 12 MONTHS  
(6 MONTHS FOR DESIGN) PRIOR TO THIS U.S. APPLICATION**

Application Number	Country	Date of Filing

**CLAIM FOR BENEFIT OF EARLIER U.S. PROVISIONAL APPLICATIONS**

I hereby claim priority benefits under Title 35, United States Code §119(e), of any United States provisional patent application(s) listed below:

☐ no such U.S. provisional applications have been filed.

☒ such U.S. provisional application have been filed as follows:

Application Number	Date of Filing	Priority Claimed Under 35 USC 119
60/319,575	September 25, 2002	<input checked="" type="checkbox"/> Yes <input type="checkbox"/> No

**CLAIM FOR BENEFIT OF EARLIER U.S./PCT APPLICATION(S)**

I hereby claim the benefit under Title 35, United States Code, §120 of the United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, §112, I acknowledge the duty to disclose all information that is material to patentability in accordance with Title 37, Code of Federal Regulations, §1.56 which became available to me between the filing date of the prior application and the national or PCT international filing date of this application:

☒ no such U.S./PCT applications have been filed.

☐ such U.S./PCT application have been filed as follows:

Application Number	Relationship	Parent Application	Date of Filing


I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

Please mail all correspondence to Jeffrey S Bergman, whose address is:

**Osha · Liang LLP**  
1221 McKinney St., Suite 2800  
Houston, Texas 77010

Please direct telephone calls to: Jeffrey S. Bergman at (713) 228-8600.

Please direct facsimiles to: (713) 228-8778

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Sole or first inventor's signature	Date
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Citizenship <b>US</b>	
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Second inventor's signature	Date
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Citizenship <b>US</b>	
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Third inventor's signature	Date
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Mailing Address  <b>1514 Dodd Lane Houston, Texas 77077</b>	

Full name of fourth inventor, if any <b>David P. Kippie</b>	
Fourth inventor's signature	Date
Residence <b>Katy, Texas</b>	
Citizenship <b>US</b>	
Mailing Address  <b>3610 Perdarnales Trails Lane Katy, Texas 77450</b>	



Full name of fifth inventor, if any <b>John W. Vian</b>	
Fifth inventor's signature	Date
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Citizenship <b>US</b>	
Mailing Address  <b>3317 Greenwood Pl. Deer Park, Texas 77536</b>	

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Sixth inventor's signature	Date
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Citizenship <b>US</b>	
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Full name of seventh inventor, if any <b>Morris Arvie</b>	
Seventh inventor's signature	Date
Residence <b>Houston, Texas</b>	
Citizenship <b>US</b>	
Mailing Address  <b>11655 Briar Forest #83 Houston, Texas 77077</b>	

Full name of eighth inventor, if any	
Eighth inventor's signature	Date
Residence	
Citizenship	
Mailing Address	

**APPLICATION FOR UNITED STATES LETTERS PATENT**

**for**

**SURFACTANT-POLYMER COMPOSITIONS FOR ENHANCING THE STABILITY  
OF VISCOELASTIC-SURFACTANT BASED FLUID**

**by**

**Robert L. Horton**

**Bethicia B. Prasek**

**Frederick B. Growcock**

**David Kippie**

**John W. Vian**

**Kamila B. Abdur-Rahman**

**and**

**Morris Arvie, Jr.**

1 This application claims the benefit of U. S. Provisional Application No. 60/319,575, filed  
2 September 25, 2002, the entire contents of which are incorporated by reference herein.

### 3 4 BACKGROUND

5 When drilling or completing wells in earth formations, various fluids typically are  
6 used in the well for a variety of reasons. For the purposes of the present disclosure, such  
7 a fluid will be referred to as a "well fluid." Common uses for well fluids include:  
8 lubrication and cooling of drill bit cutting surfaces while drilling generally or drilling-in  
9 (i.e., drilling into a targeted formation), transportation of cuttings to the surface,  
10 controlling formation pressure to prevent blowouts, maintaining well stability,  
11 suspending solids in the well, minimizing fluid loss into and stabilizing the formation  
12 through which a well is being drilled, fracturing the formation in the vicinity of a well,  
13 displacing the fluid within a well with another fluid, cleaning a well, testing a well,  
14 emplacing spacer or fluid loss control pills at various points in the displacement,  
15 completion, or work-over process, emplacing a packer fluid in the completed wellbore  
16 during production, preparing the well for abandonment, abandoning the well or,  
17 otherwise treating the well or the formation. A commonly used type of well fluid is  
18 based on water-based solutions including brines. Brines, such as  $\text{CaBr}_2$  brine, are  
19 commonly used as well fluids because of the ability to control the density of the solution  
20 over a wide density range. Further the brines are typically substantially free of suspended  
21 solids and brines typically do not damage the more common types of subterranean  
22 formations.

23 When drilling progresses to the level of penetrating a hydrocarbon-bearing  
24 formation, special care may be required to maintain the stability of the wellbore.  
25 Examples of formations in which problems often arise are highly permeable and/or  
26 poorly consolidated formation and thus a technique known as "under-reaming" may be  
27 employed. In conducting the under-reaming process, the wellbore is drilled to penetrate  
28 the hydrocarbon-bearing zone using conventional techniques. A casing generally is set in  
29 the wellbore to a point just above the hydrocarbon-bearing zone. The hydrocarbon-  
30 bearing zone then may be re-drilled to a wider diameter, for example, using an  
31 expandable under-reamer that increases the diameter of the wellbore. Under-reaming

usually is performed using special "clean" drilling fluids. Typically the "clean" drilling fluids used in under-reaming are aqueous, dense brines that are viscosified with a gelling and/or cross-linked polymer to aid in the removal of formation cuttings. The expense of such fluids limits their general use in the drilling process.

When the target subterranean formation has a high permeability a significant quantity of the drilling fluid may be lost into the formation. Once the drilling fluid is lost into the formation, it becomes difficult to remove. Removal of the aqueous based well fluids is desired to maximize the production of the hydrocarbon in the formation. It is well known in the art that calcium- and zinc-bromide brines can form highly stable, acid insoluble compounds when reacted with the formation rock itself or with substances contained within the formation. These reactions often may substantially reduce the permeability of the formation to any subsequent out-flow of the desired hydrocarbons. As should be well known in the art, it is widely and generally accepted that the most effective way to prevent such damage to the formation is to limit fluid loss into the formation. Thus, providing effective fluid loss control is highly desirable to prevent damaging the hydrocarbon-bearing formation. For example such damage may occur during, completion, drilling, drill-in, displacement, hydraulic fracturing, work-over, packer fluid emplacement or maintenance, well treating, or testing operations.

Techniques that have been developed to control fluid loss include the use of fluid loss control "pills." As the term is used in this disclosure a "pill" is a quantity of fluid added to the well fluid so as to temporarily change the properties of the well bore fluid at or near a specific point in the well bore. Significant research has been directed to determining suitable materials for the fluid loss pills, as well as controlling and improving the properties of the fluid loss pills. Typically, fluid loss pills work by enhancing filter-cake buildup on the face of the formation to inhibit fluid flow into the formation from the wellbore; however the fluids in accordance with the claimed subject matter are effective by developing extremely high viscosity in the environment at and just within the face of the formation to inhibit fluid flow into the formation from the wellbore. Because of the high temperatures, high shear (caused by the pumping and placement of the pill), high pressures, and low pH to which well fluids may be exposed (i.e., "stress

conditions”), synthetic polymeric materials typically used to form fluid loss pills and to viscosify the well fluids tend to degrade rather quickly.

One class of viscosifiers commonly used in the petroleum industry comprises polymeric structures starting with molecular weights of hundreds of thousands to several million grams per mole. These large, chemically bonded structures are often crosslinked to further increase molecular weight and effective viscosity per gram of polymer added to the fluid. These large molecules are quite stable under the thermal conditions typically encountered in a subterranean reservoir. However, this thermal stability is believed to contribute to decreased well productivity. As a result, expensive and often corrosive breakers have been designed to destroy the molecular backbone of these polymeric structures. These breakers are typically oxidizers or enzymes and are at best only partially effective with typical reservoir cleanup less than 80% complete and more usually much less than 50% complete. It is also reported in the literature that the long term stability of polymeric based thickening agents is shortened by the high temperature, high shear, high pressures, and low pH to which well fluids may be exposed (i.e., “stress conditions”).

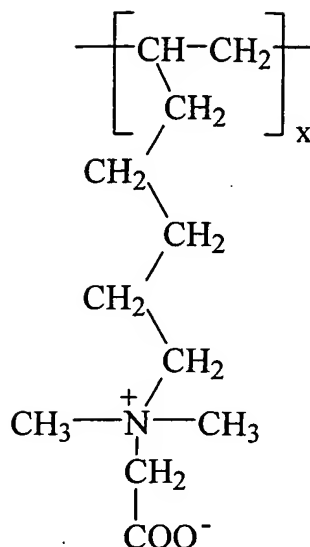
Viscoelastic surfactants are commonly used in the petroleum industry as an alternative to the above mentioned polymeric thickening agents. Viscoelastic surfactants are relatively small molecules with each molecule being typically less than 500 grams per mole (i.e., molecular weight less than 500). These small molecules will associate under certain conditions to form structures which resemble the polymer molecules but which are not stable structures. The individual molecules of surfactant begin to associate to form rod-like or spiraling-cylinder-like micelles. These micelle structures are always in an equilibrium state of breaking and reforming. As dynamic structures, these polymer-shaped micelles are readily destroyed by shear, presence of hydrocarbons or increased temperature. While these features are desirable especially in a hydrocarbon-bearing formation, there is minimal control over the conditions under which micelle breakup occurs. Therefore, under conditions of exposure to oil, high temperature, high shear, or other “stress conditions”, the viscoelastic surfactants rapidly return to their original small independent spherical micellar state. When the viscoelastic micelles are broken down to

1 this small independent spherical micellar state, the desired viscous nature of the well fluid  
2 is lost. In some cases the loss is temporary, in others the loss may be more permanent.

3  
4 Presently there exists an unmet need for a simple, inexpensive way to increase the  
5 thermal range for viscoelastic-surfactant-based viscosifying agents used in downhole  
6 applications. Preferably, this thermal extender would be applicable to various  
7 viscoelastic-surfactant-based viscosifying agents.

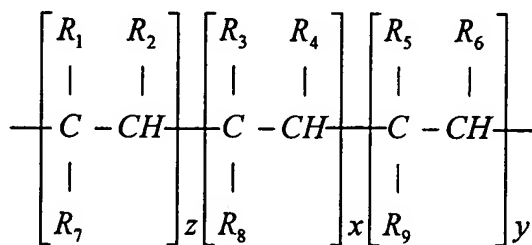
### 8 9 SUMMARY

10 Upon review of the following disclosed and claimed subject matter, one of skill in  
11 the art should appreciate and understand that one illustrative embodiment is a wellbore  
12 fluid that includes: an aqueous based continuous phase; a viscoelastic surfactant; and a  
13 surfactant-polymer compound soluble in an aqueous solution. The surfactant-polymer  
14 compound has a molecular structure including a hydrophobic backbone and a plurality of  
15 hydrophilic functional groups attached to the hydrophobic backbone. The hydrophobic  
16 backbone is the reaction product of one or more molecules having polymerizable alkene  
17 or alkyne functional groups, for example an oligo- or poly-ethylene structure. In contrast  
18 the hydrophilic functional groups can be zwitterionic surfactant functional groups,  
19 anionic surfactant functional groups, cationic surfactant functional groups, and nonionic  
20 surfactant functional groups. The illustrative embodiment is molecularly designed such  
21 that the combination of the viscoelastic surfactant and surfactant-polymer compound  
22 forms micellar assemblies in the wellbore fluid. In one illustrative and exemplary  
23 embodiment the acid form of the surfactant-polymer compound has the structure:

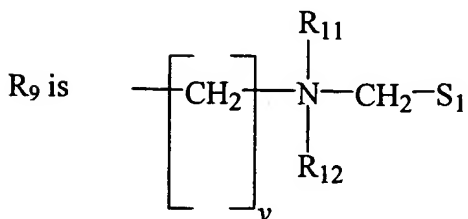
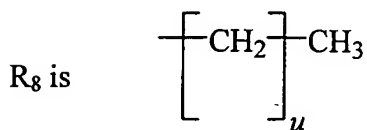
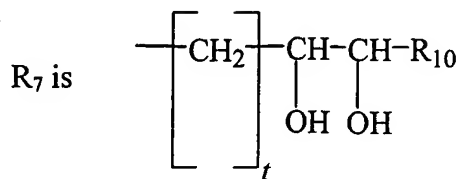


in which x has a value of 2 to 300,000 and preferably x has a value of 2 to 36  
 Alternatively, the surfactant-polymer compound can be a salt of oligo- or poly-( $\alpha$ -alkenyl  
 - $\omega$ - or  $\alpha$ -alkynyl - $\omega$ -quaternary-ammonio-N,N-dialkyl-N-alkylcarboxylate) or a mixture  
 further comprising a salt of N-alkyl-N-carboxymethyl-N,N-dimethylammonium chloride.  
 Another alternative and illustrative embodiment is where the surfactant-polymer  
 compound is a salt of oligo- or poly-(1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-  
 methylcarboxylate) or a mixture further comprising a salt of N-hexadecyl-N-  
 carboxymethyl-N,N-dimethylammonium chloride. A third alternative is where the  
 surfactant-polymer compound is a salt of oligo- or poly-( $\alpha$ -alkenyl - $\omega$ - or  $\alpha$ -alkynyl - $\omega$ -  
 quaternary-ammonio-N,N-dialkyl-N-alkylcarboxylate), or is a salt of oligo- or poly-(1-  
 hepten-7-quaternary-ammonio-N,N-dimethyl-N-methylcarboxylate).

Additional embodiments of the claimed subject matter include zwitterionic  
 surfactant heads such that the polymers or oligomers have the following structures:



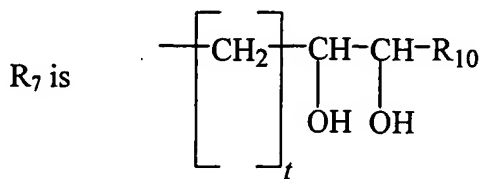
in which  $R_1, R_2, R_3, R_4, R_5, R_6 = H$  or  $CH_3$



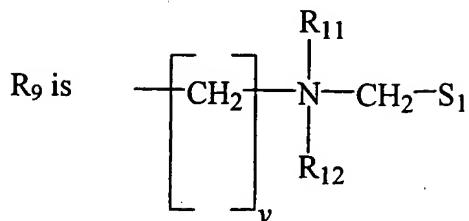
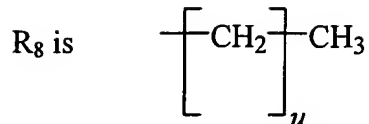
and in which  $R_{10}, R_{11}, R_{12} = H$  or  $CH_3$ , and  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  $3$ , and  $x + y + z = 3$  to  $300,000$  and  $S_1 = CO_2^-$  or  $SO_3^-$ . In a preferred illustrative embodiment,  $t = 12$  to  $16$ ,  $\mu = 6$  to  $12$ ;  $v = 12$  to  $18$ ,  $w = 1$  to  $3$ ,  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$  and  $z = 0$  to  $10,000$

Alternatively the oligomer or polymer compound can be cationic in the surfactant head and thus have a structure such as:

in which  $R_1, R_2, R_3, R_4, R_5, R_6 = H$  or  $CH_3$



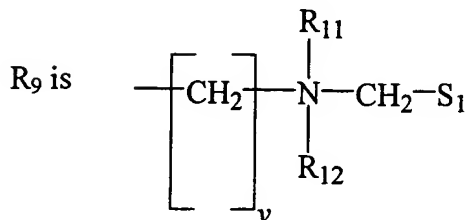
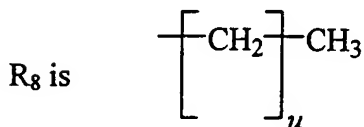
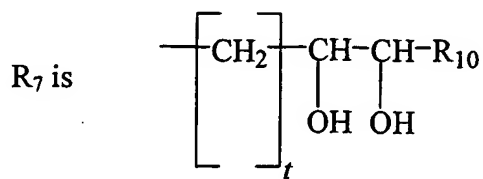




11 and in which  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ , and  $x + y + z$   
12  $= 3$  to  $300,000$ . An especially preferred and illustrative embodiment includes an  
13 oligomer or polymer in which  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $w = 1$  to  $3$ , and  $x =$   
14  $0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

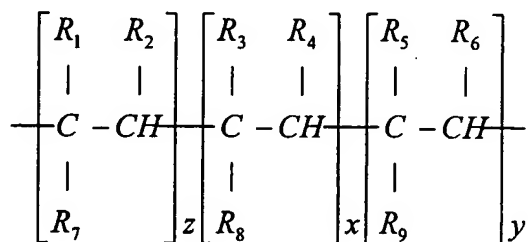
15 In yet another illustrative embodiment, the oligomer or polymer can have a  
16 molecular structure that includes an anionic surfactant functional group such as:

17  
18  
19 where  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$

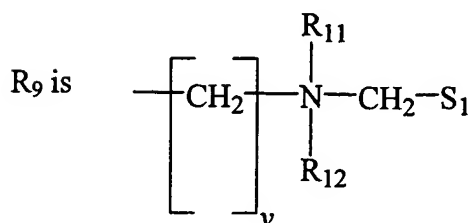
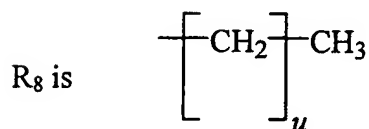
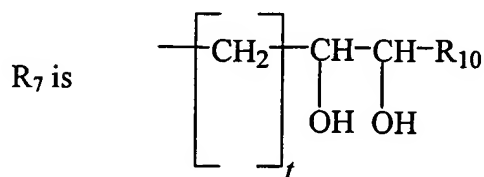


in which  $R_{10}, R_{11}, R_{12} = \text{H or CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $x + y + z = 3$  to  $300,000$ , and  $S_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ . In one such illustrative embodiment, it is preferred that  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

Further as noted above, the illustrative oligomer or polymer can have a nonionic surfactant group and preferably has a molecular structure such as:



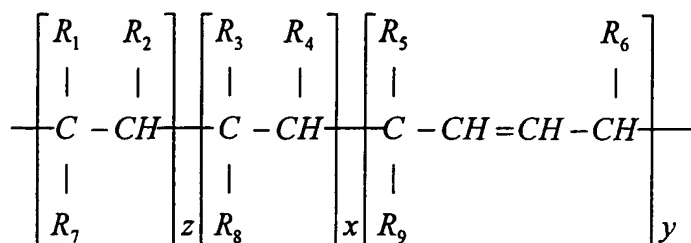
in which  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H or CH}_3$



and in which,  $R_{10}, R_{11}, R_{12} = \text{H or CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  $12$ , and  $x + y + z = 3$  to  $300,000$ . In such instances, a preferred illustrative embodiment is

achieved when t = 12 to 16, u = 6 to 12, v = 12 to 18, w = 1 to 3, and x = 0 to 10,000, y = 2 to 300,000, and z = 0 to 10,000.

Fundamentally the polymeric backbone can be saturated as noted in the above illustrative examples or unsaturated. In such illustrative embodiments, the oligomer or polymer has a back bone structure such as the following:



in which R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, R<sub>6</sub> = H or CH<sub>3</sub>. As for the R<sub>7</sub>, R<sub>8</sub> and R<sub>9</sub> groups, these may be the same as those disclosed above. Thus one of skill in the art should appreciate that zwitterionic, cationic, anionic and nonionic surfactant groups may be attached to the unsaturated backbone structure shown above and that such compounds are illustrative of the disclosed and claimed subject matter.

As discussed above, the novel oligomers and polymers taught in accordance with the claimed subject matter contain chemical functional groups that are similar to those found in conventional viscoelastic surfactants and thus are highly compatible with conventional viscoelastic surfactant systems. Accordingly, the claimed subject matter teaches blends of the viscoelastic oligomers or polymers with conventional viscoelastic surfactant systems. The conventional viscoelastic-surfactant-based fluids useful for the claimed subject matter are those in the following references, all of which are incorporated herein by reference – Canadian Patent 1,298,697, U. S. Patents 4,615,825, 4,695,389, 4,725,372, 5,258,137, 5,551,516, 5,691,292, 5,964,295, 5,965,502, 5,979,555, 5,979,557, 6,140,277, 6,194,355, 6,194,356, 6,211,120, 6,232,274, 6,239,183, Paper SPE 17,168, Paper SPE 30,098, Paper SPE 30,114, Paper SPE 30,458, Paper SPE 31,114, Paper SPE 38,622, Paper SPE 56,467, Paper SPE 57,432, Paper SPE 59,478, and Paper SPE 60,322. Of these, the preferred viscoelastic-surfactant-based fluids are those based on anionic, cationic, or zwitterionic surfactants or mixtures of anionic and nonionic surfactants or mixtures of cationic and nonionic surfactants or mixtures of zwitterionic and nonionic surfactants. And of these, the particularly preferred viscoelastic-surfactant-based fluids

1 are those based on zwitterionic surfactants or mixtures of zwitterionic and nonionic  
2 surfactants. In both the preferred viscoelastic-surfactant-based fluids, and in the  
3 particularly preferred viscoelastic-surfactant-based fluids, a minority amount of an  
4 additional surfactant, termed a "co-surfactant", such as, for example, 2-ethylhexanol or  
5 SDBS may optionally be employed. The viscoelastic oligomers or polymers of the  
6 claimed subject matter may be created in the presence of conventional viscoelastic  
7 surfactant systems or may be synthesized in separate steps, optionally processed or dried,  
8 and then subsequently admixed into a solution of conventional viscoelastic surfactants.

9 It has been found that the oligomers or polymers of the illustrative drilling fluids  
10 can be crosslinked with polyvalent metal ions, formaldehyde, or glutaraldehyde. In one  
11 such embodiment, the polyvalent metal ions are selected from:  $\text{Fe}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Ca}^{2+}$ ,  
12  $\text{Cu}^{2+}$ ,  $\text{UO}_2^{2+}$ ,  $\text{PbO}^{2+}$ ,  $\text{Al}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Cr}^{3+}$ ,  $\text{Ce}^{3+}$ ,  $\text{Ti}^{4+}$ ,  $\text{Zr}^{4+}$ ,  $\text{Sn}^{4+}$  and mixtures thereof.

13 One of skill in the art should also appreciate that the disclosed and claimed  
14 subject matter includes a method of making a wellbore fluid as is disclosed herein. In  
15 one such illustrative embodiment, the method involves the blending of an aqueous fluid  
16 phase, a viscoelastic surfactant, a water-soluble inorganic salt, and an oligomer or  
17 polymer soluble in the aqueous salt solution. The oligomer or polymer includes a  
18 hydrophobic oligomeric or polymeric backbone made from the oligomerization or  
19 polymerization of alkene or alkyne monomer groups, or mixtures thereof. The oligomer  
20 or polymer further comprises surfactant functional groups attached to the hydrophobic  
21 backbone, wherein the oligomer or polymer is hydrophilic in the surfactant functional  
22 groups and hydrophobic in the backbone hydrocarbon chain. Thus the oligomer or  
23 polymer has a molecular structure that promotes the formation of micellar assemblies  
24 such that the oligomers or polymers develop viscoelastic character prior to a  
25 polymerization step.

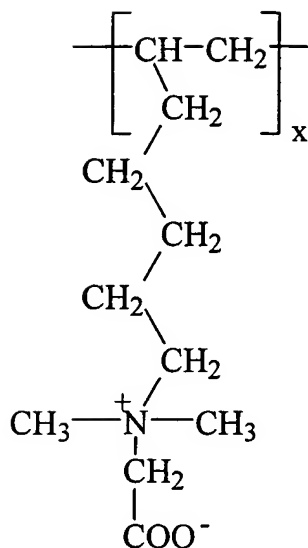
26 The claimed subject matter also encompasses a method of drilling a subterranean  
27 well as well as other uses for the wellbore fluid that should be apparent to one of skill in  
28 the art.

29 Further details regarding the claimed subject matter can be found in the following  
30 description.

## DESCRIPTION

The claimed subject matter relates to compositions for increasing the thermal durability of viscoelastic-surfactant-based well fluids. More specifically, the claimed subject matter relates to the fields of fluid rheology, thickeners, viscosifiers, viscoelastic fluids, viscoelastic surfactant fluids, drilling fluids, well fracturing fluids, well treatment fluids and fluid control pills. Further, the claimed subject matter relates to increasing the thermal and pressure stability of well fluids viscosified using viscoelastic surfactants by including an effective amount of a surfactant-oligomeric or surfactant-polymeric compound that has a thermally stable backbone structure and viscoelastic surfactant appendages. As the term is used in the present disclosure, "effective" simply means an amount sufficient to raise the temperature stability of the viscoelastic-surfactant based well fluid system by a measurable amount.

In accordance with one illustrative embodiment of the claimed subject matter, a sufficient quantity of at least one surfactant-oligomeric or surfactant-polymeric compound that is soluble in an aqueous salt solution is employed to affect the desired viscosity. In the claimed subject matter, the molecules of the surfactant-oligomeric or surfactant-polymeric compound have a hydrophobic oligomeric or polymeric backbone made preferably from the oligomerization or polymerization of alkene and/or alkyne groups. As the term is used herein, "thickener" and surfactant-oligomeric or surfactant-polymeric compound are used interchangeably and are intended to mean the compounds substantially described and claimed herein. The thickener of the claimed subject matter also includes chemical functional groups that are structurally similar to prior art viscoelastic surfactants and therefore these molecules exhibit similar chemical characteristics of prior art viscoelastic surfactants. Thus the hydrophobic backbone is chemically linked to and thus rendered at least in part hydrophilic by the presence of these chemical functional groups. One such illustrative compound is the product of the oligomerization reaction of a monomer such as the sodium salt of N-N-dimethyl-N-methylcarboxylate-N-1-hepten-7-ammonium chloride to give the sodium salt of oligo-(1-hepten-7-quaternary-ammonio-N-N-dimethyl-N-methylcarboxylate). The resulting oligomer is believed to have the simplified structure as indicated below in the acid form rather than the sodium-salt form:



in which x will have a value from about 2 to several hundred thousand, preferably from about 2 to several dozen. The monomer may be prepared, for example, by the reaction of N-hexadecyl-N,N-dimethylamine with chloroacetic acid to produce N-hexadecyl-N-methylcarboxylic acid-N,N-dimethylammonium chloride. Upon neutralization with sodium hydroxide, the final product is the zwitterionic betaine which is the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-dimethylammonium chloride – it has a negative charge on the carboxyl group and the sodium cation associated with it as a counter ion, and a positive charge on the quaternary amine group and the chloride anion associated with it as a counter ion. Alternatively, the sodium and chloride counter ions may be separated therefrom, leaving the negatively charged carboxyl group and the positively charged quaternary amine group as counter ions for each other.

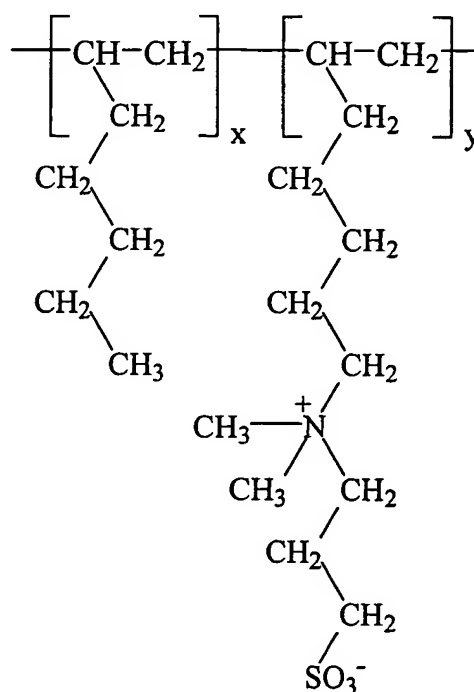
The sodium salt of oligo-(1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-methylcarboxylate) is an example of a salt of oligo- or poly-( $\alpha$ -alkenyl- $\omega$ -quaternary-ammonio-N,N-dialkyl-N-alkylcarboxylate). This example begins to define the terms  $\alpha$ -alkenyl - $\omega$ - and  $\alpha$ -alkynyl - $\omega$ -, wherein the “ $\alpha$ -“ designation denotes a location at or near one end of the hydrocarbyl chain, such as, for example at the 1-, 2-, or 3- position and wherein the “ $\omega$ -“ designation denotes a location at or near the opposite end of the hydrocarbyl chain from the  $\alpha$ -position, such as, for example at the very end of the hydrocarbyl chain, at one carbon group away from the very end of the hydrocarbyl chain,

or at two carbon groups away from the very end of the hydrocarbyl chain. An  $\alpha$ -alkene and an  $\alpha$ -alkyne are defined similarly. An  $\alpha$ -alkenyl- $\omega$ -carboxylate salt and an  $\alpha$ -alkynyl- $\omega$ -carboxylate salt are defined in a parallel manner. An  $\alpha$ -alkenyl- $\omega$ -N,N,N-trialkylammonium salt and an  $\alpha$ -alkynyl- $\omega$ -N,N,N-trialkylammonium salt are defined similarly. An  $\alpha,\alpha+2$ -alkadienyl- $\omega$ -carboxylate salt and an  $\alpha,\alpha+2$ -alkadiynyl- $\omega$ -carboxylate salt are defined in a parallel manner.

In one illustrative embodiment, a monomer like the sodium salt of N-carboxymethyl-N,N-dimethyl-N-1-hepten-7-ammonium chloride could be mixed into the solution of the conventional rod-shaped or spaghetti-like or spiraling-cylinder-like micelles of the viscoelastic surfactant that is the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-dimethylammonium chloride. This illustrative monomer, the sodium salt of N-carboxymethyl-N,N-dimethyl-N-1-hepten-7-ammonium chloride, differs only in minor ways from the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-dimethylammonium chloride. Accordingly, the monomer will be readily subsumed into the conventional rod-shaped or spaghetti-like or spiraling-cylinder-like micelles, whereupon one can take well-known steps to initiate the oligomerization or polymerization of the monomer to produce the sodium salt of oligo- or poly-(1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-methylcarboxylate). This oligomer or polymer will be inherently hydrophilic in its zwitterionic functional groups and hydrophobic in the hydrocarbon chains that link all the zwitterionic functional groups to each other. The oligomer or polymer is believed to be structurally quite similar to the viscoelastic surfactant molecules in the well fluid and therefore is soluble or dispersible in the well fluid solution. Similar oligomers or polymers are likewise soluble or dispersible in other viscoelastic surfactant solutions such as 10% XE862 (a product that is commercially available from Schlumberger) solution with 0.3% sodium dodecylbenzene sulfonate (SDBS) in 13.5 pound per barrel  $\text{CaBr}_2$ -based brine. Upon mixing the oligomer or polymer with the viscoelastic-surfactant-based fluid, the oligomer or polymer is believed to be subsumed into the rod-shaped or spaghetti-like micelles, whereupon, sitting inside these rod-shaped or spaghetti-like or spiraling-cylinder-like micelles is one or more such oligomer or polymer molecules and when configured in this fashion, the oligomer or polymer molecules impart greater thermal stability to the micelles, greater resistance to

shear stress and other stress conditions acting upon a fluid loss pill – including, for example, exposure to oil, high shear in pumping and placement, high temperature, high differential pressure, and low pH.

Another illustrative embodiment of the compounds of the claimed subject matter includes the oligomeric or polymeric products of the co-oligomerization or co-polymerization of two different monomers such as 1-heptene and the sodium salt of N-carboxymethyl-N,N-dimethyl-N-1-hepten-7-ammonium chloride. Alternatively, 1-heptene can be first co-oligomerized or co-polymerized with N,N-dimethyl-1-hepten-7-amine and in a subsequent reaction the amine groups are reacted with 1-chloro-propane-3-sulfonic acid and then caustic to produce oligo- or poly-(1-heptene-co-1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-propylsulfonate). The resulting oligomer or polymer is believed to have the simplified structure as indicated below:



in which x and y will have values dependent upon the molar ratio of reactants added to the polymerization reaction and the sum of x and y will have a value of about 2 to several hundred thousand, preferably from about 2 to several dozen. While this oligomer or polymer will be inherently hydrophilic in its zwitterionic functional groups, there are two sources of hydrophobicity: in the 1-heptene co-oligomer or co-polymer species and in the hydrocarbon chains that link all the zwitterionic functional groups to each other. By



1 varying the molar ratio of the monomers present during the oligomerization or  
2 polymerization reaction, the hydrophobicity and viscoelastic surfactant properties of the  
3 resulting oligomer or polymer may be controlled. The oligomer or polymer contains a  
4 point of structural similarity to the small molecule viscoelastic surfactants and therefore  
5 is soluble or dispersible in such conventional viscoelastic surfactant solutions. Similar  
6 oligomers or polymers are likewise soluble or dispersible in the 10% XE862 solution  
7 with 0.3% SDBS in 13.5 pounds per gallon  $\text{CaBr}_2$ -based brine discussed above. Upon  
8 admixing the polymer with the viscoelastic-surfactant-based fluid, the polymer is  
9 believed to be subsumed into the rod-shaped or spaghetti-like or spiraling-cylinder-like  
10 micelles, whereupon, sitting inside these rod-shaped or spaghetti-like or spiraling-  
11 cylinder-like micelles is one or more such polymer molecules and when configured in  
12 this fashion, the polymer molecules impart greater thermal stability to the micelles,  
13 greater resistance to shear stress and other stress conditions acting upon a fluid loss pill –  
14 including, for example, exposure to oil, high shear in pumping and placement, high  
15 differential pressure, and low pH.

16 Yet a third illustrative embodiment of the compounds of the claimed subject  
17 matter includes the co-oligomerization or co-polymerization of three or more co-  
18 monomers. For example, one could co-oligomerize or co-polymerize 1-heptene, 1-  
19 heptene-6,7-diol and N,N-dimethyl-1-hepten-7-amine. In such an illustrative  
20 embodiment, the amine groups are reacted with 1-chloro-propane-3-sulfonic acid and  
21 then caustic to produce oligo- or poly- (1-heptene -co-1-heptene-6,7-diol -co-1-hepten-7-  
22 quaternary-ammonio-N,N-dimethyl-N-propylsulfonate). The resulting oligomer or  
23 polymer is believed to have the simplified structure as indicated below:

24



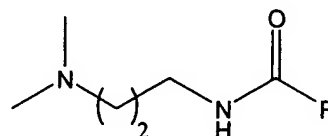
1 vicinal diol functionality to present itself at the outer surface of any rod-shaped micelle in  
2 which it has become subsumed.

3 The vicinal diol functionality provides the chemical functional group that renders  
4 the oligomers or polymers readily crosslinkable with polyvalent metal ions or complexes  
5 such as, for example, a borate, titanate, or zirconate crosslinker as taught in U. S. Patent  
6 5,062,969, (2) divalent, trivalent, or tetravalent cations such as, for example,  $\text{Fe}^{2+}$ ,  $\text{Cd}^{2+}$ ,  
7  $\text{Co}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{UO}_2^{2+}$ ,  $\text{PbO}^{2+}$ ,  $\text{Al}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Cr}^{3+}$ ,  $\text{Ce}^{3+}$ ,  $\text{Ti}^{4+}$ ,  $\text{Zr}^{4+}$ ,  $\text{Sn}^{4+}$ , and the like, (3)  
8 complexes of or other moieties containing the crosslinkers listed above in the first two  
9 categories, such as, for example, the tetrammine complex of the  $\text{Cu}^{2+}$  cation, the  
10 carbonate anion complexes of the  $\text{UO}_2^{2+}$  cation,  $\text{UO}_2(\text{CO}_3)_2^{2-}$  and  $\text{UO}_2(\text{CO}_3)_3^{4-}$ , or the  
11 triethanolamine complex of the  $\text{Ti}^{4+}$  cation, (4) so-called "organic crosslinkers" such as,  
12 for example, formaldehyde, and glutaraldehyde, and (5) mixtures of the crosslinkers  
13 listed above in the first four categories and/or reaction products therefrom. Thus it is  
14 contemplated that one of skill in the art could, if desired, crosslink the illustrative  
15 oligomers or polymers with polyvalent metal ions, complexes, organic crosslinkers, or  
16 mixtures thereof, as described above. This is believed to lead effectively to the  
17 crosslinking of the viscoelastic surfactant assemblies in which the oligomers or polymers  
18 have been subsumed.

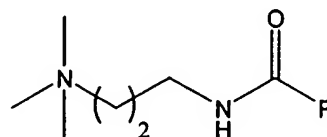
19 One of skill in the art should at this point appreciate that the co-oligomerization or  
20 co-polymerization reactions disclosed above result in the random placement of each co-  
21 monomer along the vinyl backbone of the illustrative oligomers or polymers. However,  
22 it is contemplated that block co-oligomerization or co-polymerization could also be used  
23 to achieve substantially similar results. Alternatively, block co-oligomerization or co-  
24 polymerization could also be used to carefully tailor the properties of the resulting  
25 oligomers or polymers made in accordance with the claimed subject matter. One of skill  
26 in the art should understand and appreciate that by systematically controlling the molar  
27 ratio of and concentration of monomers present in during the oligomerization or  
28 polymerization process, the order of addition, the temperature and duration of  
29 oligomerization or polymerization and the initiators and catalysts used and their  
30 concentrations, the properties of the compounds of the claimed subject matter can be  
31 carefully controlled and tailored.

Other oligomers or polymers formulated in accordance with and illustrative of the claimed subject matter include such oligomers or polymers as oligo- or poly-(N-carboxymethyl-N,N-diallyl-N-methyl ammonium chloride), oligo- or poly-(N,N-diallyl-N,N-dimethyl ammonium chloride-co-N-carboxymethyl-N,N-diallyl-N-methyl ammonium chloride), oligo- or poly-(1-butene-co-N-carboxymethyl-N,N-diallyl-N-methyl ammonium chloride), oligo- or poly- (1-butene-co-1-pentene-4,5-diol- co-N-carboxymethyl-N,N-diallyl-N-methyl ammonium chloride), oligo- or poly-alkenyl or -alkynyl DMAPA amides (see Figure below), oligo- or poly-alkenyl or -alkynyl DMAPA quats, and oligo- or poly-alkenyl or -alkynyl tallow amine quaternary amines such as

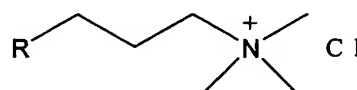
C<sub>12-22</sub> DMAPA Amide



C<sub>12-22</sub> DMAPA Amide Quaternary Amine



Tallow Amine Quaternary Amine



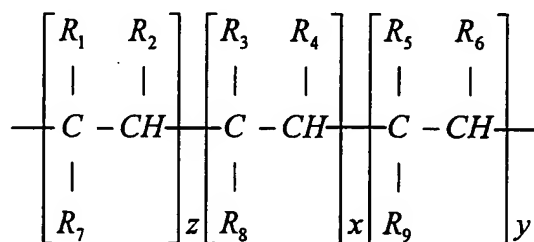
and the like.

Except for the oligo- or poly-alkenyl or -alkynyl DMAPA amides, oligo- or poly-alkenyl or -alkynyl DMAPA quaternary amines, and oligo- or poly-alkenyl or -alkynyl tallow amine quaternary amines, oligomers or polymers of this type may be unlike those of the previous illustrative embodiments of the claimed subject matter in that they are probably not capable of effectively vacuuming up the micelles and packing them about themselves until there are enough surfactant molecules present in the vicinity of the oligomer or polymer molecule so that the spherical micelles can merge into a single rod-shaped or spaghetti-like or spiraling-cylinder-like micelles with the oligomer or polymer molecule subsumed within the rod-shaped or spaghetti-like or spiraling-cylinder-like micelles. Polymers or oligomers in accordance with the present illustrative embodiment

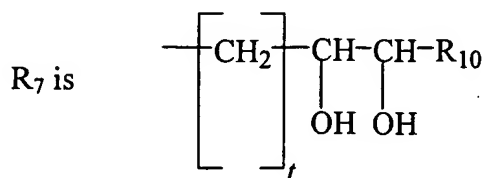
of the claimed subject matter are designed to be compatible with  $\text{CaBr}_2$ -,  $\text{CaBr}_2/\text{CaCl}_2$ -,  $\text{ZnBr}_2/\text{CaBr}_2$ -, and  $\text{ZnBr}_2/\text{CaBr}_2/\text{CaCl}_2$ -based brines.

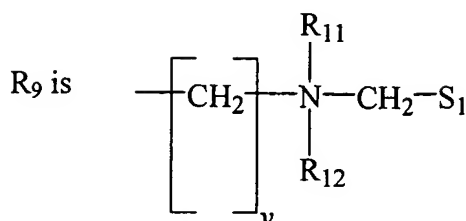
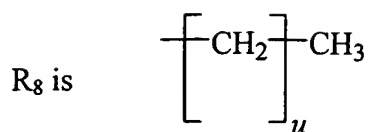
Other oligomers or polymers formulated in accordance with and illustrative of the claimed subject matter include oligomers or polymers such as oligo- or poly-(ethylene-co-N,N-diallyl-N,N-dimethyl ammonium chloride), oligo- or poly-(1-butene-co-N,N-diallyl-N,N-dimethyl ammonium chloride), oligo- or poly-(1-butene-co-1-pentene-4,5-diol-co-N,N-diallyl-N,N-dimethyl ammonium chloride), and the like. Polymers or oligomers of this type are probably not sufficiently compatible with  $\text{CaBr}_2$ -,  $\text{CaBr}_2/\text{CaCl}_2$ -,  $\text{ZnBr}_2/\text{CaBr}_2$ -, and  $\text{ZnBr}_2/\text{CaBr}_2/\text{CaCl}_2$ -based brines; however, they expand the scope of the claimed subject matter into useful dimensions to the extent that these polymers should be quite compatible with  $\text{NaCl}$ -,  $\text{NaBr}$ -,  $\text{NaBr}/\text{NaCl}$ -,  $\text{CsBr}/\text{KBr}$ -, and  $\text{CsBr}/\text{KBr}/\text{NaCl}$ -based brines and the like. Poly- or oligo-(1-butene-co-1-pentene-4,5-diol-co-N,N-diallyl-N,N-dimethyl ammonium chloride) incorporates a vicinal diol functionality, and so one of skill in the art should appreciate that this polymer or oligomer may be crosslinkable with polyvalent metal ions as disclosed above. As noted above, when the oligomers or polymers are crosslinked with polyvalent metal ions, we also effectively crosslink the viscoelastic surfactant assemblies in which the oligomers or polymers have been subsumed.

Additional embodiments of the claimed subject matter include zwitterionic surfactant heads such that the polymers or oligomers have the following structures:



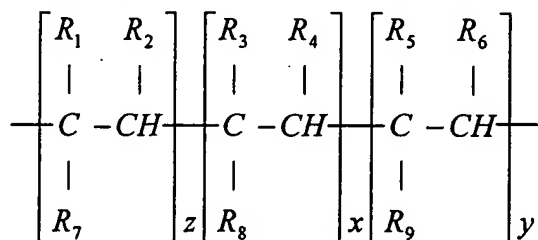
in which  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H or CH}_3$



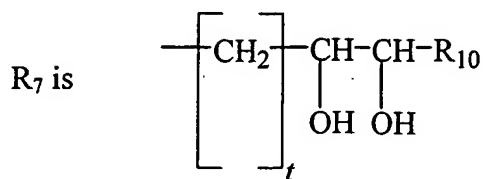


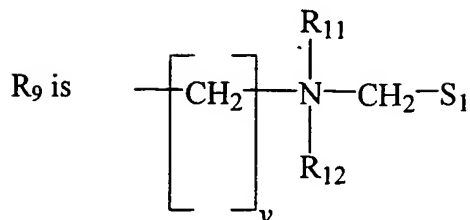
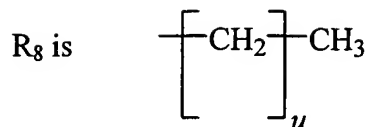
12 and in which  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ , and  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  
13  $3$ , and  $x + y + z = 3$  to  $300,000$  and  $\text{S}_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ . In a preferred illustrative  
14 embodiment,  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ;  $v = 12$  to  $18$ ,  $w = 1$  to  $3$ ,  $x = 0$  to  $10,000$ ,  $y = 2$  to  
15  $300,000$  and  $z = 0$  to  $10,000$

16 Alternatively the oligomer or polymer compound can be cationic in the surfactant  
17 head and thus have a structure such as:



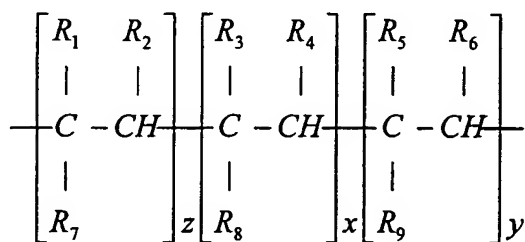
19  
20 in which  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



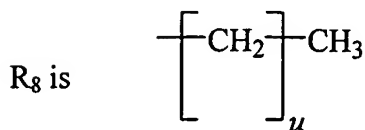
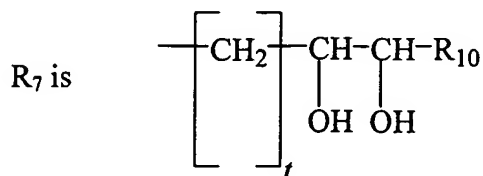


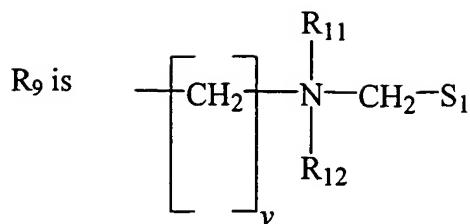
and in which  $R_{10}, R_{11}, R_{12} = \text{H or CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ , and  $x + y + z = 3$  to  $300,000$ . An especially preferred and illustrative embodiment includes an oligomer or polymer in which  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $w = 1$  to  $3$ , and  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

In yet another illustrative embodiment, the oligomer or polymer can have a molecular structure that includes an anionic surfactant functional group such as:



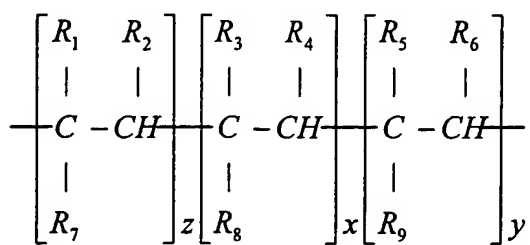
where  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H or CH}_3$



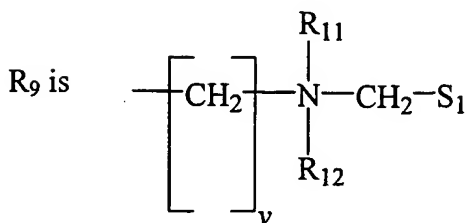
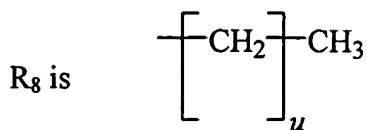
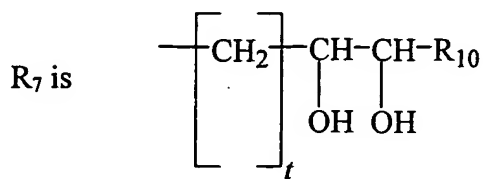


in which  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $x + y + z = 3$  to  $300,000$ , and  $S_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ . In one such illustrative embodiment, it is preferred that  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

Further as noted above, the illustrative oligomer or polymer can have a nonionic surfactant group and preferably has a molecular structure such as:



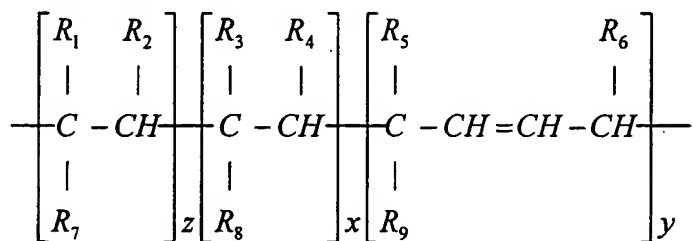
in which  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$





and in which,  $R_{10}, R_{11}, R_{12} = \text{H or CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  $12$ , and  $x + y + z = 3$  to  $300,000$ . In such instances, a preferred illustrative embodiment is achieved when  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $w = 1$  to  $3$ , and  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

Fundementally the polymeric backbone can be saturated as noted in the above illustrative examples or unsaturated. In such illustrative embodiments, the oligomer or polymer has a back bone structure such as the following:



in which  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H or CH}_3$ . As for the  $R_7, R_8$  and  $R_9$  groups, these may be the same as those disclosed above. Thus one of skill in the art should appreciate that zwitterionic, cationic, anionic and nonionic surfactant groups may be attached to the unsaturated backbone structure shown above and that such compounds are illustrative of the disclosed and claimed subject matter.

As discussed above, the novel oligomers and polymers taught in accordance with the claimed subject matter contain chemical functional groups that are similar to those found in conventional viscoelastic surfactants and thus are highly compatible with conventional viscoelastic surfactant systems. Accordingly, the claimed subject matter teaches blends of the viscoelastic oligomers or polymers with conventional viscoelastic surfactant systems. The conventional viscoelastic-surfactant-based fluids useful for the claimed subject matter are those in the following references, all of which are incorporated herein by reference – Canadian Patent 1,298,697, U. S. Patents 4,615,825, 4,695,389, 4,725,372, 5,258,137, 5,551,516, 5,691,292, 5,964,295, 5,965,502, 5,979,555, 5,979,557, 6,140,277, 6,194,355, 6,194,356, 6,211,120, 6,232,274, 6,239,183, Paper SPE 17,168,

1 Paper SPE 30,098, Paper SPE 30,114, Paper SPE 30,458, Paper SPE 31,114, Paper SPE  
2 38,622, Paper SPE 56,467, Paper SPE 57,432, Paper SPE 59,478, and Paper SPE 60,322.  
3 Of these, the preferred viscoelastic-surfactant-based fluids are those based on anionic,  
4 cationic, or zwitterionic surfactants or mixtures of anionic and nonionic surfactants or  
5 mixtures of cationic and nonionic surfactants or mixtures of zwitterionic and nonionic  
6 surfactants. And of these, the particularly preferred viscoelastic-surfactant-based fluids  
7 are those based on zwitterionic surfactants or mixtures of zwitterionic and nonionic  
8 surfactants. In both the preferred viscoelastic-surfactant-based fluids, and in the  
9 particularly preferred viscoelastic-surfactant-based fluids, a minority amount of an  
10 additional surfactant, termed a "co-surfactant", such as, for example, 2-ethylhexanol or  
11 SDBS may optionally be employed. The viscoelastic oligomers or polymers of the  
12 claimed subject matter may be created in the presence of conventional viscoelastic  
13 surfactant systems or may be synthesized in separate steps, optionally processed or dried,  
14 and then subsequently admixed into a solution of conventional viscoelastic surfactants.

15 The aqueous well fluids made in accordance with the claimed subject matter may  
16 optionally include a sufficient quantity of at least one water-soluble inorganic salt to  
17 effect formation stability. Typically, water-soluble potassium and ammonium salts, such  
18 as potassium chloride and ammonium chloride are employed. However, other shale  
19 inhibition compounds may be utilized including organic amine based compounds and  
20 other known shale inhibition agents. Additionally, calcium chloride, calcium bromide  
21 and zinc halide salts may also be used to increase the specific gravity (i.e., the density) of  
22 the solution. Formation stability and in particular clay stability are achieved at a  
23 concentration of a few percent by weight and as such the density of the fluid is not  
24 significantly altered by the presence of the inorganic salt unless fluid density becomes an  
25 important consideration, at which point, heavier inorganic salts may be employed.

26 Other compounds useful in the claimed subject matter include other viscosifiers,  
27 corrosion inhibitors, lubricants, pH control additives, surfactants, solvents, and/or  
28 weighting agents, among other additives. Some typical brine-based well fluid  
29 viscosifying additives include "natural" or biopolymers or derivatives thereof, such as,  
30 for example, xanthan gum and hydroxyethyl cellulose (HEC) or synthetic polymers and  
31 oligomers such as poly(ethylene glycol) (PEG), poly(diallyl amine), poly(acrylamide),

1 poly(aminomethylpropylsulfonate [AMPS]), poly(acrylonitrile), poly(vinyl acetate),  
2 poly(vinyl alcohol), poly(vinyl amine), poly(vinyl sulfonate), poly(styryl sulfonate),  
3 poly(acrylate), poly(methyl acrylate), poly(methacrylate), poly(methyl methacrylate),  
4 poly(vinylpyrrolidone), poly(vinyl lactam) and co-, ter-, and quater-polymers of the  
5 following co-monomers: ethylene, butadiene, isoprene, styrene, divinylbenzene, divinyl  
6 amine, 1,4-pentadiene-3-one (divinyl ketone), 1,6-heptadiene-4-one (diallyl ketone),  
7 diallyl amine, ethylene glycol, acrylamide, AMPS, acrylonitrile, vinyl acetate, vinyl  
8 alcohol, vinyl amine, vinyl sulfonate, styryl sulfonate, acrylate, methyl acrylate,  
9 methacrylate, methyl methacrylate, vinylpyrrolidone, and vinyl lactam. Yet other  
10 viscosifiers include the clay-based viscosifiers, especially laponite and other small  
11 fibrous clays such as the polygorskites (attapulgite and sepiolite).

12 One of ordinary skill in the art should appreciate that the use of "inert" filler  
13 materials can be added to impart strength to a fluid. Examples of such materials include  
14 shredded rubber tires, shredded battery casings, peanut hulls, cotton seed hulls, woody  
15 material, and other plant fibers that should be well known to one of skill in the art.

16 Additional brine-based well fluid surfactant additives useful in the claimed  
17 subject matter include nonionic surfactants, such as ethoxylated nonylphenols containing  
18 about 6 to 20 moles of ethylene oxide, or alkyl polyethyleneoxyalcohols, ethoxylated  
19 linear alcohols, ethoxylated tridecyl alcohols, ethoxylated phenols such as nonylphenols  
20 and dodecylphenols and fatty dialkanol amides. Hydroxyethyl fatty amines are also  
21 classified as nonionic surfactants, although at low pH, they may take on some cationic  
22 character. Emulsifying surfactants including oil soluble surfactants, such as fatty  
23 diethanolamides, sorbitan fatty acid esters, and ethoxylated sorbitan fatty acid esters such  
24 as sorbitan monooleate and sorbitan sesquioleate, because of their limited solubility in  
25 typical brines, may however be incidentally included in the formulation of other products  
26 – polymer solutions, emulsions, or slurries, corrosion inhibitors, lubricants, solvents, or  
27 weighting agents – that are used as additives to brine-based well fluids.

28 Other surfactants useful in the claimed subject matter are those listed in  
29 "McCutcheon's Emulsifiers and Detergents 1999: North American Edition" (ISBN:  
30 0944254624), incorporated herein by reference. They are classified as anionic, nonionic,  
31 amphoteric, zwitterionic, alcohols, alkanolamides, alkanolamines, alkylaryl sulfonates,

1 alkylaryl sulfonic acids, amine acetates, amine oxides, amines, sulfonated amines,  
2 sulfonated amides, betaine derivatives, block polymers, carboxylated alcohols,  
3 alkylphenol ethoxylates, carboxylic acids, fatty acids, ethoxylated alcohols, ethoxylated  
4 alkylphenols, ethoxylated amines, ethoxylated amides, ethoxylated fatty acids,  
5 ethoxylated fatty esters, fatty esters, fluorocarbon-based surfactants, glycerol esters,  
6 glycol esters, heterocyclic surfactants, imidazolines, imidazoline derivatives, isethionates,  
7 lanolin-based derivatives, lechithin, lechithin derivatives, methyl esters, monoglycerides,  
8 monoglyceride derivatives, olefin sulfonates, phosphate esters, phosphorous organic  
9 derivatives, polyethylene glycols, polymeric surfactants (polysaccharides, polyacrylic  
10 acids, polyacrylamides), propoxylated alcohols, propoxylated alkylphenols, propoxylated  
11 amines, propoxylated amides, propoxylated fatty acids, propoxylated fatty esters, protein-  
12 based surfactants, quaternary surfactants, sarcosamine derivatives, silicone-based  
13 surfactants, soaps, sodium isethionate, sorbitan derivatives, sucrose and glucose esters  
14 and derivatives, sulfates and sulfonates of oils and fatty acids, sulfates and sulfonates of  
15 ethoxylated alkylphenols, sulfates of alcohols, sulfates of ethoxylated alcohols, sulfates  
16 of fatty esters, sulfonates of hydrocarbons and petroleum, sulfosuccinates and derivatives,  
17 taurates, thio and mercapto derivatives, and tridecyl and dodecyl benzene sulfonic acids.

18 As mentioned earlier, the claimed subject matter concentrates on those micellar  
19 assemblies that are dominated by (1) one-dimensional extensions of the spherical micelle  
20 into rod-shaped or spaghetti-like or spiraling-cylinder-like micelles, (2) two-dimensional  
21 extensions of the spherical micelle into planar micelles or, more likely, stacks of planar  
22 micelles (as in liquid crystals), or (3) vesicular micelles. Vesicular micelles may be  
23 spherical at their very centers, but each center sphere will be surrounded by a second,  
24 third, and possibly more layers of surfactants. Moving outward radially from the central  
25 sphere of surfactant molecules, each successively outer layer of surfactant molecules will  
26 be oriented oppositely from next inner-most layer. Vesicular micelles may also be rod-  
27 shaped at their very centers, but each central rod-shaped assembly will be surrounded by  
28 a second, third, and possibly more layers of surfactants. A stack of planar micelles may  
29 be thought of as the vesicular form of a single planar micelle. These concepts comprise  
30 yet another illustrative embodiment of the claimed subject matter.

1           The above-mentioned embodiments may also comprise oligomers or polymers  
2 created by first carrying out a micellar polymerization to produce an oligomer or  
3 polymer, and then optionally processing or drying the product, and then subsequently  
4 admixing it into a solution of conventional surfactants – (1) non-viscoelastic or (2) non-  
5 viscoelastic and viscoelastic in nature. The micelles of the conventional surfactants  
6 would, in preferred instances, subsume the oligomers or polymers into the conventional  
7 surfactant micelles to create mixed micelles having superior viscoelastic properties in  
8 comparison to those of equivalent solutions not comprising the conventional surfactants  
9 and/or in comparison to those of equivalent solutions not comprising the oligomers or  
10 polymers.

11           It should also be appreciated by one of skill in the art that the claimed subject  
12 matter also relates to methods of using fluid loss control pills and similar fluids  
13 containing the compounds of the claimed subject matter, that can sustain stress conditions  
14 for extended periods of time without significant fluid loss or loss of desirable rheological  
15 properties. The stress conditions may include, for example, exposure to oil, high shear in  
16 pumping and placement, exposure to oxidizing breakers (including oxygen dissolved in  
17 the fluid), exposure to brines having high divalent cation content, high temperature, high  
18 differential pressure, low pH, extended time, and a combination of two or more of such  
19 stress conditions. These pills and fluids are advantageously applied in or in connection  
20 with drilling, drill-in, displacement, completion, hydraulic fracturing, work-over, packer  
21 fluid emplacement or maintenance, well treating, testing, or abandonment.

22           Exemplary viscoelastic surfactant compositions of improved stability in fluid loss  
23 control in accordance with the claimed subject matter are given in the following  
24 examples. The following examples are included to demonstrate preferred embodiments  
25 of the claimed subject matter. It should be appreciated by those of skill in the art that the  
26 techniques disclosed in the examples which follow represent techniques discovered by  
27 the inventors to function well in the practice of the claimed subject matter, and thus can  
28 be considered to constitute preferred modes for its practice. However, those of skill in  
29 the art should, in light of the present disclosure, appreciate that many changes can be  
30 made in the specific embodiments which are disclosed and still obtain a like or similar  
31 result without departing from the scope of the claimed subject matter.

1 **Example 1:** The oligomerization of the surfactant N-dodecene-1-yl-N,N-bis(2-  
2 hydroxyethyl)-N-methylammonium chloride:



6 in an aqueous solution was achieved by joining a few of the carbon-carbon double bonds  
7 by a free radical oligomerization reaction within the micelles using the following  
8 processes. A viscoelastic surfactant solution is produced using 30 g/l of the surfactant N-  
9 dodecene-1-yl-N,N-bis(2-hydroxyethyl)-N-methylammonium chloride with 40 g/l  
10 ammonium chloride. A volume of 100 ml of the viscoelastic surfactant solution was  
11 placed in a bottle purged with oxygen-free, dry nitrogen gas to remove any dissolved  
12 oxygen. After sufficient purging 10 mg of the free radical initiator 2,2'-azo(bis-  
13 amidinopropane) dihydrochloride was added to the viscoelastic surfactant solution and  
14 mixed thoroughly. The surfactant solution was heated at 60°C for 30 minutes under an  
15 atmosphere of oxygen-free, dry nitrogen. The resulting oligomer can be thought of as  
16 related to oligo-ethylene, a relatively short-chained polyethylene, to which relatively long  
17 pendant surfactant groups -  $[(\text{CH}_2)_{18}][\text{HO}(\text{CH}_2)_2]_2\text{CH}_3\text{N}^+\text{Cl}^-$  - are linked.  
18 Oligomerization of the surfactant monomers in micelles resulted in the viscosity of the  
19 gel becoming largely insensitive to contact with hydrocarbon. The viscosity of the  
20 surfactant gel was not materially altered by the oligomerization of the surfactant  
21 monomers. The oligomerized surfactant gel retained its gel strength after prolonged  
22 contact with water. A volume of 50 ml of the oligomerized viscoelastic surfactant  
23 solution was blended with 50 ml of the original un- oligomerized viscoelastic surfactant  
24 solution. The resulting solution was blended with 8 volume % of n-hexane with no  
25 material reduction in the viscosity of the mixture being observed, whereas mixing only 6  
26 volume % of n-hexane with the original un-oligomerized viscoelastic surfactant solution  
27 led to loss of essentially all viscosity in the mixture. Advantageously, therefore,  
28 compositions formed in accordance with the claimed subject matter provide viscoelastic  
29 surfactant fluids that are capable of controlling fluid loss, and that are capable of  
30 dissolving in fluid without leaving substantial amounts of residue. Further, the  
31 compositions are capable of controlling fluid loss at temperatures up to about 350°F or

higher, and may also control fluid loss for an extended period of at least 4 days, and finally, do not interfere with other chemical additives commonly used in the petroleum industry.

**Example 2:** The oligomerization of the surfactant potassium octadec-1-ene-18-oate:



in an aqueous solution was achieved using the following processes. The viscoelastic surfactant solution was formed by mixing 60 g/l potassium oleate with 60 g/l potassium chloride. A sample of 100 ml of the viscoelastic surfactant solution was purged with oxygen-free, dry nitrogen and mixed with 10 mg of the initiator 2,2'-azo(bisamidinopropane) dihydrochloride. The solution is heated at 60°C for 30 minutes under an atmosphere of oxygen-free, dry nitrogen. The resulting solution of oligomerized surfactants was about equally viscoelastic as the original monomeric solution but the observed viscoelasticity was insensitive to contact with hydrocarbon. The resulting oligomer can be thought of as related to oligo-ethylene, a relatively short-chained polyethylene, to which relatively long pendant groups –  $(\text{CH}_2)_{16}\text{CO}_2\text{K}^+$  – are linked. The gel formed by the oligomerized surfactant retained its viscoelasticity after prolonged contact with water. The observations led to the conclusion that the oligomer was viscoelastic in character, much as was the original un-oligomerized viscoelastic surfactant solution; but the oligomerized solution was more stable than the original un-oligomerized solution.

**Example 3:** The oligomerization of a long-chain vinyl surfactant, the potassium salt of octadeca-1,3-diene-17-oate:



in a viscoelastic solution was carried out at a concentration of 60 g/l in the presence of 40 g/l ammonium chloride. The surfactant monomers were oligomerized using 10 mg of the free radical initiator 2,2'-azo(bis-amidinopropane) dihydrochloride in 100 ml of viscoelastic surfactant solution which had been purged with oxygen-free, dry nitrogen

gas. The solution was heated at 60°C for 30 minutes under an atmosphere of oxygen-free, dry nitrogen. The resulting oligomer can be thought of as related to oligo-butadiene, a relatively short-chained polybutadiene, to which relatively long pendant groups –  $(\text{CH}_2)_{14}\text{CO}_2\text{K}^+$  – are linked. Oligomerization of the surfactant resulted in a semi-rigid gel that retained the viscoelasticity of the original monomeric surfactant solution but showed none of its sensitivity to contact with hydrocarbon or water. The observations led to the conclusion that the oligomer was viscoelastic in character, much as was the original un-oligomerized viscoelastic surfactant solution; but the oligomerized solution was more stable than the original un-oligomerized solution.

**Example 4:** Oligomers of N-(undec-11-enyl)-N,N-dimethylammonium chloride were synthesized by linking the head groups with a branched C<sub>9</sub> bridge. First, 5-methylnonane-1,8-diamine was reacted with 11-chloroundec-1-ene,  $\text{CH}_2=\text{CH}(\text{CH}_2)_9\text{Cl}$ , and subsequently quaternized to produce the following monomer:



When this monomer was oligomerized, the product was a surfactant oligomer in which the  $\text{N}^+\text{Cl}^-$  head groups, again, are almost completely free relative to their neighboring  $\text{N}^+\text{Cl}^-$  head groups – each being linked to each other through relatively long  $(\text{CH}_2)_4\text{CHCH}_3(\text{CH}_2)_4$  chains or to the oligo-ethylene backbone through relatively long pendant  $(\text{CH}_2)_9$  chains.

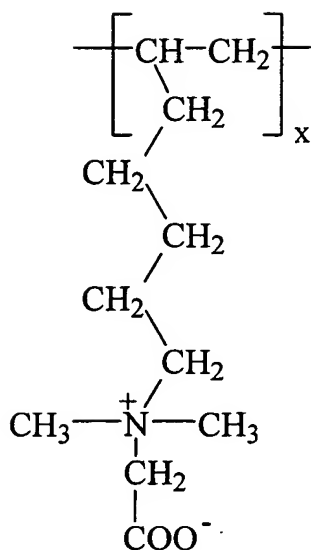
The preparation of the monomer described above began with 5-methylnonane-1,8-diamine. This precursor was reacted with 11-chloroundec-1-ene,  $\text{CH}_2=\text{CH}(\text{CH}_2)_9\text{Cl}$ , under conditions such that each diamine group was converted to secondary amines. In this first step in the synthesis, incompletely reacted material was separated from the di-secondary amine by distillation. The small part of the 5-methylnonane-1,8-diamine that had reacted too much and had become tertiarized or quaternized at one or both ends by excessive reaction with the 11-chloroundec-1-ene were separated by fractional crystallization from the desired N,N'-di-(1-decenyl)-5-methylnonane-1,8-diamine. Finally, this product was then reacted with an excess of methylbromide and quaternized to produce the monomer stated above. The oligomerization of this monomer:





in a viscoelastic solution was carried out at a concentration of 60 g/l in the presence of 40 g/l ammonium chloride. A volume of 100 ml of the viscoelastic surfactant solution was placed in a bottle purged with oxygen-free, dry nitrogen gas to remove any dissolved oxygen. After sufficient purging 10 mg of the free radical initiator 2,2'-azo(bis-amidinopropane) dihydrochloride was added to the viscoelastic surfactant solution and mixed thoroughly. The surfactant solution was heated at 60°C for 30 minutes under an atmosphere of oxygen-free, dry nitrogen. Oligomerization of the surfactant monomers in micelles resulted in the viscosity of the gel becoming largely insensitive to contact with hydrocarbon. The viscosity of the surfactant gel was not materially altered by the oligomerization of the surfactant monomers. The oligomerized surfactant gel retained its gel strength after prolonged contact with water. The observations led to the conclusion that the oligomer was viscoelastic in character, much as was the original un-oligomerized viscoelastic surfactant solution; but the oligomerized solution was more stable than the original un-oligomerized solution.

**Example 5:** The oligomerization of the viscoelastic surfactant monomer, the sodium salt of N-N-dimethyl-N-methylcarboxylate-N-1-hepten-7-ammonium chloride, was carried out to give oligo(1-hepten-7-quaternary-ammonio-N-N-dimethyl-N-methylcarboxylate), sodium salt. The resulting oligomer is believed to have the simplified structure as indicated below in the acid form rather than the sodium-salt form:

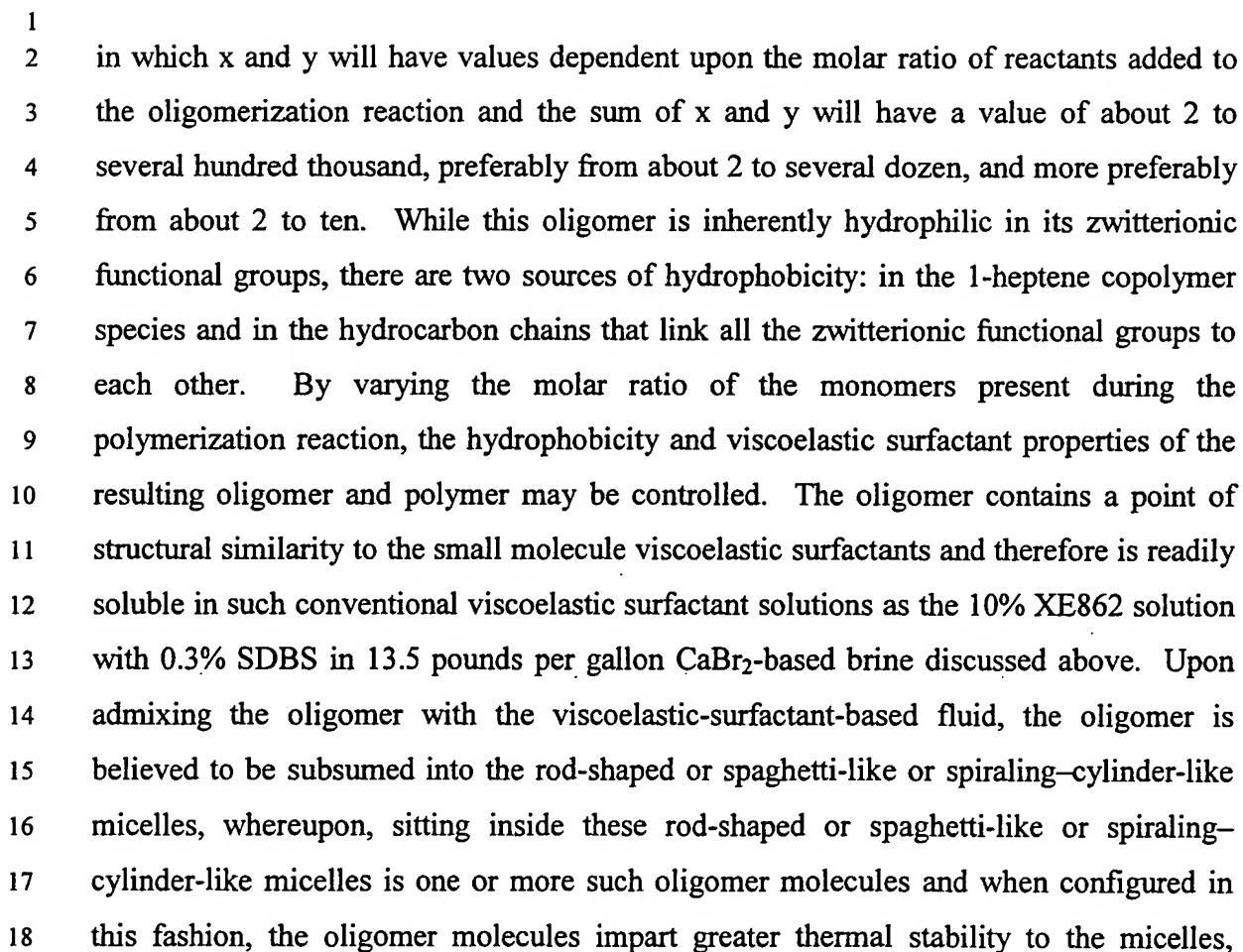


1  
2 in which x will have a value from about 2 to several hundred thousand, preferably from  
3 about 2 to several dozen, and more preferably from about 2 to perhaps 4. The monomer  
4 was prepared by the reaction of N-1-hept-6-enyl-N,N-dimethylamine with chloroacetic  
5 acid to produce N-1-hept-6-enyl-N-methylcarboxylic acid-N,N-dimethylammonium  
6 chloride. Upon neutralization with sodium hydroxide, the final product was the  
7 zwitterionic betaine which is the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-  
8 dimethylammonium chloride – it has a negative charge on the carboxyl group and the  
9 sodium cation associated with it as a counter ion, and a positive charge on the quaternary  
10 amine group and the chloride anion associated with it as a counter ion.

11 **Example 6:** The viscoelastic monomer, the sodium salt of N-carboxymethyl-N,N-  
12 dimethyl-N-1-hepten-7-ammonium chloride was mixed into the solution of the  
13 conventional rod-shaped or spaghetti-like or spiraling-cylinder-like micelles of the  
14 viscoelastic surfactant that is the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-  
15 dimethylammonium chloride. This illustrative monomer, the sodium salt of N-  
16 carboxymethyl-N,N-dimethyl-N-1-hepten-6-ammonium chloride, differs only in minor  
17 ways from the sodium salt of N-hexadecyl-N-carboxymethyl-N,N-dimethylammonium  
18 chloride. Accordingly, the monomer was readily subsumed into the conventional rod-  
19 shaped or spaghetti-like micelles, whereupon oligomerization was initiated to produce the  
20 sodium salt of oligo- (1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-

1 methylcarboxylate). This oligomer is inherently hydrophilic in its zwitterionic functional  
2 groups and hydrophobic in the hydrocarbon chains that link all the zwitterionic functional  
3 groups to each other. The oligomer is believed to be structurally quite similar to the  
4 viscoelastic surfactant molecules in the well fluid and therefore is readily soluble in the  
5 well fluid solution. the oligomers are likewise soluble or dispersible in other viscoelastic  
6 surfactant solutions such as 10% XE862 (a product that is commercially available from  
7 Schlumberger) solution with 0.3% SDBS in 13.5 pound per barrel CaBr<sub>2</sub>-based brine.  
8 Upon mixing the oligomer or polymer with the viscoelastic-surfactant-based fluid, the  
9 oligomer or polymer is believed to be subsumed into the rod-shaped or spaghetti-like or  
10 spiraling-cylinder-like micelles, whereupon, sitting inside these rod-shaped or spaghetti-  
11 like or spiraling-cylinder-like micelles is one or more such oligomer or polymer  
12 molecules and when configured in this fashion, the oligomer or polymer molecules  
13 impart greater thermal stability to the micelles, greater resistance to shear stress and other  
14 stress conditions acting upon a fluid loss pill – including, for example, exposure to oil,  
15 high shear in pumping and placement, high temperature, high differential pressure, and  
16 low pH.

17 **Example 7:** In another illustrative embodiment of the claimed subject matter, a sufficient  
18 quantity, about 10 volume%, of the co-monomer 1-heptene was mixed into a viscoelastic  
19 solution of the sodium salt of 1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-  
20 propylsulfonate chloride, causing the solution to lose most or all of its viscoelastic  
21 character. Then the 1-heptene was co-oligomerized with the sodium salt of 1-hepten-7-  
22 quaternary-ammonio-N,N-dimethyl-N-propylsulfonate chloride to produce oligo-(1-  
23 heptene-co-1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-propylsulfonate). The  
24 resulting oligomer is believed to have the simplified structure as indicated below:





1 surfactant-based fluid, the polymer is believed to be subsumed into the rod-shaped or  
2 spaghetti-like or spiraling-cylinder-like micelles, whereupon, sitting inside these rod-  
3 shaped or spaghetti-like or spiraling-cylinder-like micelles is one or more such polymer  
4 molecules and when configured in this fashion, the polymer molecules impart greater  
5 thermal stability to the micelles, greater resistance to shear stress and other stress  
6 conditions acting upon a fluid loss pill – including, for example, exposure to high shear in  
7 pumping and placement, high temperature, high differential pressure, and low pH.  
8 Additionally, the co-monomer 1-heptene-6,7-diol incorporates a vicinal diol functionality  
9 into the polymer. Surface forces should cause the vicinal diol functionality to present  
10 itself at the outer surface of any rod-shaped micelle in which it has become subsumed.  
11 The vicinal diol functionality was subsequently crosslinked with borate, leading  
12 effectively to the crosslinking of the viscoelastic surfactant assemblies in which the  
13 oligomers had been subsumed.

14 As discussed above, the novel oligomers or polymers taught in accordance with  
15 the claimed subject matter contain chemical functional groups that are similar to those  
16 found in conventional viscoelastic surfactants and thus are highly compatible with  
17 conventional viscoelastic surfactant systems. Accordingly, the claimed subject matter  
18 teaches blends of the viscoelastic oligomers or polymers with conventional viscoelastic  
19 surfactant systems. The viscoelastic-surfactant-based fluids useful for the claimed  
20 subject matter are those in the following citations, all of which are incorporated herein by  
21 reference – Canadian Patent 1,298,697, U. S. Patents 4,615,825, 4,695,389, 4,725,372,  
22 5,258,137, 5,551,516, 5,691,292, 5,964,295, 5,965,502, 5,979,555, 5,979,557, 6,140,277,  
23 6,194,355, 6,194,356, 6,211,120, 6,232,274, 6,239,183, Paper SPE 17,168, Paper SPE  
24 30,098, Paper SPE 30,114, Paper SPE 30,458, Paper SPE 31,114, Paper SPE 38,622,  
25 Paper SPE 56,467, Paper SPE 57,432, Paper SPE 59,478, and Paper SPE 60,322.

26 Furthermore, the oligomeric or polymeric viscoelastic surfactants may be added  
27 to a well fluid in substantially any convenient manner. Thus, the oligomeric or polymeric  
28 viscoelastic surfactants may be added directly to the well fluid either in solid form or in  
29 the form of an aqueous solution. Alternatively, the oligomeric or polymeric viscoelastic  
30 surfactants may be separately added to a solution already containing other surfactants or  
31 viscoelastic surfactants to provide a fluid loss controlling base solution, with the optional

1 crosslinking agents thereafter being added to the fluid loss controlling base solution  
2 immediately prior to use.

3 In preferred embodiments, the oligomeric or polymeric viscoelastic surfactants,  
4 including optional crosslinking agents, are supplied to the well fluid at a level of about 20  
5 ppm to about 20 parts per 100 parts of the well fluid, more preferably about 100 ppm to  
6 about 15 parts per 100 parts of the well fluid.

7 In more preferred embodiments, the oligomeric or polymeric viscoelastic  
8 surfactants, including optional crosslinking agents, are supplied to the well fluid already  
9 containing an amount of other surfactants or viscoelastic surfactants. When these more  
10 preferred embodiments are exercised, the oligomeric or polymeric viscoelastic surfactants  
11 are supplied at a level of about 20 ppmw to about 10 parts per 100 parts of the well fluid,  
12 preferably about 100 ppmw to about 5 parts per 100 parts of the well fluid.

13 Techniques for measuring fluid loss control are well known in the art and should  
14 be well known to one of skill and knowledge of the formulation of drilling fluids.  
15 Specifically, the fluid loss measurements of the present disclosure were made with  
16 reference to API Recommended Practice RP 13B-1, Second Edition, September 1997, pp.  
17 9-11, the contents of which are incorporated by reference.

18 While the apparatus, compositions and methods of the claimed subject matter  
19 have been described in terms of preferred or illustrative embodiments, it will be apparent  
20 to those of skill in the art that variations may be applied to the process described herein  
21 without departing from the concept and scope of the claimed subject matter. All such  
22 similar substitutes and modifications apparent to those skilled in the art are deemed to be  
23 within the scope and concept of the claimed subject matter.

24

1 **What is claimed is:**

2 1. A wellbore fluid comprising:

3 an aqueous based continuous phase;

4 a viscoelastic surfactant; and

5 a surfactant-polymer compound soluble in an aqueous solution, the surfactant-  
6 polymer compound having a hydrophobic backbone and a plurality of hydrophilic  
7 functional groups attached to the hydrophobic backbone,

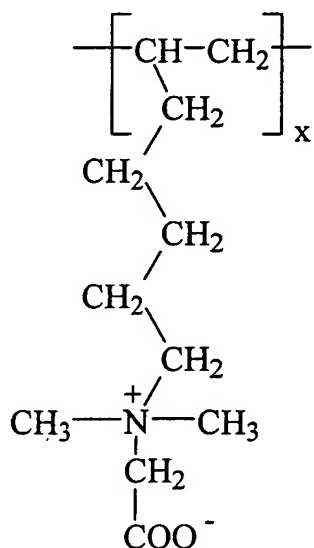
8 wherein the hydrophobic backbone is the reaction product of one or more  
9 molecules having polymerizable alkene or alkyne functional groups;

10 wherein the hydrophilic functional groups are selected from: zwitterionic  
11 surfactant functional groups, anionic surfactant functional groups, cationic surfactant  
12 functional groups, and nonionic surfactant functional groups; and,

13 wherein the combination of the viscoelastic surfactant and surfactant-polymer  
14 compound form micellar assemblies.

15  
16 2. The wellbore fluid of claim 1 further comprising a water-soluble inorganic salt.

17  
18 3. The wellbore fluid of claim 1 wherein the acid form of the surfactant-polymer  
19 compound has the structure:





wherein  $x = 2$  to 300,000.

4. The wellbore fluid of claim 3 wherein  $x = 2$  to 36.

5. The wellbore fluid of claim 1 wherein the surfactant-polymer compound is a salt of oligo- or poly-( $\alpha$ -alkenyl - $\omega$ - or  $\alpha$ -alkynyl - $\omega$ -quaternary-ammonio-N,N-dialkyl-N-alkylcarboxylate) or a mixture further comprising a salt of N-alkyl-N-carboxymethyl-N,N-dimethylammonium chloride.

6. The wellbore fluid of claim 1 wherein the surfactant-polymer compound is a salt of oligo- or poly-(1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-methylcarboxylate) or a mixture further comprising a salt of N-hexadecyl-N-carboxymethyl-N,N-dimethylammonium chloride.

7. The wellbore fluid of claim 1 wherein the surfactant-polymer compound is a salt of oligo- or poly-( $\alpha$ -alkenyl - $\omega$ - or  $\alpha$ -alkynyl - $\omega$ -quaternary-ammonio-N,N-dialkyl-N-alkylcarboxylate).

8. The wellbore fluid of claim 1 wherein the surfactant-polymer compound is a salt of oligo- or poly-(1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-methylcarboxylate).

9. A wellbore fluid comprising:

an aqueous fluid;

a viscoelastic surfactant;

a thickener soluble in the aqueous fluid,

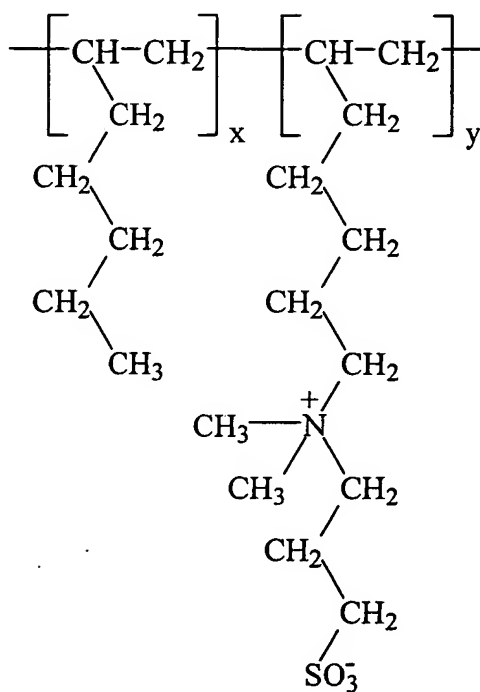
wherein the thickener has a hydrophobic oligomeric or polymeric backbone made from the reaction of alkene monomer or alkyne monomer, or mixtures thereof, and

wherein surfactant functional groups are attached to the hydrophobic backbone, wherein the surfactant functional group is selected from: zwitterionic surfactant functional groups, anionic surfactant functional groups, cationic surfactant functional groups, and nonionic surfactant functional groups; and,

wherein the thickener has a molecular confirmation such that the surfactant functional groups are hydrophilic and the hydrophobic oligomeric or polymeric backbone is hydrophobic; and

wherein the combination of viscoelastic surfactant and thickener form micellar assemblies such that the wellbore fluid thickener develops viscoelastic characteristics.

10. The wellbore fluid of claim 9 wherein the thickener has the following structure:



wherein  $x + y = 2$  to 300,000.

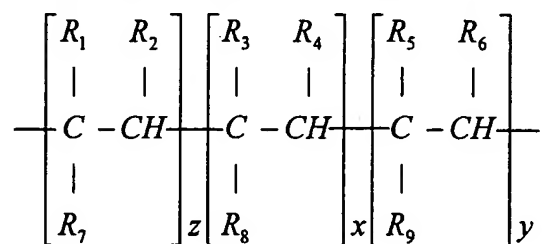
11. The wellbore fluid of claim 10 wherein  $x + y = 2$  to 36.

12. The wellbore fluid of claim 11 wherein the thickener is a salt of oligo- or poly-( $\alpha$ -alkene - $\omega$ - or  $\alpha$ -alkyne-co- $\alpha$ -alkenyl - $\omega$ - or  $\alpha$ -alkynyl - $\omega$ -quaternary-ammonio-N,N-dialkyl-N-alkylsulfonate).

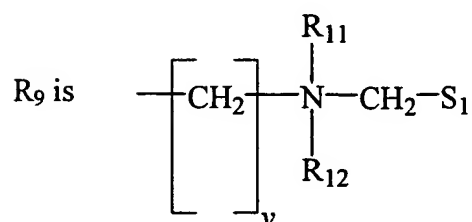
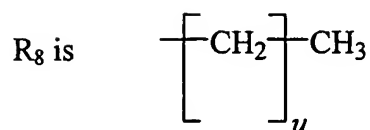
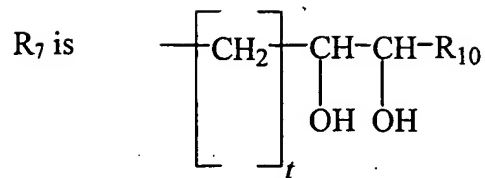
13. The wellbore fluid of claim 11 wherein the thickener is oligo- or poly-(1-heptene-co-1-hepten-7-quaternary-ammonio-N,N-dimethyl-N-propylsulfonate).

14. The wellbore fluid of claim 9 wherein the viscoelastic surfactant is selected from the following: anionic, nonionic, amphoteric, zwitterionic, alcohols, alkano-lamides, alkanolamines, alkylaryl sulfonates, alkylaryl sulfonic acids, amine acetates, amine oxides, amines, sulfonated amines, sulfonated amides, betaines, block polymers, carboxylated alcohols, alkylphenol ethoxylates, carboxylic acids, fatty acids, ethoxylated alcohols, ethoxylated alkylphenols, ethoxylated amines, ethoxylated amides, ethoxylated fatty acids, ethoxylated fatty esters, fatty esters, fluorocarbon-based surfactants, glycerol esters, glycol esters, heterocyclic surfactants, imidazolines, isethionates, lanolins, lecithins, methyl esters, monoglycerides, olefin sulfonates, phosphate esters, polyethylene glycols, polysaccharides, polyacrylic acids, polyacrylamides, propoxylated alcohols, propoxylated alkylphenols, propoxylated amines, propoxylated amides, propoxylated fatty acids, propoxylated fatty esters, protein-based surfactants, quaternary surfactants, sarcosamines, silicone-based surfactants, soaps, sodium isethionate, sorbitans, sucrose and glucose esters, sulfates and sulfonates of oils and fatty acids, sulfates and sulfonates of ethoxylated alkylphenols, sulfates of alcohols, sulfates of ethoxylated alcohols, sulfates of fatty esters, sulfonates of hydrocarbons and petroleum, sulfosuccinates, taurates, and tridecyl and dodecyl benzene sulfonic acids and mixtures thereof.

15. The wellbore fluid of claim 9 wherein the thickener has the following structure:



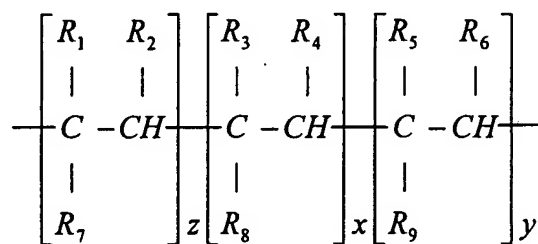
where  $R_1, R_2, R_3, R_4, R_5, R_6 = H$  or  $CH_3$



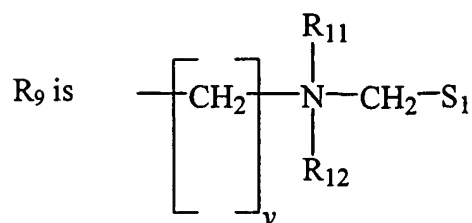
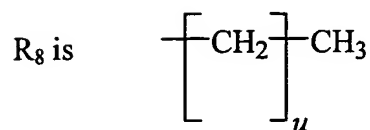
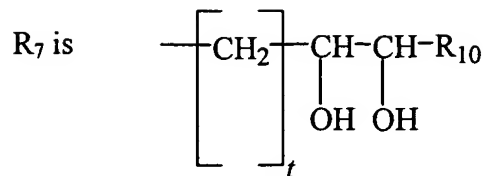
11  
12  
13  
14  
15 wherein  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ , and  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  $3$ ,  
16 and  $x + y + z = 3$  to  $300,000$  and  $\text{S}_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ .

17  
18 16. The wellbore fluid of claim 15 wherein  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ;  $v = 12$  to  $18$ ,  $w = 1$   
19 to  $3$ ,  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$  and  $z = 0$  to  $10,000$

20  
21 17. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following  
22 structure:



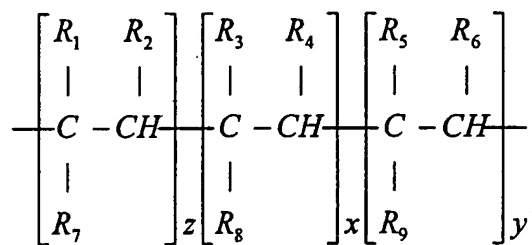
24  
25  
26 wherein  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



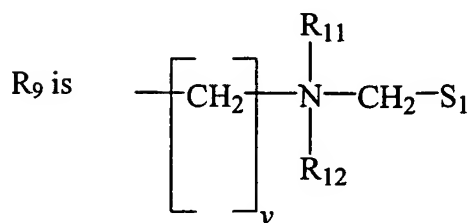
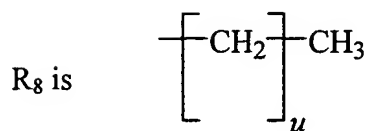
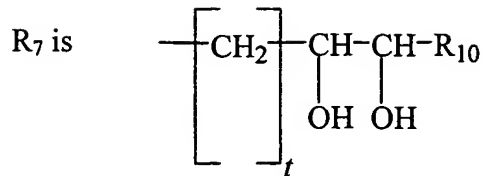
wherein  $R_{10}$ ,  $R_{11}$ ,  $R_{12}$  = H or  $\text{CH}_3$ ,  $t$  = 1 to 16,  $u$  = 6 to 12,  $v$  = 1 to 18, and  $x + y + z$  = 3 to 300,000.

18. The wellbore fluid of claim 17 wherein  $t$  = 12 to 16,  $u$  = 6 to 12,  $v$  = 12 to 18,  $w$  = 1 to 3, and  $x$  = 0 to 10,000,  $y$  = 2 to 300,000, and  $z$  = 0 to 10,000.

19. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following structure



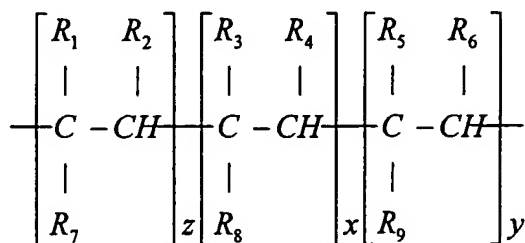
where  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  = H or  $\text{CH}_3$



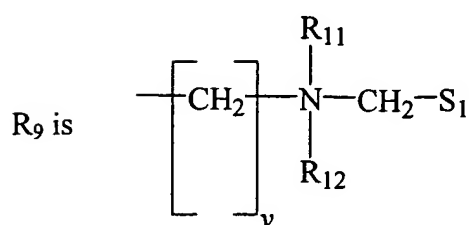
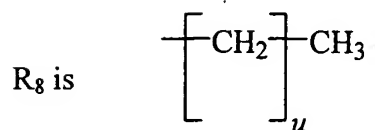
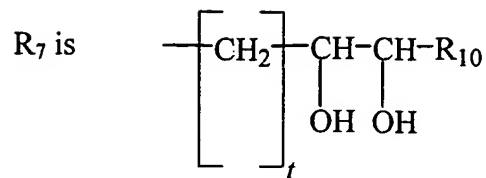
16  
17 wherein  $R_{10}$ ,  $R_{11}$ ,  $R_{12} = \text{H}$  or  $\text{CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $x + y + z = 3$  to  
18  $300,000$ , and  $\text{S}_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ .  
19

20 20. The wellbore fluid of claim 19 wherein  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $x = 0$  to  
21  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .  
22

23 21. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following  
24 structure:



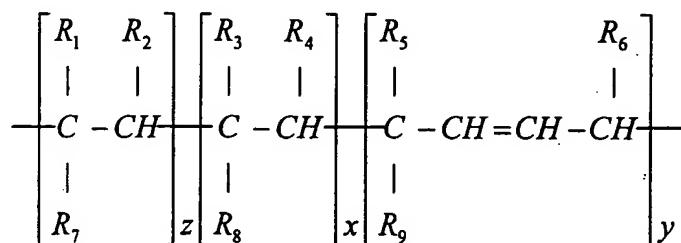
28 wherein  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



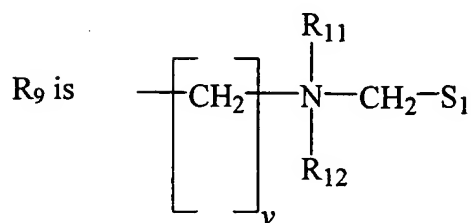
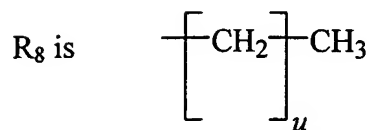
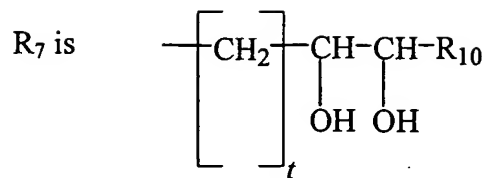
16 wherein,  $R_{10}, R_{11}, R_{12} = \text{H}$  or  $\text{CH}_3$ ,  $t = 1$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 1$  to  $18$ ,  $w = 1$  to  $12$ , and  $x$   
17  $+ y + z = 3$  to  $300,000$ .

18  
19 22. The wellbore fluid of claim 21 wherein  $t = 12$  to  $16$ ,  $u = 6$  to  $12$ ,  $v = 12$  to  $18$ ,  $w = 1$   
20 to  $3$ , and  $x = 0$  to  $10,000$ ,  $y = 2$  to  $300,000$ , and  $z = 0$  to  $10,000$ .

21  
22 23. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following  
23 structure:



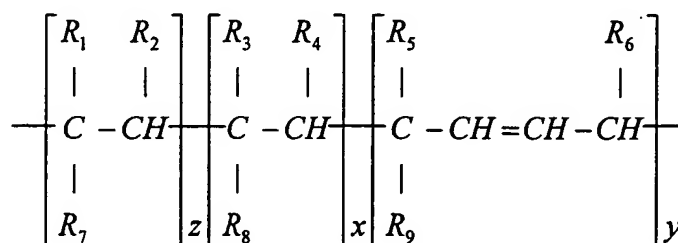
25  
26 wherein  $R_1, R_2, R_3, R_4, R_5, R_6 = \text{H}$  or  $\text{CH}_3$



wherein  $R_{10}$ ,  $R_{11}$ ,  $R_{12}$  = H or  $\text{CH}_3$ , and  $t = 1$  to 16,  $u = 6$  to 12,  $v = 1$  to 18,  $w = 1$  to 3, and  $x + y + z = 3$  to 300,000 and  $S_1 = \text{CO}_2^-$  or  $\text{SO}_3^-$ .

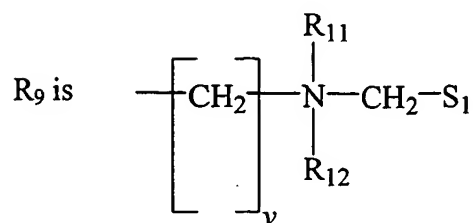
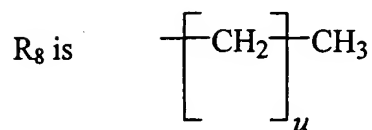
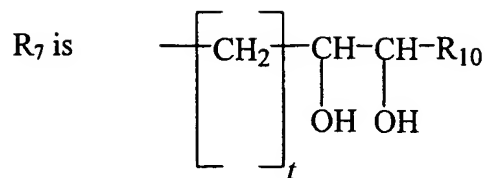
24. The wellbore fluid of claim 23 wherein  $t = 12$  to 16,  $u = 6$  to 12;  $v = 12$  to 18,  $w = 1$  to 3,  $x = 0$  to 10,000,  $y = 2$  to 300,000 and  $z = 0$  to 10,000

25. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following structure:



wherein  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  = H or  $\text{CH}_3$

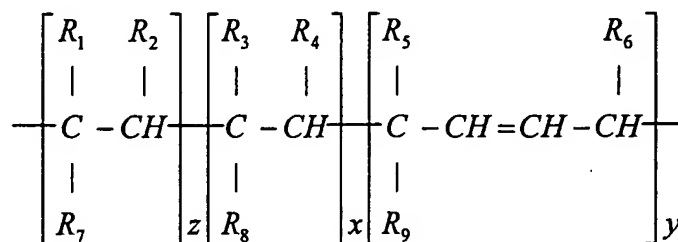




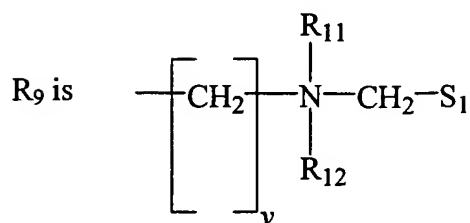
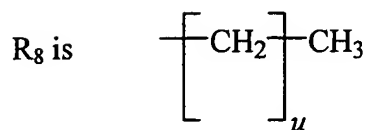
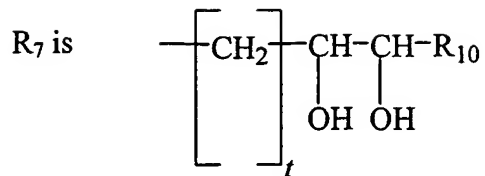
wherein  $R_{10}$ ,  $R_{11}$ ,  $R_{12}$  = H or  $\text{CH}_3$ ,  $t$  = 1 to 16,  $u$  = 6 to 12,  $v$  = 1 to 18, and  $x + y + z = 3$  to 300,000.

28. The wellbore fluid of claim 25 wherein  $t$  = 12 to 16,  $u$  = 6 to 12,  $v$  = 12 to 18,  $w$  = 1 to 3, and  $x$  = 0 to 10,000,  $y$  = 2 to 300,000, and  $z$  = 0 to 10,000.

27. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following structure:



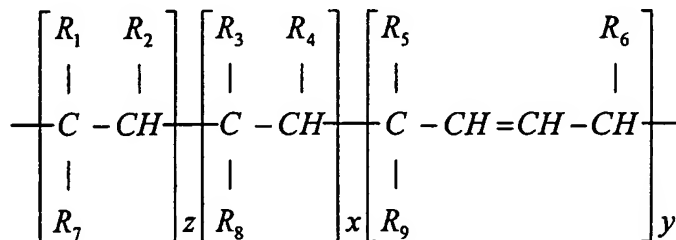
wherein  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  = H or  $\text{CH}_3$



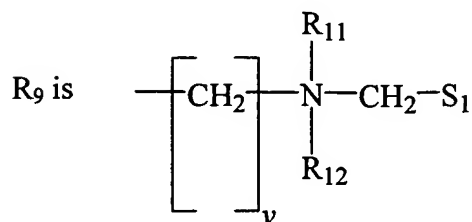
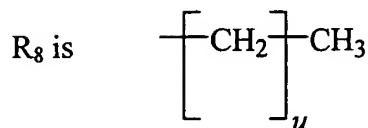
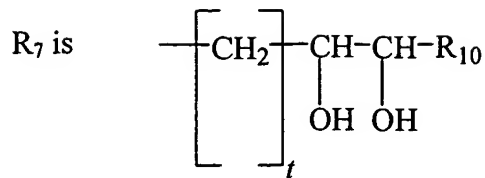
wherein  $R_{10}$ ,  $R_{11}$ ,  $R_{12}$  = H or  $\text{CH}_3$ ,  $t$  = 1 to 16,  $u$  = 6 to 12,  $v$  = 1 to 18,  $x + y + z$  = 3 to 300,000, and  $S_1$  =  $\text{CO}_2^-$  or  $\text{SO}_3^-$ .

28. The wellbore fluid of claim 27 wherein  $t$  = 12 to 16,  $u$  = 6 to 12,  $v$  = 12 to 18,  $w$  = 1 to 3, and  $x$  = 0 to 10,000,  $y$  = 2 to 300,000, and  $z$  = 0 to 10,000.

29. The wellbore fluid of claim 9 wherein the oligomer or polymer has the following structure:



wherein  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  = H or  $\text{CH}_3$



wherein,  $R_{10}$ ,  $R_{11}$ ,  $R_{12}$  = H or  $\text{CH}_3$ ,  $t$  = 1 to 16,  $u$  = 6 to 12,  $v$  = 1 to 18,  $w$  = 1 to 12, and  $x + y + z$  = 3 to 300,000.

30. The wellbore fluid of claim 29 wherein  $t$  = 12 to 16,  $u$  = 6 to 12,  $v$  = 12 to 18,  $w$  = 1 to 3, and  $x$  = 0 to 10,000,  $y$  = 2 to 300,000, and  $z$  = 0 to 10,000.

31. The wellbore fluid of claim 9 wherein the oligomers or polymers are crosslinked with polyvalent metal ions, formaldehyde, or glutaraldehyde.

32. The wellbore fluid of claim 31 wherein the polyvalent metal ions are selected from the following:  $\text{Fe}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{UO}_2^{2+}$ ,  $\text{PbO}^{2+}$ ,  $\text{Al}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Cr}^{3+}$ ,  $\text{Ce}^{3+}$ ,  $\text{Ti}^{4+}$ ,  $\text{Zr}^{4+}$ ,  $\text{Sn}^{4+}$  and mixtures thereof.

33. A method of making a wellbore fluid comprising blending:

- an aqueous fluid phase;
- a viscoelastic surfactant;
- a water-soluble inorganic salt;

an oligomer or polymer soluble in an aqueous salt solution, the oligomer or polymer comprising a hydrophobic oligomeric or polymeric backbone made from the oligomerization or polymerization of alkene or alkyne monomer groups, or mixtures thereof, the oligomer or polymer further comprising zwitterionic functional groups attached to the hydrophobic backbone, wherein the oligomer or polymer is hydrophilic in the zwitterionic functional groups and hydrophobic in the backbone hydrocarbon chain to form micellar assemblies such that the oligomers or polymers develop viscoelastic character prior to a polymerization step.

34. The method of claim 33 further comprising a polymerization step of the polymer or oligomer, then drying the product and subsequently admixing it into a solution of conventional surfactants.

35. A method of drilling a subterranean well, the method comprising:  
drilling the subterranean well using a rotary drilling rig and circulating a drilling fluid in the subterranean well, wherein the drilling fluid is the wellbore fluid of claim 1.

36. A method of reducing the loss of fluid out of a subterranean well, the method comprising injecting into the subterranean well a wellbore fluid as recited in claim 1.

## ABSTRACT

1  
2 Compositions for increasing the thermal and pressure stability of well fluids  
3 viscosified using viscoelastic surfactants, the compositions including an effective amount  
4 of an oligomeric or polymeric compound that has a thermally stable backbone structure  
5 and at least one pendent viscoelastic surfactant functional group. Preferred compositions  
6 for increasing the stability of well fluids viscosified using monomeric viscoelastic  
7 surfactants include an effective amount of an oligomeric or polymeric compound that has  
8 a thermally stable backbone structure and a multiplicity of pendent viscoelastic surfactant  
9 functional groups attached to said backbone structure through relatively long  
10 hydrocarbon chains, 1 to 18 carbons in length.



IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Application of:  
Robert Horton, et al.

Serial No.: 10/620,041

Filed: July 15, 2003

For: SURFACTANT-POLYMER COMPOSITIONS OF  
ENHANCING THE STABILITY OF  
VISCOELASTIC-SURFACTANT BASED FLUID

Group Art Unit: 1712

Confirmation No.: 7482

Examiner: Charles R. Richards

Atty. Dkt. No.: PA-00146US

**DECLARATION OF LEAH BARDIN  
IN SUPPORT OF PETITION UNDER 37 C.F.R. § 1.47**

Commissioner for Patents  
P.O. Box 1450  
Alexandria, VA 22313-1450

I, Leah Bardin, hereby declare the following:

1. I am currently employed as the patent paralegal of M-I L.L.C. and was at the time of the following occurrences.
2. On or about September 21, 2005, I attempted to locate Ms. Kamliya B. Abdur-Rahman via telephone book, internet and telephone, without success.
3. On September 21, 2005, I sent a certified letter, return receipt requested, to Ms. Abdur-Rahman enclosing the patent application and transmittal letter, Declaration she had previously signed but not dated, and a new Declaration for her execution. Also enclosed was a FedEx return envelope for her use in returning the document to us. I received the entire package back from the post office stating "return to sender, attempted not known."
4. On April 21, 2006, I contacted Human Resources at M-I L.L.C. and received the last known address for Ms. Abdur-Rahman on file, for which I then sent another certified letter, return receipt requested, to her enclosing the patent application and transmittal letter, Declaration

she had previously signed but not dated, the Assignment she signed and dated, July 1, 2003 and a new Declaration for her execution. Also enclosed was a return FedEx envelope for her use in returning the document to us. To date, I have not received the newly executed Declaration.

5. I hereby declare that all statements made of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

May 11, 2006  
Date

Leah Bardin  
Leah Bardin